

## Review Article

### A review of methods for removal of contaminants in used lubricating oil

#### Abstract

Management and disposing of used lubricating oil (ULO) poses deleterious effects to air, land and water pollution. These contaminants not only causes environmental problems, they also have bio-accumulation effects on living organisms, reduces the ~~inhabitants~~ lifespan ~~of inhabitants~~ as a result of the ~~diseases~~ spread ~~of diseases~~, poisoning and fouling of catalyst as well as corrode processing equipment. ~~The contaminants removal~~ Removal of contaminants in used lubricating oil is a major step to avoid pollution as discussed thoroughly by many researchers in literature. In addition, to curbing pollution, another advantage is converting waste to wealth. This review paper presents insight into various methods for removal of contaminants in used lubricating oil. The advantages and drawbacks of each method were earmarked for further study.

**Keyword:** *Used lubricating oil, Contaminants. Removal Methods, Treatment*

#### Introduction

Lubricating oils (LOs) are conventionally obtained from crude oil. Chemical composition of LOs consists on average ~~of~~ about 80–90% base oil and about 10–20% chemical additives and other compounds (Rincón et al., 2005). Lubricating oils mainly helps in reducing friction, dust, corrosion, protection against wear and tear and provision of heat transfer medium in various equipment or machineries. (Shri et al., 2014). During operation time, LOs deteriorate, as well as their additives, and its physical and chemical properties become unsuitable for further use (Tsai, 2011). In the process of lube oil usage ~~of lube oil~~, temperature build up occurs which breaks down ~~the~~ oil and weakens its properties which include pour point, flash point, specific gravity, viscosity etc. (Udonne and Bakare, 2013). These renders the oil unsuitable for regular usage as results of contaminants in ~~the~~ lube oil such as water, wear metals, carbon residue, ash content, gums, varnishes etc. Chemical changes in ~~the~~ oil occurred due to thermal degradation and oxidation. Europe represents 19% of total worldwide market volume of lubricants, consuming around 6.8 million tons in 2015 (Kupareva *et al.*, 2013).

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Used lubricating oils (ULO) are classified as hazardous wastes, and constitute a serious pollution problem not only for ~~the~~ environment, but also for human health due to ~~the harmful~~ contaminants presence ~~of harmful contaminants~~, such as heavy metals, polychlorinated biphenyls (PCBs) and polycyclic aromatic hydrocarbons (PAHs) (Kanokkantapong *et al.*, 2009). Poor management and careless disposal of used lube oil can affect the environment negatively (Lam and Chase, 2012 and Lam *et al.*, 2016). Scientists have reported that in some geographical region e.g. West Africa, the dispersion of ~~the~~ air pollutants could travel at a speed of 10-12 m/s (Emetere, 2017). The implication of this report is that air pollution from burning of waste lubricant is not localized to the pollution source ~~the source of pollution~~ but could travel with time to other locations. For example, it was recently reported that black soot

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42 covered a metropolitan city of Port Harcourt while remote sources were at the suburb  
43 settlement (-about 22 km away from the city) (Temitayo *et al.*, 2018).

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44 On the other hand, ULOs can be considered as valuable resources, in the sense that it is  
45 possible to recover energy or profitable materials for further use (Guerin, 2008). The best  
46 environmental options, for the management of used lube oils follow the 'waste hierarchy' by  
47 recycling, recovering and then disposing. Used lube oils can be used as an alternative fuel in  
48 a variety of engine configurations and other applications. Its gross calorific value is greater  
49 than 42.9 MJ/kg (Ketlogetswe, ~~C.~~, 1998). The principal objective of any waste management  
50 plan is to ensure safe, efficient and economical collection, transportation, treatment and waste  
51 ~~disposal of waste~~ and as well as satisfactory operation for current and foreseeable future  
52 scenarios (Stoll and Gupta, 1997). The treatment of used lube oil is important due to: (1) It  
53 ~~requires less energy and cost compared to conventional refining of crude oil.~~ (2) It helps in  
54 improving air quality, land and water pollution in the environment. The most preferred option  
55 by experts is the reuse of ~~the~~ used lube oil generated by consumers (Jafari and Hassanpour,  
56 2015). In this paper, a thorough review on various removal and treatment methods for used  
57 lube oil would be considered starting from conventional to ~~the~~ most current methods and their  
58 limitations. ~~;~~ further developments of these fields were also touched. In addition,  
59 environmentally friendly and affordable solvent extraction and adsorbents would be  
60 developed as a means of removing contaminants in used lube oil.

