	Original Research Article
	Preservation of the Moringa oleifera
	Constituents by Freeze-Drying
ABSTRACT	

Aims: The objective of this research was to evaluate the preservation of the constituents of the powder obtained from the moringa seeds by freeze-drying comparing it with the *in natura*.

Place and Duration of Study: Sample: The work was conducted at the Laboratory of Processing and Storage of Agricultural Products, Department of Agricultural Engineering, Federal University of Campina Grande, Brazil, in the period from August to November 2019. **Methodology:** To obtain the samples the seeds were peeled and macerated manually. Freeze-drying was done in a Liotop® L101 benchtop freeze drier. After dehydration the samples were disintegrated and the physical and physico-chemical constituents were evaluated before and after freeze-drying in terms of apparent density, real density, porosity, compacted density, compressibility index, Hausner factor, solubility, moisture content and activity, ash, titratable total acidity, pH, protein, lipids and carbohydrates.

Results: The *in natura* powder presented better results for the physical analyzes of the densities: apparent, real and compacted, however, it was observed that for the other physical parameters and physicochemical constituents the freeze-drying promoted the preservation of these in front of the *in natura*. Freeze-drying caused a significant reduction in moisture content, pH and lipid activity, making the powders more stable and contributing to the maintenance of their physico-chemical qualities. The inverse was observed for the ash, protein and carbohydrate contents, where freeze-drying promoted increases in their contents.

Conclusion: Freeze-drying presented to the moringa as an adequate method in the preservation of its constituents, with emphasis on physicochemical.

- 11 12
- Keywords: Conservation, Drying, leguminous, Plant, Seeds
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1. INTRODUCTION

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Moringa was introduced in Brazil as an ornamental plant around 1950 and since then has been widely cultivated because of its high food value, mainly leaf and seed, and because of its high medicinal value [1]. In Brazil, in some regions, it is used as an alternative solution to clarify water supply in rural communities [2]. The efficiency of the use of moringa as a natural coagulant in water treatment is reported by several researchers, among them Baptista et al. [3].

In this context, the development of new technologies has been pointed to the use and preservation of the constituents of the moringa. Drying technologies are used to meet industry needs by significantly reducing the costs of operations such as packaging, transportation, storage, and providing the consumer with a quality product for a longer period of time [4]. Among these technologies, freeze-drying, due to its low temperature and
absence of atmospheric air, preserves the constituents of the natural product, also allowing
the chemical, nutritional and sensory properties of the powder to be practically unchanged,
besides having characteristics that hinder the development of microorganisms that could
promote its deterioration [5].

32 According to Park et al. [6] the type of drying to be used depends, among other factors, on 33 the product to be dehydrated, its chemical constitution and the physical characteristics of the desired product. Freeze-drying has been used and recommended to dry products with high 34 added value, which have delicate aromas or textures or that are sensitive to the use of heat. 35 36 For this reason, the objective of this research was to evaluate the preservation of the constituents of the powder obtained from the moringa seeds by freeze-drying comparing it 37 with the in natura, since the moringa drying process should be ensured not to exceed 60 °C 38 39 avoiding that the protein content is damaged [7].

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41 2. MATERIAL AND METHODS

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43 **2.1 Place of research**

44 The work was developed at the Laboratory of Processing and Storage of Agricultural 45 Products, Federal University of Campina Grande, Campina Grande, Brazil. To obtain the 46 samples, the seeds were peeled and macerated manually with the aid of mortar and pistil. 47 Freeze-drying was done in a Liotop® L101 benchtop freeze drier. The pulps, to be 48 lyophilized, were obtained by the addition of 50 ml of distilled water to 100 g of the in natura powder. Then they were inserted in plastic forms and subjected to freezing in a freezer at -49 18 °C for 24 hours. Then, the frozen samples were lyophilized at -50 °C for 25 hours [8]. 50 After dehydration, they were disintegrated and the physical and physicochemical 51 constituents of the powders were evaluated in natura and lyophilized. 52

53 2.2 Physical analyzes

54 Density apparent - was calculated according to the adapted method of Caparino et al. [9]; 55 actual density - this was obtained by weighing 1 g of the powder in a 10 mL graduated 56 cylinder, completing the volume of the beaker with oil, determining the amount of oil needed 57 to complete the 10 mL beaker; porosity - was determined by the method of Krokida and 58 Maroulis [10]; compacted density - was determined from the mass contained in a 10 mL 59 beaker after being manually tapped 50 times on the surface of a bench according to Tonon et al. [11]; compressibility index (CI) - was obtained by comparing the apparent density and 60 the compacted density of the powder; Hausner factor - was determined from apparent and 61 compacted density [12]; Solubility - determined according to the methodology adapted from 62 [13]; hygroscopicity - was obtained following the methodology proposed by Goula and 63 Adamopoulos [14]; moisture content - determined by the oven drying method at 105 °C for 64 24 hours [15]; water activity - obtained through direct reading on the "Aqua-Lab" equipment, 65 66 Decagon brand, model 3TE.

