A review on microbial surfactants: production, classifications, properties and characterization

Abstract: Surfactants are a surface-active group of molecular compounds with hydrophobic and hydrophilic 6 7 moieties in one single molecule that distributes themselves between two immiscible fluids, reduce surface/ 8 interfacial tensions and cause the solubility of non-polar compounds in polar solvents. Besides surface and 9 interfacial activities, they display properties such as solubilization, detergency, lubrication, emulsification, 10 stabilization and foaming capacity. Microbiologically derived surfactants are called biosurfactants. They are 11 produced as either metabolic products or as the surface chemistry of an actual cell. The employment of screening 12 techniques such as surface tension measurements, drop collapse test, oil spreading assay, emulsification index 13 (%EI24), CTAB/methylene blue agar plate test and strain characterization. Others are analytical techniques 14 including liquid chromatography-mass spectroscopy, thin layer chromatography, high-performance liquid 15 chromatography, Fourier transform infrared spectroscopy, nuclear magnetic resonance, fast atom bombardmentmass spectrometry and electrospray ionization-mass spectrometry. These hadve led to the identification of 16 biosurfactant producing microorganisms, properties and characterization of biosurfactants. Therefore, this review 17 18 tends to provide the current knowledge of the screening techniques and chromatography/spectroscopic tools 19 employed to study biosurfactants. Results from a detailed study of these tools can unveil new surfactant producing 20 microorganism, decipher chemical diversity and multifunctional properties of biosurfactants critical for applications 21 in diverse industrial sectors.

22 Keywords: Biosurfactants; chromatography; screening techniques, spectrometry; surface tension

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24 1 Introduction

25 Surfactants are surface-active group of molecular compounds with hydrophobic and hydrophilic moieties in 26 one single molecule and tend to distribute themselves between two immiscible fluids, reduce surface/interfacial 27 tensions (ST/IFT) and cause the solubility of non-polar compounds in polar solvents [1,2]. They display properties 28 such as solubilization, detergency, lubrication; have stabilizing and foaming capacity [3,4]. Surfactants are either 29 produced chemically or biologically. The biologically derived surfactants are known as biosurfactants (BSs) since 30 they are produced from living entities especially microorganisms. These molecules are produced as metabolic 31 products or the surface chemistry of the cells themselves [5]. Majorly, BSs are produced from aerophilic microbes in 32 aqueous media with carbon source feedstock such as hydrocarbons, carbohydrates, fats and oil which are mostly 33 from bacteria genera (Pseudomonas, Bacillus and Acinetobacter), fungi genera (Aspergillus, Fusarium) and yeast 34 (Candida, Pseudozyma) [6]. The most common BSs are rhamnolipids, surfactins, sophorolipids, emulsans, 35 mannosylerythritol lipids. These surface-active compounds play a physiological role for the benefit of the BSs 36 producing microorganisms to grow on water-immiscible substrates, ensure exponential biomass increase, exhibit 37 antimicrobial activities against possible predators, make them survive inhospitable environmental conditions, 38 virulence and cell desorption for survival [7] The physiological roles differ with the class a particular biosurfactant 39 belongs to.

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40 Broadly, biosurfactants are grouped into low and high molecular weight (LMW and HMW) biosurfactants. The LMW-BSs lower ST and ITF while the HMW-BSs are more of emulsion-stabilizing agents. Glycolipids, 41 42 lipopeptides and phospholipids belong to the LMW biosurfactants while the HMW biosurfactants are particulate and 43 polymeric [8,9]. Based on chemical composition, biosurfactants are grouped into glycolipids (rhamnolipids, 44 sophorolipids, trehalolipids, mannosylerithritol lipids), lipopeptides (surfactin, lichenysin, iturin, fengycin, 45 serrawettin), fatty acids/phospholipids/neutral lipids (phosphatidylethanolamine, spiculisporic acid), polymeric 46 biosurfactants (emulsan, alasan, biodispersan, liposan) and particulate biosurfactants (vesicles, whole-cell) [10-12]. 47 This classification is made possible by chromatographic and spectroscopic studies. Studies conducted with these 48 analyses had proven the hydrophobic moiety of BSs to be a long-chain fatty acid while the hydrophilic component 49 could either be alcohol, amino acid, carbohydrate, phosphate, carboxyl acid or cyclic peptide [13].

50 These components of the biosurfactants elicited in the supernatant are subjected to different surfactant 51 activity tests which include hemolytic activity test, ST measurements, drop collapse test, oil spreading assay, 52 emulsification index (%EI24), CTAB/methylene blue agar plate test and strain characterization [14-18]. In an 53 optimum environmental condition (including availability of carbon and other nutrients sources) a competent 54 microorganism can yield enough biosurfactants in a bioreactor which can be extracted by several available options 55 (acetone precipitation, ethanol precipitation, acid precipitation, solvent extraction technique, etc.) which can be 56 purified by either __dialysis/lyophilization, thin layer chromatography or isoelectric focusing [19-22]. Purified 57 biosurfactants can be characterized using nuclear magnetic resonance (NMR), liquid chromatography-mass 58 spectroscopy (LC-MS), high-performance liquid chromatography (HPLC), Fourier transform infrared spectroscopy 59 (FT-IR) [19,23-25]. The employment of these techniques has led to the identification of novel biosurfactant 60 producing microorganisms and uncommon biosurfactants as reported in scientific literature. Therefore, the purpose 61 of this review is to provide current knowledge on biosurfactant-producing microorganisms, the screening techniques 62 and the chromatography/spectroscopic tools employed to study biosurfactants.

64 **2 Biosurfactants**

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Biosurfactant is a portmanteau word that means surfactant from biological origin. Surfactant represents a 65 66 group of molecular compounds made up of tensio-active agents incorporated with both hydrophobic and hydrophilic 67 components that reduce ST and ITF by being distributed at the adjoining point of the two immiscible fluids, thereby 68 causing the solubility of one of the fluids in the other [2,18]. Their ability to reduce ST and IFT in most cases is 69 accompanied by detergency, lubrication, solubilisation and phase dispersion [4]. Therefore, from a technical point of view, biosurfactant represents surface-active agents which are either metabolic products produced by cells or the 70 71 surface chemistry of the actual cells [5] with an amphiphilic property that enable them to shape micelles that collect 72 at the interface between fluids of varying polarities with ultimate reduction of pressure and interfacial tension 73 pressure [26]. They are different from synthetic surfactants in that they are non-toxic, biodegradable, specific and 74 tolerant to extreme conditions [9]. They are majorly produced by aerophilic microorganisms in aqueous media with 75 either carbohydrates or hydrocarbons or fats, and oils as carbon source feedstock. These BSs are mostly from bacteria, fungi and yeast [27, 6] though plants and humans also produce biosurfactant [28] 76