## 61 2. Conventional Methods

62 The conventional methods of contaminants removal ~~of contaminants~~ in used lube oil either  
63 requires a high cost technology such as vacuum distillation or ~~the~~ use of toxic materials such  
64 as sulphuric acid. These methods also produce contaminating by-products which have highly  
65 sulphur levels, especially in ~~the~~ Kurdistan region/Iraq (Hamawand *et al.*, 2013).

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### 66 2.1. Acid-clay

67 Assessment of different contaminants removal ~~of contaminants~~ processes in used lube oils  
68 revealed that acid-clay process had the highest environmental risk and lowest cost. The  
69 method involve treatment of used oil with acid and clay (Udonne and Bakare, 2013;  
70 Hamawand *et al.*, 2013; Abu-Elella *et al.*, 2015). They all used the clay as an adsorbent to  
71 remove the odour and dark colour. What makes acid-clay method unique from others are; with  
72 its simple method, affordable capital investment, low operating cost and does not need skilled  
73 operators (Giovanna *et al.*, 2012; Nwachukwu *et al.*, 2012 and Isah, ~~A.G.~~, 2013).

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74 However, this method has many disadvantages: ~~;~~ It also produces large quantity of pollutants,  
75 is unable to treat modern multigrade oils and it's difficult to remove asphaltic impurities  
76 (Fox, ~~M.F.~~, 2007). To reduce these hazardous contaminants from this method, ~~;~~ the acid  
77 treatment stage of the process can be done under the atmospheric pressure to remove the  
78 acidic products, oxidized polar compounds, suspended particles and additives (Falah and  
79 Hussein, 2011). —Princewill and Sunday, (2010) observed that high recovery rate of treated  
80 lube oil from used lube oil depend largely on the source of ~~the~~ used lube oil, pre-treatment  
81 mechanisms, extent of contamination and the grade of ~~the~~ acid used. He also showed that the  
82 volume of ~~the~~ adsorbent (clay) used could affect the rate at which contaminants are removed  
83 and ~~the~~ recovery percentage ~~of recovery~~ of the method.

84 In Abu-Elella *et al.*, (2015) worked on used motor oil. He treated used motor oil with  
85 phosphoric acid, sulphuric acid, methanoic acid and acetic acid. -He observed that methanoic  
86 acid, sulphuric acid and acetic acid have great changes on ~~the~~ kinematic viscosity while  
87 phosphoric acid is not affected by used lube oil. He therefore concluded that treatment with  
88 acetic acid showed better results than formic acid-clay.

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## 89 2.2 Solvent Extraction

90 This method has replaced acid-clay treatment as ~~the~~ preferred method for improving ~~the~~  
91 oxidative stability and viscosity as well as temperature characteristics of ~~the~~ base oils. Base  
92 oils obtained from Solvent Extraction are of good quality and contains less amounts of  
93 contaminants. In contrast to acid-clay treatment, it operates at higher pressures, requires  
94 skilled operating system and qualified personnel (AERCO 1995). The solvent selectively  
95 dissolves the undesired aromatic components (the extract), leaving ~~the~~ desired saturated  
96 components, especially alkanes, as a separate phase (the raffinate). (Rincon *et al.*, 2005).

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97 Different solvents types ~~of solvents~~ have been used for solvent extraction such as 2-propanol,  
98 1-butanol, methyl ethyl ketone (MEK), ethanol, toluene, acetone, propane etc. (Quang *et al.*,  
99 1974) and (Rincon *et al.*, 2003) used propane as a solvent. He found out that ~~the~~ propane was  
100 capable of dissolving paraffinic or waxy material and intermediately dissolved oxygenated  
101 material. Asphaltenes which contain heavy condensed aromatic compounds and particulate  
102 matter are insoluble in liquid propane. These properties make propane ideal for recycling the  
103 used engine oil, but there are many other issues that have to be considered. Propane is  
104 hazardous and flammable therefore this process is regarded as hazardous method.

