		Effects on Human Who ctroscopy in Vitro Stu
ABSTRACT		
FTIR and UV-Vis spetthe serum of whole block	ctroscopic technique is o bod samples	een used numerously in medical applicati employed to study the spectral difference
irradiated to different t Study design: Huma Place and Duration of	imes from 10 min to 50 n n Whole Blood Irradiated	I to (He-Ne) laser(λ = 632nm, power=2mWer, Sudan university of science and techno
Methodology: Blood	samples were collect er and control compare	ed from healthy volunteers; blood sar d; UV-vis spectrophotometer and FTIR v
He-Ne laser radiation exposed blood show group), and C-H (aron transmittance in FTIR increases the most e	i shows, a significant d the peaks due to O-H (f natic group). N-H (Amin spectra for C=O group a ffect are found when w	ctra of whole are compared before and a becrease in intensity. FTIR spectrum of free group), C=O (amide I group), N=O (o acid (amide II) Laser radiation change and O-H, N=O, percentage of transmitta hole blood irradiated to He-Ne laser radia ses for C-H, and N-H, due to denaturation
Conclusion: Photode		ponents due to absorption of laser radia ormational changes in the polypeptide

Keywords: Laser, blood, UV-Vis, FTIR, spectroscopic

14 **1. INTRODUCTION** 15

- Low-intensity helium neon laser has been used extensively in medical applications 16
- 17 lately. Interaction of lasers with biological materials such as blood, skin, and tissues
- 18 is an important to be understood. The study of blood change by spectroscopic
- 19 techniques can be used for understanding the biological nature of the disease, and
- 20 also for the diagnosis of the disease. [1,2]
- 21 Photobiomodulations involves exposing tissues to low level light. This type of
- 22 therapy called Low level laser therapy (LLLT), also known as cold laser therapy as

the power densities used produces no heating effect on the tissues. LLLT has a
photochemical affect which means the light is absorbed and cause a chemical
change. [3,4,5]

FTIR and UV-Vis spectroscopic technique is employed to study the spectral 26 27 differences in the serum of normal blood samples[2],Blood samples were irradiated by He-Ne laser (Wavelength λ = 632.8 nm, Power = 3mW). The FTIR spectra for 28 29 FTIR spectra of irradiated blood samples. Show significant changes.[1] He Ne laser 30 $(\lambda = 632$ nm, power=2mW) is used to irradiate human Red blood cells Absorption 31 spectrum, FTIR and fluorescence spectra of RBC The absorption spectrum of RBC 32 after exposure to He-Ne laser shows a significant decrease in absorbance. The FTIR spectrum of irradiated RBC clearly show changes in transmittance, [6] some 33 rheological factors of the human blood, such as complete blood count (CBC) 34 35 parameters and blood sedimentation rate (BSR) effected by low-level laser radiation 36 (LLLR) laser blood biostimulations investigated the effect of LLLT on rheological parameters of human blood, they noticed a change in both viscosity and size 37

- of erythrocytes, [7].human blood exposed to low-intensity He–Ne-laser radiation causes clearly defined changes in the IR and visible absorption spectra of the blood and erythrocytes. These spectral changes arise as a result of partial photo
- 41 dissociation of hemoglobin–ligand[8]

this paper investigate the effect of He-Ne laser (Wavelength λ = 632.8 nm, Power = 2mWith different exposure time using UV-Vis spectrophotometer and FTIR

44 spectrometer.

45 2. MATERIAL AND METHODS

46 **2.1 Samples Collection**

Blood samples were taken from healthy volunteers; 3 ml of each volunteer by medical standard laboratory conditions and blood samples were saved in tube to prevent from coagulation to (EDTA) and each sample was divided into two samples one sample was control and other exposed to helium-neon laser with different exposure times.

51 2.2 laser irradiated

52 Samples were exposed to a Helium-Neon laser beam, operating in continuous wave mod, as 53 a radiation source (632.8 nm, 2 mW), for (10, 20, 30, 40 and 50) minutes The distance 54 between the laser source and the samples was set to be 10 cm and the diameter of laser 55 spot was chosen to be 1.5 cm.. to studied the effect of laser radiation were used UV-vis 56 spectrophotometer (Jasco-670) and Fourier Transform Infra Red Spectra (FTIR) were 57 obtained used FTIR spectrophotometer (shimadzo) for control, and He-Ne laser irradiated 58 blood serum samples.

59 3. RESULTS AND DISCUSSION

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61 3.1 UV-vis spectra

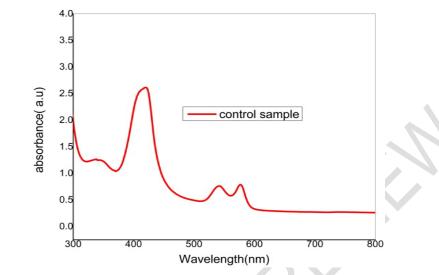
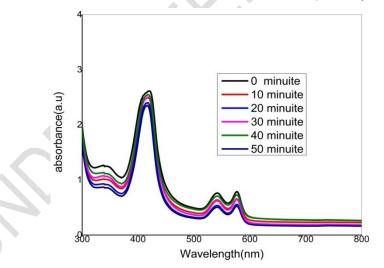




Figure .1 Spectrum of non- irradiated blood sample (control).