77 Genera of Pseudomonas, Bacillus and Acinetobacter dominate the literature space of biosurfactants [4]. 78 production Notable genera among the veasts 79 are Torulopsis, Pseudozyma, Saccharomyces, Rhodotorula and Kluyveromyces [29-31]. For the 80 fungi, Aspergillus, Ustilago, Fusarium, Trichoderma, Penicillium [32-37] are well reported in research publications. A summary list of some species of bacteria, yeast and fungi that produce biosurfactants are listed in Table 1. 81

82

83 **3 Factors affecting biosurfactant production**

84 **3.1 Biosurfactant production**

85 Production of biosurfactants starts from sampling location. There are three commonly reported 86 environmental media that are commonly sampled which are stressed soil, stressed aquatic environments or pristine 87 environmental media [15-17] However, the most interesting results emanate from ecologically compromised 88 environments, mostly with hydrophobic compounds. Examples of sites where biosurfactants producing microbes 89 hadve been isolated are diesel polluted soil; [14] in an oilfield; [38] from extreme environments; [16] from an oil 90 reservoir; [39] in an automobile garage; [40] from sea harbor [41] and unpolluted soil. Media used for the isolation 91 of the microbes are mostly mineral salt media (MSM) and incorporated with organic substrates as the sole carbon 92 source and nutrient broth. The carbon source mostly used is hydrophobic [41-43] however; cheap renewable hydrophilic carbon sources are currently used. Nwachi et al. [44] Glucose ais the sole carbon source in MSM. 93 94 Competent microorganisms are inoculated into a broth media having suitable carbon, nitrogen and in a controlled 95 environment within optimum conditions. Biosurfactant (BS) production can be done through laboratory-scale or on 96 large scale (fermentation). A laboratory-scale production of BSs is illustrated in Fig. 1. 97

98 **3.2 Factors affecting biosurfactant production**

99 3.2.1 Nutrient factors and salt concentration

Carbon sources do influence the quantity and quality of BS production [45]. Crude oil, diesel, sucrose, glucose,
 glycerol are good carbon sources for BS production [7] However, researchers are now focusing attention on the use

102 of wastes to cut down the cost of downstream processing.

Nitrogen is a limiting nutrient and it is essential in the formulation of medium for BS production because it is very
 critical for microbial growth, protein and enzyme syntheses. Yeast, meat and malt extracts, urea, peptone,
 ammonium sulphate, nitrate and sodium nitrates are common nitrogen sources used in BS production [46,7]

- Phosphate is also very important for the growth of microorganisms. It is usually provided in triphosphates form.
 Maqsood and Jamal [47] reported that cultivation of gram-negative bacterium on ethanol with a low phosphate
 concentration yielded a maximum concentration of rhamnolipids. A mutant strain of *P. aeruginosa* (mutation caused
 by N-methyl-N-nitrosoguanidine) in a study produced 10 times more of rhamnolipid in comparison to the parental
 strain at 200 rpm/37°C [48].
- 111 Salt concentration is expected to influence BS production since cellular activities of microorganisms are affected
- by salt concentration [49]. Md [7] noticed in their study that some biosurfactant products were not affected by salt
- 113 concentrations up to 10% (weight/volume), though a slight reduction in the critical micelle concentration (CMC)

114 value was detected. A range of 1-10% of NaCl concentration has been proven to have an optimal influence

on *Pseudomonas aeruginosa*, which produces rhamnolipids [49,50]. Table 2 shows the optimum conditions underwhich some microorganisms produce maximum yield of BSs.

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118 **3.2.2 Environmental factors**

Environmental factors are extremely important because it affects the characteristics and output of BS. To
obtain an appreciable yield of BS, it is vital to optimize the bioprocess as the product may be susceptible to changes
in pH, temperature, agitation speed or aeration.

122 Temperature between 25-37 °C influences the growth of biosurfactant producing organisms [51]

A pH of culture medium around 8 has been reported to accompany enhance the best production of biosurfactants
[31] However, Jagtap *et al.* [52] reported the optimal pH to be 7 for most microorganisms. Bacteria tend to do best
at alkaline pH while yeast and fungi thrive best in acidic condition, but there are some exceptions. For
instance, *Yarrowia lipolytica* experience its optimal growth at pH 8 [53] and *Lactobacillus* spp. thrives at a pH of 6
[54].

128 Incubation periods also affect biosurfactants production. Auhim and Mohamed51 demonstrated that optimal
129 incubation period for *Azotobacter chrococcum* is 4 days. Fontes *et al.* [53] noted 24 h for *Yarrowia lipolytica* while
130 Bhardwaj *et al.* [36] reported 8 days for *Candida lipolytica*.

Aeration and agitation are very important factors that influence BS production since they both facilitate oxygen transfer into the culture medium [26]. Adamczak and Bednarski [46] demonstrated that improved yield value of BS (45.5 g/l) was achieved when the air-flow-rate was 1 vvm and the dissolved O₂ concentration was sustained at 50% saturation. Agitation of between 120 rpm to 200 rpm is most common in microbial growth studies [56]. It is safe to conclude that incubation period remains the most unpredictable.

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137 **4 Properties of biosurfactants**

Biosurfactants properties such as ST reduction, detergency, emulsifying capacity, foaming capacity, stabilizing capacity, low-CMC and solubility, are key in performance evaluation of BS and selection of microorganisms with BS producing potentials [57] Though chemical composition diversity and properties may differ, some properties are common to most of the biosurfactants [58]

142 **4.1 Surface and interface activity**

An efficient BS reduces fluid ST at a lower concentration in comparison to synthetic surfactants or ineffective BS. The CMC of biosurfactants (a measure of effectiveness) ranges from 1-2000 mg/l, whereas IFT (oil/water) and ST are around 1 and 30 mN/m respectively [4]. According to [26] a good surfactant should lower ST of water from 72 to 35 mN/m and the IFT of water/hexadecane from 40 to 1 mN/m. For example, rhamnolipids lower the ST of water and IFT of water/hexadecane to 26 mN/m and 1 mN/m respectively; surfactin from *B*. *subtilis* reduces the ST of water to 25 mN/m while the IFT of water/hexadecane to less than 1 mN/m and

- sophorolipid from *T. bombicola* reduces the ST to 33 mN/m and the IFT to 5 mN/m. In general, BSs are moreeffective and powerful since their CMC are lower than chemical surfactants [26].
- 151