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105 Katiyar and Husain (2010); Sterpu *et al.*, (2012) and Hassan *et al.*, (2012) found out that  
106 methyl ethyl ketone has the highest performance due to its low oil percentage losses and high  
107 sludge removal while Hussein *et al.*, (2014) and -Aremu *et al.*, (2015) -found out extraction  
108 using butan-1-ol solvent produces the highest sludge removal rate. (Rincon *et al.*, 2005) and  
109 Oladimeji *et al.*, 2018) used a composite solvent of methyl ethyl ketone and 2-propanol the  
110 oil resulting from this process is comparable to that produced by acid-clay method, its cost  
111 was high.

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112 Solvent extraction, in general, involves solvent losses and highly operating maintenance.  
113 Also, it occurs at pressures higher than 10 atm and requires high pressure sealing systems  
114 which makes solvent extraction plants expensive to construct, operate and the method also  
115 produces remarkable amounts of hazardous by-products. (Quang *et al.*, 1974); (Rincon *et al.*,  
116 2003) and Hamawand *et al.*, (2013).

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117 Mineral Oil Raffinerie Dollbergen (MRD) solvent extraction process using N-methyl-2-  
118 pyrrolidone. The applied oil re-refining process is based on a patent held by AVISTA OIL.  
119 (P'ohler *et al.*, 2004) The 'Enhanced Selective Refining' process uses solvent N-methyl-2-  
120 pyrrolidone (NMP), which is commonly used in ~~the~~ petroleum refining industry. NMP is a  
121 powerful, aprotic solvent with low volatility, which shows selective affinity for unsaturated  
122 hydrocarbons, aromatics, and sulphur compounds. Due to its relative non-reactivity and high  
123 selectivity, NMP finds wide applicability as an aromatic extraction solvent in lube oil re-  
124 refining. The NMP advantages ~~of NMP~~ over other solvents are ~~the~~ non-toxic nature and high  
125 solvent power, absence of azeotropes formation with hydrocarbons, the ease of recovery from  
126 solutes and its high selectivity for aromatic hydrocarbons. Being a selective solvent for

127 | aromatic hydrocarbons and PAH, NMP can be used for ~~the~~ re-refining of waste oils with  
128 | lower sludge, carbonaceous particles and polymer contents, such as waste insulating,  
129 | hydraulic and other similar industrial oils.( Lukic, ~~J~~ *et al.*, 2005). The MRD solvent  
130 | extraction process uses the liquid–liquid extraction principle.

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131 | The average base oil yield within the process is about 91 %.( Schiessler, ~~N~~ *et al.*, 2007). The  
132 | base oils produced have high quality (Kupareva *et al.*, 2013). The process is characterized by  
133 | optimized operating conditions which allow elimination of toxic polyaromatic compounds  
134 | from the re-refined base oil and preservation of ~~the~~ synthetic base oils like polyalphaolefin  
135 | (PAO) or hydrocracked oils, which are increasingly present in used oils. However, this  
136 | method need skilled personnel, proper disposal and management of it waste.

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### 137 | 2.3 Vacuum distillation

138 | Extensive research work have been done on vacuum distillation on used oil by the following  
139 | Martins, ~~J.P.~~ (1997); Shakirullah *et al.*,2006; Bridjanian and Sattarian (2006); Emam and  
140 | Shoaib, (2012); Hamawand *et al.* (2013) and Kannan *et al.*, (2014). In this method, used  
141 | lube oil collected is heated at a temperature of 120°C to remove the water added to the oil  
142 | during combustion. Then the dehydrated oil is subjected to vacuum distilled at a temperature  
143 | of 240°C and pressure 20 mmHg. This results ~~to~~ the production of ~~a~~ light fuel oil at a  
144 | temperature of 140°C (the light fuel oil can be used as fuel source for heating) and lubricating  
145 | oil at 240°C. The lubricating oil vapour is condensed and sent for next stage. (Kannan *et al.*,  
146 | 2014). The advantages of vacuum distillation process over atmospheric pressure distillation  
147 | are: ~~C~~eolomns can be operated at lower temperatures; ~~M~~more economical to separate high  
148 | boiling point components under vacuum distillation; ~~A~~void degradation of properties of  
149 | some species at high temperatures therefore thermally sensitive substances can be processed  
150 | easily.