67 **2.3 Physicochemical analysis**

The physico-chemical determinations, except for lipids, were performed according to the methodology of Brazil [15]: Ash - by incineration of samples in muffle at 550 °C for 24 hours; total acidity - given by titration; pH - was obtained by direct reading of the samples in digital pH meter; total protein content - determined by the Micro-Kjeldahl method; lipids - by the modified method of Bligh and Dyer [16]; carbohydrates - removing from 100 the sum of water, lipids, proteins and ashes.

74 **2.4 Statistical analysis**

The results were submitted to analysis of variance (ANOVA) and the means comparison was performed by the Tukey test at 5%, using the statistical program Assistat 7.7 [17].

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79 3. RESULTS AND DISCUSSION

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81 Table 1 shows the results obtained for the physical and physicochemical analyzes of the 82 constituents of the in natura and lyophilized moringa. It was observed that the physical 83 constituents of the moringa powder, with the exception of solubility and hygroscopicity, 84 presented a statistical difference, and the apparent, real and compacted densities showed 85 higher values for the *in natura* samples. Behavior that is due to the fact that the powder *in* 86 natura present greater and apparent density, due to the greater compaction of its particles. On the other hand, it was verified that the lyophilized powder is more porous and tends to 87 present lower densities, because of the smaller pores. The powders had different values of 88 89 density than those obtained by Zea et al. [18] for guava powder and lyophilized guava and pitahie mix (1.474 and 1.503 g/cm³, respectively). According to Ceballos et al. [19] the 90 density is one of the factors that interferes in the wettability of powders, an important 91 characteristic as it affects the first phase of the reconstitution of a powder product. Thus, 92 93 because the lyophilized powder is presented with greater porosity than the in natura, it may 94 behave differently regarding the resistance and the movement of the air during the drying 95 and storage process.

As for the compressibility values, the in natura and lyophilized moringa powders fall within 96 the classification of Santhalakshmy et al. [20] (20%) to poor (26%), since values between 15 97 and 20% indicate good fluidity, between 20-35% poor fluidity, between 35-45% poor fluidity 98 99 and greater than 45% very poor fluidity. As for the Hausner factor, the in natura and lyophilized powders had an intermediate-to-easy flow, since materials with a Hausner 100 101 number greater than 1.4 are classified as cohesive and when less than 1.25 are easily flow 102 able, the *in natura* powder presented the best result because it was easily drained. This property is directly linked to the moisture content of the studied material, that is, the wetter 103 104 the powder, the greater the cohesiveness, making it more difficult to flow the powder due to 105 the formation of liquid bridges between the particles.

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Table 1. Physical and physico-chemical constituents of the moringa powder *in natura* and freeze-dried.

Constitutions of the moringa	Powder		
	In natura	Freeze-dried	CV* (%)
Apparently density (g/cm ⁻³)	0.49a	0.30b	1.97
Actual density (g/cm ⁻³)	0.65a	0.49b	5.43
Porosity (%)	18.51b	31.34a	3.68
Compressed density (g/cm ⁻³)	0.62a	0.41b	1.12
Compressibility Index (%)	20b	26a	4.74

Hausner factor	1.27b	1.34a	1.40
Solubility (%)	74.03a	73.89a	3.13
Hygroscopicity (%)	97.26a	95.08a	1.80
Moisture content (%)	5.24a	1.76b	0.96
Water activity (a _w)	0.62a	0.51b	0.94
Ashes (%)	2.77b	3.19a	0.12
Total acidity (%)	0.22b	0.61a	2.29
рН	6.68a	5.34b	0.88
Proteins (%)	31.92b	34.31a	1.35
Lipids (%)	31.46a	27.33b	3.77
Carbohydrates (%)	28.59b	34.06a	4.33

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Note: The averages in the row followed by the same letter do not differ statistically from each other by the Tukey test at 5% probability.*CV: Coefficient of Variation.