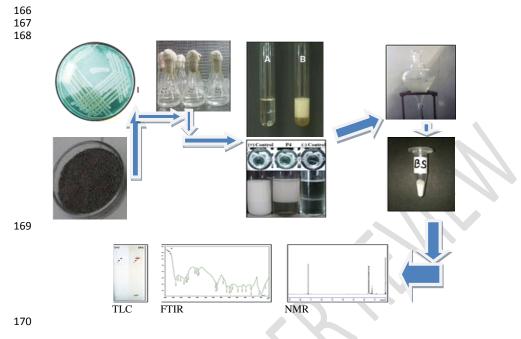
152 **4.2** Tolerance to pH, ionic strength, temperature

153 Many BSs and their surface activities are not much affected by environmental parameters such as pH and 154 temperature. For example, the lipopeptide produced by Bacillus licheniformis JF-2 was tolerant to a temperature of 155 75 °C for up to 140 hours and a pH range of 5-12 [4]. Biosurfactants also tolerate high salt concentrations up to 5 156 times the concentration (2%) that could inactivate synthetic surfactants [21]. A lipopeptide from B. subtilis was 157 subjected to different extreme conditions (autoclaving condition (121°C/15 minutes), -18 °C for 6 months, varying pH between 5-11 and 20% NaCl concentrations) without losing its surface activity property [26]. Mukherjee [58] 158 159 demonstrated that a BS produced by Arthrobacter protophormiae withstood a temperature of 30-100 °C and a pH of 160 2 to 12. Since industrial processes pass through extreme pH, temperature, and pressure [2], it is expedient to use 161 biosurfactants in industries that require extreme conditions.

162

164 Table 1. Biosurfactant producing species of bacteria, fungi and yeast

Genus	Phylum/Division	Class	Species	Note
	Bacteria			
Pseudomonas	Proteobacteria	Gammaproteobacteria	P. aeruginosa	Gram negative, rod-shaped, an opportunistic pathogen, versatile
Bacillus	Firmicutes	Bacilli	B. subtilis	Gram positive, rod-shaped, found commonly in soil, tolerant
Acinetobacter	Proteobacteria	Gammaproteobacteria	A. calcoaceticus	Non-motile, gram negative coccobacillus, commensal in humans
Rhodococcus	Actinobacteria	Actinobacteria	R. erythropolis	Nonsporulating, non-motile, gram positive, hydrocarbon degrader
Mycobacteriu	n Actinobacteria	Actinobacteria	M. aurum	Non-pathogenic, fast-growing saprophytic mycobacterium
Serratia	Proteobacteria	Gammaproteobacteria	S. marcescens	Gram negative, rod-shaped opportunistic pathogen
Corynebacteri	um Actinobacteria	Actinobacteria	C. kutscheri	Gram negative, pathogenic to rodents but can exist as free living
Nocardia	Actinobacteria	Actinobacteria	N. amarae	Slow-growing, often gram positive, alkyl benzenes degrader
Lactobacillus	Firmicutes	Bacilli	L. casei, jensenii	Probiotic found in gut, have wide pH and temperature range
Arthrobacter	Actinobacteria	Actinobacteria	A. paraffineus	A soil microorganism, Gram-positive obligate aerobes
	Fungi		1 00	
Aspergillus	Ascomycota	Eurotiomycetes	A. flavus	Ubiquitous mold few of which causes illness in humans
Fusarium	Ascomycota	Sordariomycetes	F. proliferatum	Filamentous fungi widely found in soil and associated with plants
Penicillium	Ascomycota	Eurotiomycetes	P. crysogenum	Industrially important fungi and usually the most abundant in soil
Ustilago	Basidiomycota	Ustilaginomycetes	U. maydis	Smut fungi parasitic to grasses
Trichoderma	Ascomycota	Sordariomycetes	T. viride	Many species are opportunistic avirulent plant symbionts
Mucor	Zygomycota	Zygomycetes	M. mucedo	Ubiquitous filamentous fungi that causes mucormycosis
Rhizopus	Zygomycota	Zygomycetes	R. oryzae	Ubiquitous filamentous fungi that seldom causes serious infections
Phoma	Ascomycota	Dothideomycetes	P. complanata	A dematiaceous filamentous fungus found in plants and soil
Curvularia	Ascomycota	Dothideomycetes	C. clavata	A facultative pathogen of many plants and common in soil
	Yeast	-		
Candida	Ascomycota	Ascomycetes	C. albicans	Yeast found in soil/humans and when overgrown causes disease
Saccharomyce	s Ascomycota	Ascomycetes	S. cerevisiae	Brewer's or baker's yeast important in food production
Pseudozyma	Basidiomycota	Ustilaginomycetes	P. rugulosa	Environmental yeast that rarely cause diseases
Yarrowia	Ascomycota	Saccharomycetes	Y. lipolytica	A yeast that can use unusual carbon sources
Rhodotorula	Basidiomycota	Microbotryomycetes	R. babjevae	An environmental yeast that acts as an opportunistic pathogen
Kluyveromyce	s Ascomycota	Saccharomycetes	K. marxianus	A probiotic yeast with industrial applications
Aureobasidiun	a Ascomycota	Dothideomycetes	A. pullulans	A yeast-like fungi that is ubiquitous, isolated as saprophytes
Geotrichum	Ascomycota	Saccharomycetes	G. cadidum	A ubiquitous filamentous yeast-like fungi
Galactomyces	Ascomycota	Saccharomycetes	G. geotrichum	A yeast used as moisturizing agent with antioxidant effect
Apiotrichum	Basidiomycota	Tremellomycetes	A. loubieri	An anamorphic basidiomycetous yeast genus



171 Fig. 1. Schematic presentation of a laboratory-limited biosurfactant production procedure: sampling to isolation to

- 172 growth in MSM to screening of extraction and purification of biosurfactants to characterization of extracted
- 173 biosurfactant. Adapted from Singh and Tiwary51