Figure .1 shows the spectrum of non- irradiated blood sample (control). This spectrum referred to non- irradiated blood sample which specified by peaks at (576.0, 542.0, 416.0 and 340.0) nm with intensities 0.793, 0.755, 2.604 and 1.253 respectively.



67

68 Figure 2. Relation between Absorbance (a) and wavelength (λ) for whole blood before

69 and after irradiated to (He-Ne) laser power 2 mW

- The absorption spectra of the whole blood recorded in the range of 300–800 nm Figure 2.
- 71 Contain absorption bands with λ_{max} = 340, 416 nm, a doublet band with λ_{max} = 542 and 576
- nm. We investigated only those changes in the absorption spectra of the whole blood
- radiation that were detected for all of the samples studied.

Table.1 The intensity of normal and irradiated samples

Wave	Absorbance a.u					
number 1/cm	control	10 min	20 min	30 min	40 min	50 min
340	1.253	1.01	0.933	1.065	1.12	0.868
416	2.604	2.49	2.391	2.501	2.538	2.347
542	0.755	0.633	0.536	0.614	0.699	0.492
576	0.793	0.653	0.547	0.633	0.718	0.525

Different serum samples are analyzed quantitatively by calculating the intensities among the absorption peaks which is show decrease intensity, all irradiated serum sample less than control serum sample. These results indicate to that there is photo degradation happened to the blood components.

Absorption intensity slightly decreases for all peaks at, due to increasing ligand electronegativity [9].

In the UV-visible absorption spectrum of the irradiated blood, (figure.2 and table1) the most intense absorption band at 416 nm, the light with this wavelength that strikes this biological tissues will be highly absorbed. This phenomenon is the key for the desired effect on the tissues [10]. Figure 2 compared the light absorption at 340nm, 414nm, 542nm and 576nm for different irradiation time. The minimum light absorption occurred at 50 minutes of irradiation with the less intensities recorded.

88 3.2 FTIR spectra

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Table .2 FTIR spectral data (wave number, function group and transmission) fornormal blood control

FTIR spectral data for normal blood (control)					
Sr. No	Wave number 1/cm	Group	% T		
1	3444.63	O-H	0.48		
2	1650.95	C=0	1.19		
3	1548.73	N=O	6.36		
4	1452.30	C-H	14.26		
5	1317.29	N-H	15.3		
6	1168.78	C-O	17.12		

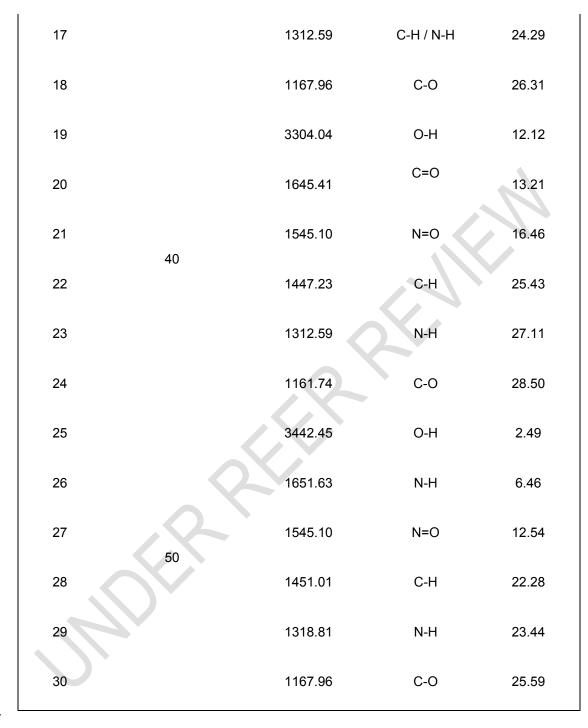
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95 Table .3 FTIR spectral data (wave number, function group and transmission) for

96 irradiated blood sample blood control

FTIR spect	rum of blood irradiated	d with he-ne laser for	duration 10, 20), 30,40 and 50
min				
Sr. No	Irradiated Time	Wave number	Group	Т%
	(minute)	1/CM		
1		3396.77	O-H	0.77
2		1650.96	C=O	1.78
3		1545.10	N=O	4.49
4	10	1450.73	C-H	15.20
5		1312.59	N-H	16.12
6		1161.74	C-0	18.70
7		3442.45	O-H	0.65
8		1651.63	C=O	1.68
9		1545.10	N=O	4.68
10	20	1451.01	C-H	11.43
11		1312.59	N-H	12.58
12		1161.74	C-0	13.76
13		3410.57	O-H	4.92
14	20	1651.63	C=O	6.50
15	30	1551.23	N=O	12.82
16		1451.01	C-H	22.14