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151 | However, the remaining oil generated at this temperature (240°C) contains the dirt, degraded  
152 | additives, metal wear parts and combustion products like carbon and is collected as residue.  
153 | ~~The residue is in the form similar to that of tar,~~ which can be used as a construction material,  
154 | for example, road and bitumen production. (Giovanna *et al.*, 2003). The disadvantage of this  
155 | method is the high investment cost and/or the use of toxic materials such as sulphuric acid.  
156 | (Havemann, 1978 and Puerto-Ferre, & Kajdas, 1994).

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### 157 | 2.4 Hydrogenation

158 | To avoid formation of harmful products and environmental issues based on above methods,  
159 | some modern processes have been used and the best one is hydrotreating. (Bridjanian and  
160 | Sattarin, 2006). This method follows vacuum distillation. In this process, the distillate from  
161 | vacuum distillation is hydrotreated at high pressure and temperature in the presence of  
162 | catalyst for the purpose of removing chlorine, sulphur, nitrogen and organic components. The  
163 | treated hydrocarbons resulted in products of improved odour, chemical properties and colour.  
164 | (Temitayo *et al.*, 2018).

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165 | Another important aspect of this method is that, this process has many advantages: Produces  
166 | of high Viscosity Index lube oil with well oxidation resistance and a good stable colour and  
167 | yet having low or no discards. At the same time, it consumes bad quality feed. In addition to  
168 | that, this method has advantage that all of its hydrocarbon products have good applications

169 and product recovery is high with no (or very low) disposals. Other hydrocarbon products  
170 are: In oil refinery the light—cuts can be used as fuel in—the plant itself. Gas oil may be  
171 consumed after being mixed with heating gas oil and the distillation residue can be blended  
172 with bitumen and consumed as—the paving asphalt, because it upgrades a lot its rheological  
173 properties. Also, it can be used as a concentrated anti-corrosion liquid coating, for vehicles  
174 frames. (Hassan—A—D, 2014).

175 The disadvantage of this method is that the residue resulting from the process is of high  
176 boiling range of hydrocarbon product fractionated into neutral oil products with varying  
177 viscosities which can also be used to blend lube oil (Basel Convention, 2002).

## 178 2.5 Membrane Technology

179 Membrane technology is another method for removal of contaminants of used lubricating  
180 oils. In this process, three types of polymer hollow fibre membranes [polyethersulphone  
181 (PES), polyvinylidene fluoride (PVDF), and polyacrylonitrile (PAN)] (Lam *et al.*, 2016) were  
182 used for recycling the used engine oils. The process is carried out at 40°C and 0.1 MPa  
183 pressure. The process is a continuous operation as it removes metal and particles and dusts  
184 from used lube oil and improves the recovered oils liquidity and flash point. (Dang—C—S,  
185 (1997) and Hamawand *et al.*, (2013).

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186 Despite the above mentioned advantages, the expensive membranes may get damaged and  
187 fouled by large particles with time. (Dang—C—S, (1997) and Hamawand *et al.*, (2013).

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## 188 2.6 Catalytic Process

189 For example, Hylube process from Germany. This process allows production of mainly base  
190 oils. The Hylube process is a proprietary process developed by Universal Oil Products (UOP)  
191 for—the catalytic processing of used lube oils into re-refined lube base stocks for re-blending  
192 into saleable lube base oils (Kalnes *et al.*, 2006). This is the first re-refining process in which  
193 as received used oil is processed, without any pre-treatment, in a pressurized hydrogen  
194 environment. A typical HyLube process feedstock consists of a blend of used lube oils  
195 containing high concentrations of particulate matter such as iron and spent additive  
196 contaminants such as zinc, phosphorous, and calcium (Chari—K—R, 2012).