Biosurfactant producer	Carbon source	Nitrogen source	Temperature	pH	Reference
Pseudomonas aeruginosa	Glycerol and other water- soluble carbons	Sodium and ammonium nitrates	30 -37 °C	7-8	Ikhwani et al. [49], Maqsood and Jamal [47]
Bacillus subtilis	Glucose (40 g/l) in the presence of activated carbon	Urea (6 g/l)	37 °C	7	Mnif et al. [59], Asad et al.60, Yeh et al. [56]
Acinetobacter calcoaceticus	Hydrocarbons	Sodium nitrate (2 g/l)	25 – 33 °C	8	Ohadi et al. [61], Vigneshwaran et al. [62]
Lactobacillus fermentum	Lactose	Peptone/ meat extract	25 – 37 °C	б	Satpute et al. [63], Gudina et al. [64]
Candida lipolytica	Glucose and canola oil (10% each)	Yeast extract and ammonium nitrate	27 °C	5	Bhardwaj et al. [36]
Saccharomyces cerevisiae	Galactose and fructose	Peptone water 94.0 mg/100ml	32 °C	4-11	Salari <i>et al.</i> [65]
Pseudozyma antarctica	Soybean oil	Yeast extract and urea	30 °C	-	Bhardwaj et al. [36]
Penicillium spp	Soybean oil (20 g/l)	Yeast extract (30 g/l)	35 °C	9	Sena <i>et al.</i> [66]
Fusarium spp	Ethanol acetate/methanol (5;1)		40 °C	5-9	Qazi <i>et al.</i> [67]

186 **4.3 Biodegradability and low toxicity**

187 Microbial surfactants, like other microbiologically derived compounds, can be easily degraded unlike the 188 synthetic surfactants so they can be applied in bioremediation, biosorption and waste management [68, 69]. Though 189 very little data are available that give credence to BS toxicity, they are generally accepted as non-toxic, proving they 190 can be used in health-related industry [26] The latter author disclosed that Corexit (a synthetic anionic surfactant) is 191 ten times more lethal than rhamnolipids through the use of LC50 test against Photobacterium phosphoreum. This 192 signifies that Corexit is far more toxic than rhamnolipids and possibly the comparison will be so between synthetic 193 surfactants and microbial surfactants. A comparative study between biosurfactant from P. aeruginosa and a popular 194 industrial synthetic surfactant revealed that the synthetic surfactants are toxigenic and mutagenic unlike the 195 biosurfactants- [26].

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197 4.4 Physiological properties

Microbial surfactants are secreted either extracellularly or attached to parts of cells during growth onhydrophobic substrates. Biosurfactants:

- 200 I) Allow microbes to grow on water-immiscible substrates by reducing the surface tension at the interface, thereby
- 201 making substrates/nutrients soluble for uptake, which is necessary for metabolism.
- II) Ensure exponential biomass increase needed by microorganisms by way of making soluble hydrocarbons ascarbon substrates and energy source, and also utilizing the biosurfactants themselves.

204 III) Modify bacterial cell surface properties. Kaczorek [70] highlighted some of the effects of biosurfactants on

bacterial cells which include among other things alteration in biomorphology, cell surface hydrophobicity, surfacefunctional groups, and electrokinetic potential.

IV) Exhibit antimicrobial activities towards various microorganisms. Yuliani *et al.* [71] demonstrated
 that *Bacillus subtilis* C19 produced lipopeptide that had selective antimicrobial effects against *Candida albicans*.

209 Biosurfactants also dissolve cell surface structure by their detergency property.

- 210 V) Impart stability under hostile environmental conditions, virulence and in cell desorption when organisms need to
- find new habitats for survival [26].
- 212 213

214 5 Basic analysis in biosurfactant study

215 **5.1 Biosurfactant screening methods**

Isolates (or supernatants) that exhibit good growth are subjected to various biosurfactants activity tests (Table 3) which include hemolytic activity test, surface tension measurements, drop collapse test; [17,72] oil spreading assay; [18] emulsification activity; [73] emulsification index (%EI24) [74] CTAB/methylene blue agar plate test [14], Penetration assay for high throughput screening [25], and molecular characterization [16] Biosurfactants producing microorganisms are characterized through this order of steps: cultural isolation, purification of isolates, DNA extraction, a polymerase chain reaction (PCR) of DNA, sequencing and phylogenetic analysis [16] Each of these steps is described in Table 4. 223 Table 3. Description of the basic analysis carried out in biosurfactants study

Inoculate isolates on blood agar medium (5% of fresh human blood) and incubate at 28 OC for 48-72 hours. The hemolytic activity will be assessed based on α , β and γ type hemolysis to ensure preliminary conformation on biosurfactant activity	If agar under the colony is dark and greenish (α -hemolysis); yellow and transparent (β -hemolysis) or remain unchanged then (γ -hemolysis)
Drops of oil placed on a slide and then add 10 μ l of the supernatant by piercing the drop using micropipette without disturbing the dome shaped of the oil.	If the drop collapsed within 1 minute then the test is considered to be positive for the drop collapse test
Add 40 μ l of distilled water into a Petri dish followed by the addition of 20 μ l of diesel oil to the surface of water then 10 μ l of supernatant dropped on to the oil surface.	If biosurfactant is present in the supernatant, the oil is displaced and a clearing zone is developed.
Add 2 ml of oil to the same amount of supernatant in a glass tube, then mixing it with a vortex for 2 minutes and leaving it to stand for 24 hours.	Biosurfactants float on the upper part of the tube. %EI24 is calculated by height of the biosurfactant divided the total liquid in the tube multiplied by 100
Prepare Bushnell Hass agar medium containing glucose (2%), CTAB (0.5 mg/ml) and methylene blue (0.2 mg/ml). Create equidistant wells using cork borer (4 mm) Add 30 μ l of supernatant into the labeled wells and incubate at 37 0C for 48- 72 hours	If test is positive dark blue complex will be formed indicating the presence of anionic biosurfactants
Pre-cultures of strains were prepared in a nutrient broth. A volume of 1 ml of inoculum was added to 100 ml mineral salt solution and 1% of filtered oil as hydrocarbon source. The mixtures with control samples (100 ml MSS and 1% filtered oil without bacterial strains) were incubated at 30° C on shaker at 150 rpm for 3 days. Measure surface tension with a tensiometer	If surface tension in the test sample is significantly lower than the surface tension in the control then test is positive
The cavities of a 96 microplate wells are filed with 150 μ l of hydrophobic paste consisting of silica gel and oil. The paste is covered with 10 μ l of oil followed by placing of coloured supernatant consisting of 90 μ l of supernatant and 10 μ l of a red staining solution	Colour will change from red to cloudy white if biosurfactants are present within 15 minutes.
	 blood) and incubate at 28 OC for 48-72 hours. The hemolytic activity will be assessed based on α, β and γ type hemolysis to ensure preliminary conformation on biosurfactant activity Drops of oil placed on a slide and then add 10 µl of the supernatant by piercing the drop using micropipette without disturbing the dome shaped of the oil. Add 40 µl of distilled water into a Petri dish followed by the addition of 20 µl of diesel oil to the surface of water then 10 µl of supernatant dropped on to the oil surface. Add 2 ml of oil to the same amount of supernatant in a glass tube, then mixing it with a vortex for 2 minutes and leaving it to stand for 24 hours. Prepare Bushnell Hass agar medium containing glucose (2%), CTAB (0.5 mg/ml) and methylene blue (0.2 mg/ml). Create equidistant wells using cork borer (4 mm) Add 30 µl of supernatant into the labeled wells and incubate at 37 0C for 48-72 hours Pre-cultures of strains were prepared in a nutrient broth. A volume of 1 ml of incculum was added to 100 ml mineral salt solution and 1% of filtered oil as hydrocarbon source. The mixtures with control samples (100 ml MSS and 19% filtered oil without bacterial strains) were incubated at 30°C on shaker at 150 rpm for 3 days. Measure surface tension with a tensiometer The cavities of a 96 microplate wells are filed with 150 µl of hydrophobic paste consisting of 90 µl of supernatant and 10 µl of a red