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121 Both powders (in natura and freeze-dried) showed similar solubility values. The rehydration 122 capacity of dry products is of fundamental importance to characterize the quality of products that will be reconstituted, so that the absorption must be fast and in the largest possible 123 volume in order to increase the yield of the products [21]. The powders also presented high 124 hygroscopicity, being defined as the ability of the powder to absorb water from an 125 environment of relative humidity higher than equilibrium. Therefore, depending on the use of 126 the powders, the high hygroscopicity of the powders is difficult to use the product due to high 127 128 affinity for water and because of its complex composition. High hygroscopicity is undesirable for a powdered product when used in food production in view of promoting sticky 129 appearance and hinders solubility of the product, which impairs the quality of the product as 130 a whole, the powder being used in the treatment of water this may not be a problem. 131 132 According to Tonon [11], a higher hygroscopicity can be observed in powders with lower 133 humidity, due to the difference of the water concentration gradient between the product and 134 the environment, however this behavior was not verified in the obtained results, which show 135 a higher hygroscopicity in the powder in natura whose humidity value was higher.

136 The lyophilized powder had low content and water activity, and it may be possible to prolong 137 the shelf life of the lyophilized powder by inhibiting the growth of microorganisms and enzymatic activity, without exposing them to high temperatures and, as a result, greater 138 139 preservation of nutritional quality and sensory characteristics [22]. The powder in natura also 140 presented low moisture content, thus, both powders presented values within those required 141 by current legislation, RDC nº 270 - ANVISA, which describes the maximum acceptance limit 142 of 15% [23]. As for the water activity, the powder in natura showed intermediate water 143 activity, which may hinder the growth of fungi and bacteria.

144 Regarding the analyzes of the physicochemical constituents of the powders, it was observed 145 that there was difference for all parameters evaluated. The ash content in the lyophilized 146 samples was higher than in the in natura samples. In a study carried out by Passos et al. 147 [24] with the in natura moringa powder, the authors found 0.95% of ash, much lower than 148 that found in this research. The powders presented low acidity, meeting the requirements of Brazilian legislation, which determines a minimum of 0.8% acidity in citric acid [23]. In a 149 study on pulps marketed in Alagoas, Temóteo et al. [25] observed acidity of 0.94% in citric 150 151 acid for lyophilized acerola pulp powder, values that are in accordance with current 152 legislation and are superior to that found in this study for Moringa oleifera powder. Different results (3.18%) were also found by Oliveira [26] when the cassava pulp was dried by freeze-153 154 drying.

For the pH it was found that the lyophilized powder was acidic. Passos et al. [24], working with the *in natura* powder of moringa seeds, found a much higher value (7.47%) than this work. Considering the possible toxic effects of microorganisms, when they are at an unfavorable pH, it can be verified that the acid pH value verified in this research is beneficial to lyophilized powder, since it promotes a longer shelf life for the same, without prejudice to its stability.

161 There was an increase in the ash, protein and carbohydrate content of the lyophilized powder and a reduction in the amount of lipids; this is probably due to the addition of distilled 162 water to form the pulp to be lyophilized. Passos et al. [24] found lower values for proteins 163 164 (23.29%) and lipids (17.37%) for the in natura powder of moringa seeds. Basso [27] verified in his study about the chemical composition of the jackfruit that the freeze-drying process did 165 not reduce the amount of ashes, proteins and lipids. Celestino [28] cites as advantages of 166 167 freeze drying the concentration of nutritional components, increasing their value in the 168 product. Affirmative in part is in agreement with Ghribi et al. [29], Oberoi and Sogi [30] and 169 Samoticha et al. [31], who proved in their research the efficiency of the process of freeze-170 drying against the preservation of its constituents.

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173 4. CONCLUSION

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For the physical analysis of the moringa seed powder (apparent density, real and compacted), it was verified that the powder *in natura* presented better results, however, it was observed that for the other physical parameters and physicochemical constituents the freeze-drying promoted the preservation of these in front of the *in natura*.

The freeze-drying of the moringa powder caused a significant reduction in moisture content, water activity, pH and lipids, making it more stable and contributing to the maintenance of its physical-chemical, nutritional and sensorial qualities. The inverse was observed for the ash, protein and carbohydrate contents, where freeze-drying promoted increases in their contents when compared to *in natura*. Freeze-drying presented to the moringa as an adequate method in the preservation of its constituents, with emphasis on physicochemical.

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186 **COMPETING INTERESTS**

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188 Authors have declared that no competing interests exist.

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