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197 The Hylube unit operates with reactor section pressures of 60–80 bar and reactor  
198 temperatures in the range 300–350°C (Kalnes—T—N and Schuppel—A, 2007).— The Hylube  
199 process achieves more than 85% of lube oil recovery from the lube boiling range  
200 hydrocarbon in—the feedstock (Kupareva *et al.*, 2013). Besides the advantages of these  
201 process, this method is very expensive. This method requires high level personnel due to high  
202 temperature and pressure operations.

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## 203 3. Combined Technologies/methods

204 These are advance methods that combines two or more generic methods in its process. Due to  
205 the complex nature of contaminants removal ~~of contaminants~~ in used lube oils, using a single  
206 method may not give you the desired standard ~~emission-controlled~~ emission-controlled  
207 process. Therefore, some companies have developed specific processes for treatment and  
208 contaminants removal ~~of contaminants~~ in used lube oils (Basel Convention, 2002; Brinkman,  
209 2010 and Kupareva *et al.*, 2013), these methods require sophisticated technologies,  
210 equipment and processes. Some of these complex processes are briefly discussed below;

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### 211 3.1 Vaxon process

212 -This process contains chemical treatment, vacuum distillation and solvent refining units. The  
213 advantage of ~~the~~ Vaxon process is the special vacuum distillation, where the cracking of oil is  
214 strongly decreased. (Chari, ~~K.R.~~, 2012).

215 The chemical final stage does not, however allow the high-quality base oils production ~~of~~  
216 high-quality base oils; although in Spain the Catalonia refinery produces base stocks accepted  
217 by an original equipment manufacturer (OEM). In connection with this fact, the lube distillate  
218 obtained from ~~the~~ Vaxon process (Denmark) or North Refining (Netherlands) are precursors  
219 for ~~the~~ Avista Oil base. (Kupareva *et al.*, 2013).

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### 220 3.2 CEP process

221 -This process combines thin film evaporation and hydroprocessing. The used oil is chemically  
222 pre-treated to avoid precipitation of contaminants which can cause corrosion and fouling of  
223 the equipment. The pre-treating step is carried out at temperatures within 80–170°C. The  
224 chemical treatment compound comprises sodium hydroxide, which is added in a sufficient  
225 amount to give a pH about 6.5 or higher. (Magnabosco ~~L.M.~~ and Rondeau ~~W.A.~~, 1993).

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226 Heavy materials (such as residues, mMetals, additive degradation products, etc.) are passed to  
227 a heavy asphalt flux stream. The distillate is hydropurified at high temperature (315°C) and  
228 pressure (90 bar) in a catalytic fixed bed reactor. (Merchaoui *et al.*, 1994). This process  
229 removes nitrogen, sulphur, chlorine and oxygenated organic components. In ~~the~~ final stage of  
230 ~~the~~ process, three hydrotreating (Hydrofinishing) reactors are used in series to reduce sulfur  
231 to less than 300 ppm and to increase the amount~~number~~ of saturated compounds to over 95%,  
232 in order to meet the key specifications for API Group II base oil. The final step, in this  
233 process is vacuum distillation to separate the hydrotreated base oil into multiple viscosity cuts  
234 in ~~the~~ fractionator. The yield of base oils is about 70%. (Kupareva *et al.*, 2013).

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### 235 3.3 Ecohuile process

236 The re-refining process was based on vacuum distillation and acid-clay treatment steps until  
237 the end of 2000. (Audibert, ~~F.~~, 2006). Clay adsorption was banned on 1 January 2001 and the  
238 plant was modified and upgraded to the Sotulub process. (Sotulub re-refining process. 2005).  
239 Moreover, the addition of injection facilities of so-called Antipoll-additive (1–3 wt% of pure  
240 sodium hydroxide) has been provided and has allowed solving the following basic problems:

- 241 • Corrosion of dehydration column and cracking column top section due to ~~the~~ organic  
242 acidity of ~~the~~ used oil;
- 243 • Plugging of equipment and piping due to polymer formation in ~~the~~ cracking section;
- 244 • High losses of base oil in ~~the~~ oily clay due to ~~the~~ high consumption of clay.