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Table 4. Steps and common procedures involved in molecular charaterisation of biosurfactantsproducing microbes

Stages	Bacteria	Fungi	Yeast
Isolation and purification	Enrichment with Bushnell Hass (BH) both supplemented with 1% diesel and culturing in BH medium with 1% diesel [75]	About 1 ml of soil suspension is plated out on potato dextrose agar [33]	Enrichment and culturing of yeast was carried out on yeast extract peptone dextrose broth and agar plate respectively [80]
DNA extraction PCR amplification	Conventional (CTAB method) and rapid approaches The 27F and 1492R primer pairs used for amplification [16]	Conventional (CTAB method) and rapid approaches The ITSF and ITS4R primer used for amplification [78]	Conventional (CTAB method) and rapid approaches The ITSF and ITS4R primer used for amplification [81]
Sequencing	The amplicons were sequenced using a model ABI 3700 capillary sequencer [76]	The PCR products were sequenced by mean of the mentioned primers in an Applied Biosystem 3130 sequencer [78]	The PCR amplicons sequenced using an automatic sequencer (ABI 3730) [81]
Phylogenetic analysis	BLAST software is used in comparing sequence in GenBank. Sequences are aligned using the software CLUSTALX Phylograms were constructed using MEGA software, with a 1,000-repetition bootstrap [77]	IT Sequences are downloaded from Gene Bank. Alignment done using the clustal W. in MEGA 7. Maximum parsimony analysis is done and branches is supported by the bootstrap (1000 replicates) method [79]	Sequence comparisons were performed using the BLAST program. Alignment by CLUSTALW and phylogenetic tree constructed with MEGA 7 software with bootstrap of 1000 repetition [82]

230 5.2 Estimation of biosurfactant activity

Biosurfactant activity can be estimated by measuring its ability to change ST and hydrophilic-lipophilic balance (HLB). When a significant amount of (bio)surfactant is introduced into a liquid system, a critical value (CV) is reached where the ST decreases no further. Above this CV, biosurfactant monomers aggregate to form bilayers, vesicles and micelles. This CV represents CMC which can be measured. Reduction of ST, IFT and CMC values can be measured. A new surfactant is usually compared with a surfactant of known HLB value to predict its property. The HLB value is between 0 and 20 [83]. The HLB can be calculated as follows:

237 HLB = 20(MWHP/MWSA)

238 Where MWHP stands for the molecular weight of the hydrophilic moiety, and MWSA stands for the molecular

239 weight of the whole surfactant. The HLB value provides grounds for prediction of a surfactant or biosurfactant

240 property as depicted in Table 5.

241

242 Table 5. Corresponding predicted properties of (bio)surfactants to HLB value.

HLB value	Predicted function	Application		
0-3 Anti-foaming agent		Used in fermentation process		
4-6	Water/oil emulsifiers	Improving diesel fuel		
7-9	Wetting agent	Aid nutrient uptake in plants		
8-18	Oil/water emulsifier	Bioremediation of pollutants		
13-15	Typical detergent	Industrial laundry detergents		
10-18	Solubilizer	In enhanced oil recovery		
A dented form Dented (1921				

Adapted from De *et al.* [83]

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247 6 Crude extraction of biosurfactants

248 A good number of methods exist for extracting biosurfactants among which are centrifugation, acetone 249 precipitation, ethanol precipitation, acid precipitation, ion-exchange chromatography, adsorption-desorption, 250 filtration and precipitation, foam fractionation, isoelectric focusing, ultrafiltration, dialysis and lyophilization and 251 solvent extraction [20] Solvent extraction will be explained here while others are summarized in Table 6. The 252 hydrophilic moieties of biosurfactants are soluble in non-polar solvents which make the extraction easy. Organic 253 solvents such as chloroform, methanol, butanol, hexane, acetic acid and isopropanol are commonly used for 254 biosurfactants extraction. To execute solvent extraction, the microorganisms is cultured in MSM broth for an 255 optimum incubation period on a shaker at 120 rpm at 37 °C, centrifuged at 15 x 10³ rpm for 15 minutes at 4 °C. The 256 supernatant is then treated with concentrated HCl until the pH is two; and left for 24 hours at 4 °C. After 24 h 257 centrifuge the acidified supernatant at 15 x 10³ rpm for 15 minutes at 4 °C and collect grey white precipitate that will 258 be formed for further extraction of the biosurfactants. Chloroform and methanol in the ratio (2:1 v/v) should be 259 added to precipitate the pellet and incubate at 30 °C for 15 minutes. Then centrifuged for 20 minutes under cooling 260 conditions and allow supernatant to evaporate by air drying. Dispense the product in sodium phosphate buffer (pH 261 7) and stored at 4 °C [42].

263 7 Purification of biosurfactants

- 264 There are good numbers of biosurfactants purification techniques, but the common ones are discussed here.
- 7.1 Thin-layer chromatography is a method used for the exploratory characterization of BSs. A part of the crude
 BS is separated on a silica-gel-plate using chloroform: methanol: water (10: 10: 0.5 v/v/v) mixture. The type of
 biosurfactant is characterized by utilising a developing solvent system with different colour developing reagent
 like ninhydrin. This reagent is applied to detect lipopeptide as a red spot, produced by biosurfactant [84]
 Sumaiya *et al.* [85] carried out TLC analysis and spotted sediments recovered from extracted biosurfactants on
 a TLC plate and sprayed with phenol sulphuric acid reagent. Brown spots were developed with an Rf value of
- 271 0.65 which indicates lipopeptide. Rhamnolipid was the standard biosurfactant they used.
- 7.2 Dialysis and lyophilization method is easy and cost-effective and widely exploited to enhance the purity of
 biosurfactant by using seamless cellulose dialysis bags. The collected precipitate containing the biosurfactant is
 dissolved in 5 -10 ml of sterile distilled water and dialyzed against double distilled water for 48 hours at 10 °C.
 The dialysate is stored at 4 °C in an airtight container for further use [20]
- 7.3 Isoelectric focusing (IEF) is one of the new approaches used for purification of biosurfactants. Its unit
 comprises of a single column, filled with density gradient solutions, electrolyte and non-ion conducting
 polymers. In the presence of electric influence, pH, density gradient, the ampholyte moves in the column until it
 reaches a neutral pH. The columns help to segregate fractions based on changes in pH. Once total separation
 occurs, electro-focusing is discontinued and the activity of purified BE is compared with the crude form [20]
 This procedure requires 10-12 hours at 400 V and a current of 1.5 A [86].
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Table 6. Selected techniques for biosurfactants extraction