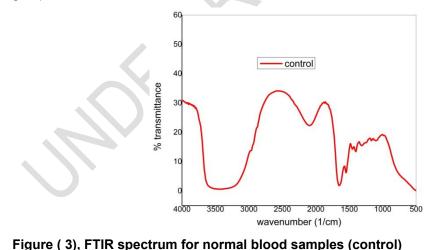
An FTIR spectrum of whole blood in vitro without laser radiation is shown in (Figure) 3.
Table2. Shows the groups OH, C=O, N=O, C-O and C-H in the region between the wave
number 4000 1/cm to 500 1/cm. The most intense absorption band in proteins is the amide
I peak, which is observed at 1650.95 1/cm. Amide I is mainly associated with C=O

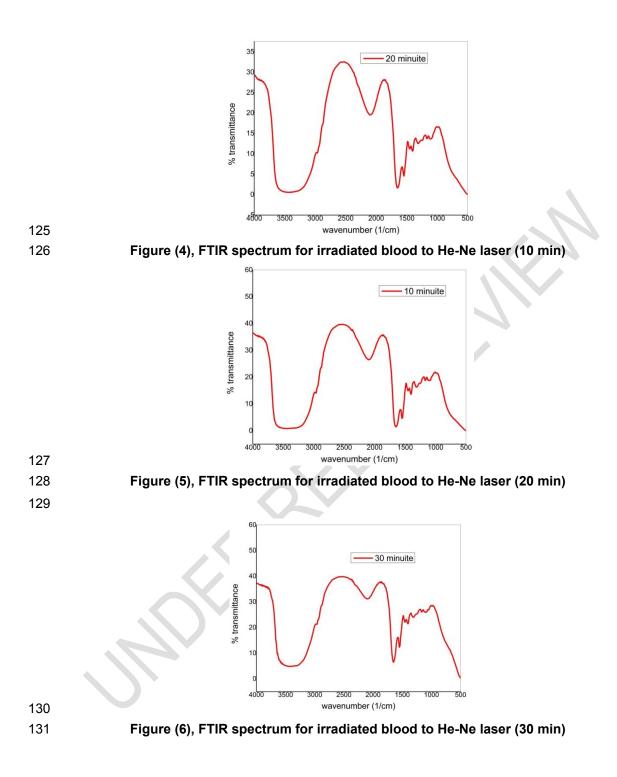
102 and or C-O stretching vibrations. There are anther very strong symmetric stretching 103 prominent amide absorptions one at at 1545 1/ cm due to strong N-H in plane bending and 104 termed as an Amide II band. The strong characteristic band at 3295 1/cm due to N-H 105 symmetric stretching confirmed the existence of amino acid group [2] The medium band at 106 2873 1/cm due to C-H asymmetric and symmetric stretching of CH3 group established the 107 presence of lipids and the medium bands at 2854 1/ cm due to C-H symmetric stretching of 108 CH2 group established the presence of lipids, fatty acids[11,12,13,14]. The FTIR spectra of 109 blood showed clear bands at1080, and 12451/ cm, are composed of mononuclear cells 110 containing nucleic acids such as DNA and RNA. The nucleic acid components found in 111 WBCs.[9]

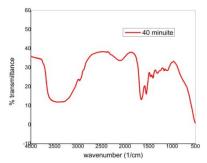
Whole blood sample is irradiated to He-Ne laser radiation for 10, 20,30,40min. and 50 min
duration respectively, figure (4 to 8) table 3. Shows the groups associated with spectral
peaks whole sample irradiated to

He-Ne laser radiation for 10 min duration shows increase in transmittance for all groupsexcept for C-H dicreases due to denaturation of protein .

FTIR spectra of whole blood irradiated with He-Ne laser for 20 minuite show decreases in transmission for group, C-H, and N-H, to denaturation of protein. i.e. it breaks the polypeptide bonds due to conformational changes of proteins , But in 30, 40 ,50 minutes show increase in transmittance for all groups for all groups is observed. Laser irradiation of blood causes changes in absorption band in stretching and bending Vibrations of peptide group.







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Figure (7), FTIR spectrum for irradiated blood to He-Ne laser (40 min)

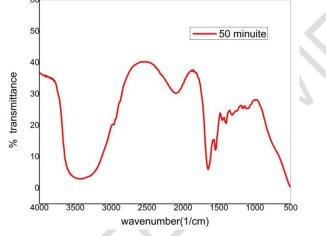


Figure (8), FTIR spectrum for irradiated blood to He-Ne laser (50 min)

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138 4. CONCLUSION

140 It has been shown that laser radiation effect on blood at the molecular level. Hemoglobin is a 141 blood photoacceptor that selectively absorbs He-Ne Laser radiation power 2mW, (632.8 142 nm). The absorption of laser radiation by blood leads to partial photodissociation. show 143 decrease intensity, all irradiated serum sample less than control serum sample This result 144 indicate to that there is photodegradation happened to the blood components. This causes 145 changes in the structure and conformational changes in the polypeptide of N-H and CO and 146 COO– groups in the regions 1500–1700 and 3000–3500 cm–1 of the IR spectrum.

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