245 The Sotulub process (Merchaoui, ~~M.H.~~ *et al.*, 1994) is based on treatment of ~~the~~ used oil with  
246 an alkali additive called Antipoll and high vacuum distillation. The used oil is pre-heated to  
247 about 160°C and mixed with a small amount of Antipoll-additive, which decreases equipment  
248 fouling. This process, allows a final product to be obtained with acceptable quality without  
249 any additional finishing stage. Oil obtained is additionally fractionated to obtain various base  
250 oil cuts. The process provides base oils with a yield of 82–92%. (Sotulub re-refining process.  
251 2005).

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### 252 3.4 Cyclon process

253 | -This process combines the technology of vacuum distillation and hydrofinishing.  
254 | (Havemann, 1997). The process licence belongs to Kinetic Technology International (KTI).  
255 | (Kajdas, ~~C~~, 2000). In this process, used oils taken from storage tanks are dewatered and the  
256 | light hydrocarbons are removed by distillation. The heavier fraction is sent to high vacuum  
257 | distillation, where the majority of base oil components are evaporated from the heavy  
258 | residue. The oils in ~~the~~ residues are extracted with propane in ~~the~~ de-asphalting unit and sent  
259 | to ~~the~~ hydroprocessing unit where ~~the~~ other oils are processed. Then they are treated with  
260 | hydrogen and fractionated based on ~~the~~ desired base oil features. The re-refined base oil  
261 | products obtained are of high quality due to ~~the~~ hydrogenation. (Schiessler, ~~N~~, *et al.*, 2007  
262 | and Tsalavoutas, ~~S~~, *et al.*, 2002)

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### 263 3.5 STP method

264 | This is another advance method that combines vacuum distillation and hydrofinishing process  
265 | (Basel Convention, 2002). It produces less harmful pollutants therefore its environmentally  
266 | friendly (Kupareva *et al.*, 2013). This method involves dehydration, vacuum distillation,  
267 | separation of ~~the~~ lubricating fraction and hydrofinishing of base oil separation from ~~the~~  
268 | residue.

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### 269 3.6 Interline process

270 | -Interline proposes a process based on propane de-asphalting at ambient temperature and  
271 | under a pressure that facilitates separation in ~~the~~ liquid phase. The lubricating oil yield  
272 | declared for ~~the~~ Interline process is 79 %.(Monier, ~~V~~ and Labouze, ~~E~~, (2001); European IPPC  
273 | Bureau, Spain (2003)- and Aramburu, ~~J.A~~, (2003). The extraction process removes the  
274 | majority of additives. The process is interesting from ~~the~~ economics point of view because it  
275 | eliminates thin film distillation and ~~the~~ need for hydrogenation. Both investment and  
276 | maintenance costs are low.

277 | The drawbacks of ~~the~~ Interline process are that ~~the~~ feed should not contain polychlorinated  
278 | biphenyls (PCBs), and its chlorine content should be below 1000 ppm, since this process has  
279 | no final hydrofinishing step.

### 280 3.7 Propak thermal cracking process

281 | The Propak process consists of screening and dewatering sections, followed by a thermal  
282 | cracking section, a separation or distillation depending on ~~the~~ product state desired and  
283 | finally purification and stabilization stages. In certain plant configurations, a heavy boiling  
284 | fraction is recycled back to ~~the~~ fired process heater. Gasoil in ~~the~~ liquid state is led to ~~the~~  
285 | stabilization section from distillation.

286 | This technology is characterized by a large operational and product flexibility. Process  
287 | operating conditions (temperature, pressure, residence time) can be varied to produce a  
288 | desired product such as heavy fuel oil, gasoil or base oil. (Kupareva *et al.*, 2013)

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### 289 4.0 Current Technologies for Used Oil Re-refining

290 | Used lube oil normally tends to have a high concentration of potentially harmful pollutant  
291 | materials and heavy metals which could be dangerous to both living and non-living things on  
292 | the earth. Used lube oil may cause damage to ~~the~~ environment when dumped into ~~the~~ ground

293 or into water streams including sewers. This may result in ground water and soil  
294 contamination (Hopmans, 1974). Therefore, development of environmentally safe,  
295 sustainable and cost-effective solution is required for recycling of used lubricant. (Stehlik,  
296 2009).