Method	Description	Reference
Acetone precipitation	Culture is grown in a mineral salt medium supplemented with required constituents. Cell-free supernatant is mixed with ice-cold acetone to precipitate biosurfactants which is further suspended in phosphate buffer. Then mixture is incubated at 4 0C for 15–20 h to get the precipitated biosurfactants.	Patil and Chopade [87]
Ethanol precipitation	Culture broth is centrifuged at $11,000 \ge g$ for 20 minutes at 4 °C and biosurfactant is precipitated from the supernatant by using cold ethanol.	Phetrong et al. [88]
Acid precipitation	Acid hydrolysis is carried out by adding concentrated HCl to the supernatant to bring down the pH to 2 for the precipitation the biosurfactants at 4 °C. Centrifugation is followed and the pellet is further extracted by using appropriate solvent. Extracted material is filtered for removal of residues and evaporated to dryness using rotary evaporator.	Nitschke and Pastore [21]
Centrifugation	Following acid precipitation, biosurfactants-containing broth can be centrifuged at 12,000 rpm for 15 min at 4 0C to be easily collected as crude product. Once the pellet is obtained, it can be dried under N2 and extracted with solvents.	Nitschke and Pastore [21]
Ammonium sulphate precipitation	This method is used to precipitate high molecular weight biosurfactants such as emulsan, biodispersan. In this case the biosurfactant is precipitated by salting out process and the product is purified by dialysis procedure and lyophilized	Vandana and Singh [19]
Ion exchange chromatography	This method is carved out for anionic biosurfactants. Ion exchange resin is used to attract the biosurfactants at higher pH. The biosurfactant is eluted with a buffer	Satpute et al. [20]

	containing 10% (v/v) ethanol. Addition of a minimum of 0.6 NaCl to the buffer releases the biosurfactant from the resin	
Adsorption-desorption	Cell-free supernatant is added directly to the adsorbent column and 0.1 M phosphate buffer (pH 6.1) is used to equilibrate it. Exhaustion of the adsorbent resin is observed by ultra violet (U.V.) absorption. A wash of distilled water is given to the resin for removal of pigments and free fatty acids. Further, elution is carried out with methanol, which can be evaporated to obtain crude biosurfactants.	Abalos et al. [89]
Foam fractionation	Foam is collected through fractionation column and acidified with HCl down to pH $1.0-2.0$ to precipitate biosurfactants, which can be extracted with solvents. High yield of biosurfactants can be achieved by increasing the residence time of foam in the fractionation columns	Satpute et al. [20]
Filtration and precipitation	Precipitation was carried out with ethanol, acetone, ethanol acetic acid (1%)/5 N HCl in an equal volume of culture liquid. Extraction was performed twice to enhance the yield of biosurfactants	Turkovskaya <i>et al</i> . [22]

291 8 Characterisation of biosurfactants

There are many chromatography and spectroscopic methods used to characterize biosurfactants common among them are thin chromatography (TLC), Nuclear magnetic resonance (NMR), liquid chromatography-mass spectroscopy (LC-MS), Fourier transform infrared spectroscopy (FT-IR), high-performance liquid chromatography (HPLC). Each technique has its own strength and drawbacks as indicated in Table 7. Liquid chromatography-mass spectroscopy is the most commonly used instrument [23]

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298 8.1 Spectroscopy methods

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FT-IR can elucidate some components of an unknown mixture based on functional groups. In the process, 1 mg of
 purified biosurfactant (dried in freeze dryer) is ground with potassium bromide (100 mg), pressed for 30 s to achieve
 translucent pellets. Then analyze in an FT-IR device with the spectrum ranging from 450 - 4000 cm-1 at a
 resolution of 4 cm⁻¹ [90,16]

NMR provides information regarding the functional groups about the position of linkages within the lipid and
 carbohydrate molecules. This is based on transitions in atoms with a magnetic moment when an external magnetic
 field is applied. Smyth *et al.* [91] characterized glycolipid biosurfactant using NMR.

Fast atom bombardment-mass spectrometry uses a high energy beam of xenon atom and caesium ions to
stammer the sample and matrix (m-nitrobenzyl alcohol) from the probe's surface. Usually, the biosurfactants are
dissolved in methanol, mixed with matrix [20].

Electrospray ionization-mass spectrometry is a soft ionization method used for the production of gas-phase ions

311 for biological molecules with high molecular weight. It is so flexible that it can be used with MS (ESI-MS/MS), LC

312 (LE/ESI-MS), HPLC/ESI-MS) for a detailed insight of structural properties of molecules.92 Sabturani *et al.*93 used

- **313** ES-MS to characterize BS derived from *P. aeruginosa* UKMP14T.
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316 8.2 Chromatography methods

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Liquid chromatography-mass spectroscopy (LC-MS) analysis of biosurfactants requires an initial purification by removing the worst interferences and also to concentrate the sample to a significant quantity [23]. The LC-MS utilizes differences in hydrophobicity to achieve partitioning between a non-polar stationary phase and a polar mobile phase. The LC-MS technique is highly efficient in purifying and separating lipopolysaccharides (LP) congeners. Liquid chromatography-MS is best suited for a characterizing an unknown lipopolysaccharide.

324 Gas chromatography-mass spectroscopy (GC-MS) is used in characterizing biosurfactants where the mass 325 spectroscopy measures the MW of the compound. For this device, the sample needs hydrolytic cleavage between the 326 peptide/protein or carbohydrate/lipid portions present in the biosurfactant. The GC-MS results are analysed by fatty 327 acid derivatization to fatty acid methyl esters (FAME) and further conversion to trimethylsilyl (TMS) derivatives 328 [19].

High-performance liquid chromatography (HPLC) is a special kind of column chromatography used in the chemical and biochemical analysis in that it can separate a mixture of surface-active compounds, identify, quantify and purify separate components of biosurfactant mixture [23]. The use of HPLC has been reported in the characterization, quantification and purification of BSs [94] For example, purification of LP by HPLC was carried out by reversed-phase (RP)-HPLC using a semi-preparative C18 column and 0.1% trifluoroacetic acid/methanol/H2O as a mobile phase [95]

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Table 7. Chromatography and spectroscopic methods used to characterize biological molecules

Method	Advantages	Disadvantages
LC-MS	Large commercial and public libraries	Slow
	No derivatization required	Limited commercial libraries
	Many modes of separation available	
	Large sample capacity	
GC-MS	Sensitive	Slow
	Robust	Often requires derivatization
	Large linear range	Many analytes thermally unstable or too
		large for analysis
NMR	Rapid analysis	Low sensitivity
	High resolution	Convoluted spectra
	No derivatization needed	More than one peak per component
	Non-destructive	Libraries of limited use due to complex
		maxtrix
HPLC	Amenable to diverse sample types	Lack of ideal universal detector
	Accurate	Less separation efficiency
	Sensitive	Arduous for regulatory testing
	Speed	Costly
	Can analyze neutral, anions and cations on	1a
	single run	

338 Adapted from Satpute *et al* [20]

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341 9 Classification of the five groups of biosurfactant

Biosurfactants are classified based on their biochemical constituents or the species producing them.
Rosenberg and Ron [96] grouped biosurfactants into LMW molecules and HMW polymers. The former efficiently
lower ST and IFT while the latter are expert emulsion-stabilizing agents. The main classes of LMW-BSs
are lipopeptides, glycolipids and phospholipids, while the HMW-BSs are particulate and polymeric surfactants (Fig.
The hydrophobic moiety of BSs is long-chain fatty acids while the hydrophilic moiety either be alcohol, amino
acid, carbohydrate, cyclic peptide, or phosphate carboxyl acid [13].

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349 9.1 Classification based on molecular weight

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The LMW biosurfactants are biosurfactant compounds that lower the ST and IFT at the air/water interface. They
are generally glycolipids (rhamnolipids, sophorolipids, trehalolipids, mannosylerythritol lipids) or lipopeptides [97]
and are better reducers of ST and IFT [45]

The HMW biosurfactants are known as bioemulsifiers. They show effective stabilization property with respect to oil-in-water emulsions [26]. Besides, they can work at low concentrations and show considerable substrate specificity [98]. Examples include emulsans, alasans, biodispersans etc. Each of the specific class is discussed below.

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- 359 9.2 Classification based on chemical composition
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361 **9.2.1** Glycolipids

362 Glycolipids constitute a hydrophilic carbohydrate component and a hydrophobic fatty acid chain.
363 According to Marchant and Banat [97], the hydrophilic end is made up of different sugars: rhamnose (in rhamnolipids), sophorose (in sophorolipids), and mannose and erythritol (in mannosylerythritol lipids). Trehalose
365 and cellobiose lipids are other examples of glycolipids. However, most studied glycolipids are rhamnolipids.

- a) Rhamnolipids are amphiphilic compounds ideally comprised of 3-hydroxy fatty acids (hydroxydecanoic acid) linked through a β-glycosidic bond to mono- or di-rhamnose [26,99,100]
- 368 b) Sophorolipids are made up of disaccharide-sophorose β-linked to a long fatty acid with a chain length of
 369 16 18 carbon atoms with the presence of unsaturation [101] They can exist in a lactonic form [97] or in an
 acidic form [102]
- c) Mannosylerythritol lipids (MELs) have 4 major structural groups having 4-O-b-D-mannopyranosyl-D erythritol linked to 2 medium-length chains of fatty acyl esters [29,103]. Though MELs exist as MEL-A,
 MEL-B and MEL-C, the MEL-A is the most dominant [32]
- 374 d) Trehalolipids biosurfactants exist in various structural types. In some microorganisms, the disaccharide 375 trehalose linked at C-6 and C-6 to mycolic acid which is long-chain α-branched and β-hydroxy fatty acids
 376 [104]
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378 9.2.2 Lipopeptides and lipoproteins

This class of BSs, in general, comprises of cyclic peptides connected to a fatty acid. *Bacillus* cycliclipopeptides are formed of three different categories: fengycin, iturin and surfactin [2] Surfactin is the most studied among them. Structure of surfactin is made up of 7 amino acid cyclic peptide connected to a C13–C16 fatty acid, whereas iturin consists of 7 amino acids linked to C14–C17 and fengycin is composed of 10 amino acids with a fatty acid chain length of C14–C18 [105]. Other examples of lipoprotein include viscosin, lichenysin, serrawettin, gramicidin, polymyxin [106].

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9.2.3 Fatty acids and lipids (phospholipids and neutral lipids)

Many bacteria and yeasts yield appreciable amounts of these molecules during their growth on n-alkanes.
 The HLB these molecules relate to the hydrocarbon chain length in direct proportion [45]. In *Acinetobacter* sp.,
 phosphatidylethanolamine rich vesicles are synthesized and form optically clear micro-emulsions of oil-in-water.

Phosphatidylethanolamine synthesized by *R. erythropolis* while growing on n-alkane, lowers the IFT between

396 hexadecane and water to less than 1 mN/m and a CMC less than 30 [9].

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398 9.2.4 Polymeric and particulate biosurfactants

399 The best-studied polymeric BSs are emulsan, alasan, liposan, lipomanan and some other lipopolysaccharide 400 and polysaccharide-lipid (or protein) complexes. The lipopolysaccharides consist of lipid component, a core 401 polysaccharide and O-specific side chain polysaccharide bond together by covalently. Emulsan is an effective 402 emulsifying agent for oi-in-water, even at a very low concetration. Liposan is an extracellular water-soluble 403 emulsifier from Candida lipolytica and has 83% of carbohydrate and 17% of protein [26]. Extracellular membrane 404 vesicles (particulate BSs) can form microemulsions by partitioning hydrocarbons. These microemulsions aid alkane 405 metabolism by microbial cells [9] Vesicles of Acinetobacter spp. having a diameter of 20-50 nm and a buoyant 406 density of 1.158 g/cm, were screened to possess protein, phospholipids and lipopolysaccharides. Table 8 summarises 407 the major groups of BSs produced by microorganisms.