297 Nowadays due to different treatment and finishing methods, there are currently available  
298 many new technologies (Bridjanian, H and Sattarin, M., 2006) such as pyrolytic distillation  
299 method (PDM), pyrolysis process (PP), thin film evaporation (TFE), including combined  
300 TFE and clay finishing, TFE and solvent finishing, TFE and hydrofinishing, thermal de-  
301 asphaltting (TDA), TDA and clay finishing, TDA and hydrofinishing etc. In addition,  
302 environmentally friendly and affordable solvent extraction and adsorbents are being  
303 developed as a means of removing contaminants in used lube oil. Some of the current  
304 methods are briefly discussed below.

305 From the research conducted by Arpal *et al.*, (2010), a fuel named as diesel-like fuel (DLF)  
306 was produced by applying pyrolytic distillation method. Filtration of the waste engine oil  
307 sample was done using a quantitative filter. Three additives known as Na<sub>2</sub>CO<sub>3</sub>, zeolite and  
308 CaO were blended with the purified oil at different ratios and were exposed to thermal and  
309 pyrolytic treatment to convert them into a diesel-like fuel. Conclusively, effects of DLF on  
310 the oil properties shows a closer range to that of diesel fuel. (Temitayo *et al.*, 2018)

311 Also, Pyrolysis process (PP) has been used as an alternative means of effective conversion of  
312 used lubricants to a refined one (Lam *et al.*, (2016); and Manasomboonphan and Junyapoon,  
313 (2012). Lam *et al.*, (2016), describe pyrolysis as a thermal process that heats and decomposes  
314 substance at high temperature (300-1000<sup>o</sup>C) in an inert environment without oxygen.  
315 Pyrolysis process is not yet widespread but it has been receiving much attentions nowadays  
316 due to its potential to produce energy-dense products from materials. Examples of pyrolysis  
317 process includes Microwave Pyrolysis Process (MPP) and Conventional Pyrolysis Process  
318 (CPP). The MPP is a thermo-chemical process applied to waste to wealth process of electrical  
319 power input of 7.5kW at a flow rate of 5kg/h. (Temitayo *et al.*, 2018).

320 Thin film evaporation technology includes a rotating mechanism inside the evaporator vessel  
321 which creates high turbulence and thereby reduces the residence time of feed-stock oil in the  
322 evaporator. This is done in order to reduce coking, which is caused by cracking of the  
323 hydrocarbons due to impurities in the used oil. Cracking starts to occur when the temperature  
324 of the feedstock oil rises above 300<sup>o</sup>C.

325 However, any coking which does occur will foul the rotating mechanism and other  
326 mechanisms such as tube-type heat exchangers are often found in thin film evaporators.  
327 Solvent extraction processes are widely applied to remove asphaltic and resinous  
328 components.

329 Liquid propane is by far the most frequently used solvent for de-asphalting residues to make  
330 lubricant bright stock, whereas liquid butane or pentane produces lower grade de-asphalted  
331 oils more suitable for feeding to fuel-upgrading units. The liquid propane is kept close to its  
332 critical point and, under these conditions, raising the temperature increases selectivity. A  
333 temperature gradient is set up in the extraction tower to facilitate separation. Solvent-to-oil  
334 ratios are kept high because this enhances rejection of asphalt from the propane/oil phase.

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335 Counter-current extraction takes place in a tall extraction tower. Typical operating conditions  
336 can be found in the work by (Mortier and Fox, 2010).

337 Recent studies showed that propane can be replaced by an alcohol–ketone mixture, which  
338 reduces coking and fouling problems during distillation. (Kamal and Khan, 2009 & Durrani  
339 *et al.*, 2010). The solvent chosen should meet the following requirements: *M*maximum  
340 solubility for the oils and minimum solubility for additives and carbonaceous matter; ability  
341 to be recovered by distillation. New plant units increasingly use N-methylpyrrolidone  
342 because it has the lowest toxicity and can be used at lower solvent/oil ratios, saving energy.  
343 Independent of ~~the~~ contacting method used, the end result is two product streams. The  
344 raffinate stream is mainly extracted oil containing a limited amount of solvent, while the  
345 extract stream is a mixture of solvent and aromatic components. The streams are handled  
346 separately during solvent recovery and ~~the~~ recovered solvent streams are recombined and  
347 recycled within the plant.