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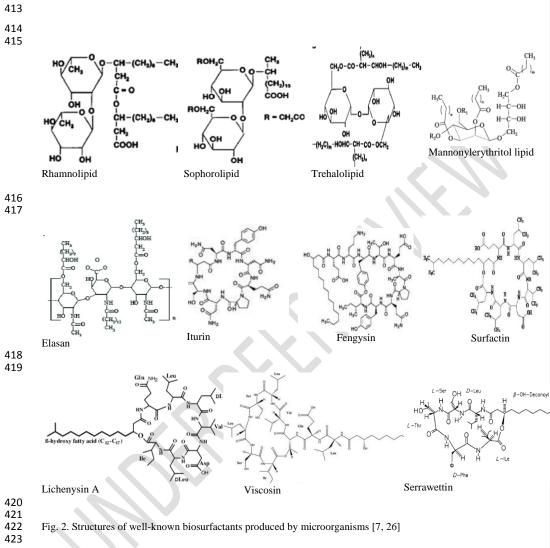


Fig. 2. Structures of well-known biosurfactants produced by microorganisms [7, 26]

424 Table 8. Classification if biosurfactants based on chemical structure and the key microorganisms that produces the specific type of

425 biosurfactants

Group	Туре	Microbial identity	Reference
Glycolipids	Rhamnolipids	Pseudomonas aeruginosa, Serratia rubidaea SNAU02	Whang <i>et al.</i> [107], Nalini and Parthasarath [108]
	Sophorolipids	Torulopsis bombicola, Trichosporon asahii, Mucor mucedo, Aspergillus flavus, Trichoderma viridis, Fusarium sp. S33, Rhizopus oryzae	Adekunle et al. [109], Balaji et al. [35], Lima et al. [34]
	Mannosylerythritol lipid	Candida antarctica Ustilago scitaminea,	Yu et al. [110]
	Annosylerythritol lipids	Pseudozyma rugulosa	Morita et al. [29]
	Trehalolipids	Arthrobacter paraffineus, Rhodococcus erythropolis, Gordonia amarae, Nocardia sp	Vigneshwaran et al. [62]
	Cellobiolipids	Ustilago maydis	Shekhar et al. [9]
Lipopeptides	Surfactin, iturin, fengycin	Bacillus subtilis, Bacillus mojavensis	Hmidet <i>et al</i> . [111]
	Lichenysin	Bacillus licheniformis	Madslien et al. [112]
	Viscosin	Pseudomonas fluorescens	Alsohim et al. [113]
	Serrawettin	Serratia marcescens	Hage-Hulsmann et al. [114]
	Phomafungy	Phoma sp. S31	Lima <i>et al.</i> [34]
Fatty acids, phospholipids	Spiculisporic acid	Penicillium spiculisporum	Ishigami et al. [115]
and neural lipids	Diglycosyl diglycerides	Lactobacillus fermentum,	Saharan and <i>et al.</i> [18]
	Glycerol-liamocin	Aureobasidium pullulans	Kim <i>et al.</i> [116]
	Phosphatidylethanolamine	Rhodococcus erythropolis	Stancu [10]
Polymeric biosurfactants	Emulsan	Acinetobacter calcoaceticus	Shekhar et al. [9]
	Alasan	Acinetobacter radioresistens	Toren <i>et al.</i> [117]
	Yasan	Yarrowia lipolytica	Yalcin et al. [118]
	Biodispersan	Acinetobacter calcoaceticus RAG-1	Rahman and Gakpe [45]
	Liposan	Acinetobacter radioresistens KA-53,	B H H H H H H H H H H
	Mannoprotein	Saccharomyces cerevisiae, Kluyveromyces marxianus	Dikit <i>et al.</i> [118]
	EPS	Galactomyces sp. Z3, Apiotrichum loubieri sp. TEMOS16, Geotrichum spp. Curvularia lunata IM 2901	Yalcin <i>et al.</i> [119]
Particulate biosurfactants	Vesicles Whole cell	Acinetobacter calcoaceticus	Muthusamy et al. [12]

10 Advantages and disadvantages of biosurfactants 426

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Biosurfactants has its merits and draw backs as reflected in Table 9.

428

429 Table 9. Advantages and disadvantages of biosurfactants

429	Table 7. Advantages and disadvantages of blost	urractants	
	Advantages	Disadvantages	
	Biosurfactants are easily degraded in the environment		
	Biosurfactants exhibits lower toxicity than the synthetic ones	Hemolytic activity of certain biosurfactants can rupture erythrocytes at 37 0C	
	Surfactants of biological origin have feature of	Biosurfactants is characterized with very low	
	compatibility thus being used in pharmaceuticals,	productivity. This is because over producing strains	
	cosmetics, food industries etc.	and recombinant stains are very rare	
		To get pure biosurfactants require multiple steps with	
	Biosurfactants can be produced from a variety of relatively cheap raw materials	attendant cost	
	Biosurfactants are effective surface and interfacial	Strong foam formation hampers the improvement in	
	tensions reducers	production yield	
	Biosurfactants can be produced from industrial waste	Production of biosurfactants in large scale is capital	
	and by-products thus key into acceptable production economics	intensive	
	Many biosurfactants are stable at extreme pH, salinities		
	and temperature		
	Biosurfactants are specific in their action, hence play		
	specific functions		
430	Adapted from De et al. [83]		
431			
432	11 Conclusion		
433	Biosurfactants are tensio-active molecules from	microorganisms as metabolic products or the actual cells of	
121	the surface showing Decides the larger big of that a decime Decide and Decide A is the test		

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433 434 their surface chemistry. Besides, the known biosurfactant producers: Pseudomonas, Bacillus, Acinetobacter, 435 Candida other genera such as, Apiotrichum, Aureobasidium, Galactomyces, Geotrichum Gordonia, Kluveromyces, 436 Phoma, and Yarrowia, and host of others are now included in the list. Biosurfactants have a unique property of 437 reducing ST and IFT of adjoining liquids. Biosurfactants which are not efficient in reducing surface tensions but are 438 efficient in stabilizing emulsions are known as bioemulsifiers. Gold standard techniques employed to determine biosurfactant properties are surface tension measurements, emulsification activity and emulsification index (%EI24). 439 440 Crude extraction of biosurfactants can be achieved through a number of methods including: centrifugation, acid 441 precipitation, ion-exchange chromatography, adsorption-desorption, foam fractionation. The most common 442 technique used in purifying crude biosurfactants are thin-layer chromatography, dialysis and lyophilization, and 443 isoelectric focusing. Characterisation of biosurfactants can be achieved by using chromatography and spectroscopy methods such as TLC, LC-MS, HPLC, FT-IR, NMR. Biosurfactants help microorganisms to metabolise 444 445 hydrocarbons, solubilize hydrophobic compounds and exhibit antimicrobial activities, thus have multifunctional 446 properties that can be relevant in industrial applications.

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