348 However, solvent recovery is an energy-intensive part of ~~the~~ solvent extraction process. For  
349 several years, catalytic hydrotreatment stood out as ~~the~~ modern and successful refining  
350 treatment from the point of view of ~~the~~ yield and quality of ~~the~~ finished products.  
351 Hydroprocessing is more often applied as a final step in ~~the~~ rerefining process in order to  
352 correct problems such as poor colour, oxidation or thermal stability, demulsification and  
353 electrical insulating properties. (Kupareva *et al.*, 2013).

354 In hydrofinishing, used oil and hydrogen are pre-heated and then oil allowed to trickle  
355 downwards through a reactor filled with catalyst particles where hydrogenation reactions take  
356 place. The oil product is separated from ~~the~~ gaseous phase and then stripped to remove traces  
357 of dissolved gases or water. Typical reactor operating conditions for hydrofinishing can be  
358 found (Mortier and Fox, 2010).

359 The following reactions can be operative: *H*hydrorefining reactions with ~~the~~ objective of  
360 removing heteroelements and to hydrogenate olefinic and aromatic compounds, and  
361 hydroconversion reactions aiming at modifying the structure of hydrocarbons by cracking  
362 and isomerization. (Audibert, 2006).

363 Hydrotreatment catalysts are made of an active phase constituted by molybdenum or tungsten  
364 sulfides as well as by cobalt or nickel on oxide carriers. Generally applied combinations are  
365 Co-Mo, Ni-Mo, and Ni-W for ~~the~~ active phase and high surface area  $\gamma$ -alumina (transition  
366 alumina) carrier. The metal content, expressed as oxides can reach 12–15 wt. % for Mo and  
367 3–5 wt. % for Co or Ni. Co-Mo catalysts are preferentially used for hydrodesulphurization  
368 and Ni-Mo for hydrogenation and hydrodenitrogenation. Ni-W catalysts are applied for low-  
369 sulphur feeds. The most-used carriers are alumina and alumina-silica, the latter being  
370 characterized by a higher cracking activity. (Audibert, 2006).

371 The currently applied catalysts in rerefining are modified in order to improve ~~the~~ product  
372 base oil quality and to decrease the coke formation, however, their composition is typically  
373 not disclosed in an open literature. The technologies applying hydroprocesses require  
374 relatively high investments compared with others. However, depending on ~~the~~ technology  
375 adopted, the total cost might be lower than in solvent extraction process due to the high  
376 operating costs to make up for ~~the~~ solvent losses. On ~~the~~ other hand, solvent extraction and  
377 chemical treatment processes do not require catalyst regeneration. Moreover, it is not

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378 necessary to establish a hydrogen gas supply facility in these methods which in addition  
379 reduces a risk concerning operation safety. (Kupareva *et al.*, 2013).

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## 380 5.0 Conclusion

381 Currently applied technologies can be compared in terms of their operating and capital costs,  
382 quality of feedstock and products obtained. These advance combine technology processes  
383 and/or methods are mainly found in developed countries but not available in developing  
384 countries. These methods when applied generates reduced concentrations of pollutant but  
385 require complex and expensive equipment which are rarely found in developing countries.  
386 Under increasing environmental pressure of ~~the~~ conventional treatment method such as acid-  
387 clay treatment, which was the first oil regeneration process used, it was substituted in ~~the~~  
388 majority of European countries with new technologies based on solvent extraction, pyrolysis,  
389 membrane etc. The modern technologies based on solvent extraction, pyrolysis, membrane  
390 etc. are environmentally controllable but their operating and capital costs are high, low yields  
391 and requires highly skilled personnel (–compared to conventional method) is the major  
392 drawback. Also, the challenge of cost reduction resulting from ~~the~~ vacuum distillation and  
393 hydroprocessing technique. The combined treatment methods have shown remarkable well  
394 with high treatment efficiency, environmentally friendly. However, the problem of high cost  
395 and season skilled operating personnel remains a major gap in used lube oil treatment.  
396 Therefore, there is the need to developed viable, efficient, environmentally friendly,  
397 affordable treatment and high yield technique such as solvent extraction coupled with  
398 adsorption process to remove contaminants in used lube oil.

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