

1 **Degradative Effect of I.R radiations on the Constituents of Bitumen**

2 **Abstract**

3 Sample of natural bitumen were taken from bitumen well in Agbabu town in
4 Odigbo Local Government of Ondo State. These samples were separately
5 irradiated with ultraviolet, infrared and X-ray radiations for a period of seven
6 hours. Part of the sample was withdrawn at interval of One, Three and Seven
7 hours. The withdrawn sample was later separated into maltene and asphaltene
8 fractions. The maltene fraction was further separated into saturated, aromatic and
9 polar fraction. The saturated and aromatic fractions were subjected to gas
10 chromatography analysis. The Saturated and aromatic profiles of the bitumen were
11 found to vary with the period of irradiation. Both the amount of saturated and
12 aromatic compounds in the bitumen decreased with the period of irradiation. Thus,
13 the radiations were found to have a degradative effect on the composition of
14 bitumen.

15 **Keywords: Bitumen, radiations, Gas Chromatography, degradative effect**

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17 **Introduction**

18 The greatest use of bitumen is in paving and road building, particularly flexible
19 pavements. Examples of such areas of usage are in Highways Street and
20 driveways, airfields, Parking areas, service (petrol) stations and industrial floors
21 among several others [1]. Bitumen is thermoplastic as it consistency or degree of
22 hardness varies with temperature. On exposure to different radiation, bitumen
23 behaves in a different manner which will affect it quality. Bitumen is very sensitive
24 to any form of radiation which leads to a degradative effect on its constituent
25 thereby affect its usefulness for engineering purposes. [2]

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27 **Materials and Methods**

28 The bitumen used for the degradation experiments were collected from one of the
29 observatory wells in Agbabu, Ondo State, Nigeria. Agbabu is one of the major
30 towns located in the Nigerian natural bitumen belt and the place where bitumen
31 was first discovered in Nigeria (Adegoke, 2000) [3]. The raw natural bitumen

32 obtained from Agbabu was purified as described by the method employed by
33 Olabemiwo *et al* (2008) .The Chemicals used for this research are products of
34 BDH Limited which includes iso – octane (2,2,4-trimethylpentane) [4]

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36 **Irradiation of Bitumen Samples**

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38 Dry Petri dish (Pyrex) was weighed and 10g of purified natural bitumen was put on
39 it. Thin layer of the natural bitumen was formed on the petri dish. The petri dish
40 containing the purified natural bitumen was subjected to different
41 radiations(wavelength) for a period of Seven hours at interval of One, Three and
42 Seven hours respectively. Some of the irradiated sample was withdrawn into petri
43 dish at interval of One, Three and Seven hours to be analyzed. From the withdrawn
44 irradiated sample, 0.6g of it was carefully and accurately weighed into a beaker
45 and 20cm³ of iso – octane was added to precipitate out the Asphaltene component.
46 Filtration process of the solution was now carried out by making use of filter paper.
47 From the filtration process, two components were obtained which was residue and
48 filtrate. The residue is asphaltene and filtrate is maltene. The Maltene was
49 collected into a sample bottle while the asphaltene was washed about five times
50 with 20ml iso – octane. By the method employed by Olabemiwo et al (2008) using
51 Column Chromatography, maltene fraction which is the filtrate was separated into
52 saturated hydrocarbons, polycyclic aromatic hydrocarbon and polar compounds.

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54 **Analysis of Gas Chromatographic**

55 The gas chromatographic analysis used was 5890 series (Hewlet Packard) that is
56 equipped with flame ionization detector. The stationary phase used for the analysis
57 is a fused – silica capillary column coated with 0.25m film of HP-5. For
58 hydrocarbons that are saturated, about 3μL of sample was injected. The column
59 temperature started at 60⁰C, held for 2minutes isothermally and then increased to
60 200⁰C at the heating rate of 10⁰C/min for 20minutes. It was held at this
61 temperature for 2munutes and then increased to 320⁰C at the heating rate of 12⁰C
62 for 5minutes. The carrier gas used was nitrogen at a pressure of 30 psi. At pressure
63 of 22 and 28 psi, Hydrogen and air were introduced respectively. 300⁰C and 320⁰C
64 were used for Injector and detector temperature respectively.

65 The column temperature was held for 2minutes at about 70⁰C column temperature
66 for the aromatic hydrocarbons and later increase to 250⁰C at heating rate of 15⁰C
67 for 20minutes.It was held at 260⁰C for 6 minutes isothermally and then increased
68 to 320⁰C for 6mnutes at heating rate of 15⁰C and it was at this temperature for
69 10minutes. Nitrogen, which is the carrier gas for this experiment was used at a
70 pressure of 35psi. At a pressure of 25 and 30 psi, hydrogen and air was introduced

71 respectively. At temperature of 300 and 320⁰C was when injector and temperature
 72 was used respectively and the sample of injected volume is 2 μ L. By making use of
 73 the standards supplied by the Gas Chromatography equipment manufacturer,
 74 Calibration curves for the standard and aromatic hydrocarbon were prepared.

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Result and discussion

78 Gas Chromatography result of the saturated and aromatic fraction of irradiated
 79 bitumen with infrared.

80 **Table 1: Aliphatic Hydrocarbon Profile of Bitumen Irradiated with Infra – red Radiation**

PAH	AMOUNT (g/mg)			
	RAW BT	IRO 1 SAT	IRO 3 SAT	IRO 7 SAT
C ₁₁	304.33	371.745	227.238	194.829
C ₁₂	3.7772	2.5146	1.5918	2.3541
C ₁₃	37.3414	25.7805	56.4245	24.6311
C ₁₄	4.4330	2.99145	1.7349	3.1931
C ₁₅	45.1838	32.5694	19.69	27.9708
C ₁₆	4.1187	2.7674	1.6251	2.6861
C ₁₇	6.6893	5.0672	2.8083	4.3397
C ₁₈	4.0167	3.3583	1.9249	2.7063
C ₁₉	3.9041	2.9776	1.7255	2.5150
C ₂₀	8.9376	9.1302	5.0815	6.2955
C ₂₁	3.9808	6.6905	3.5853	4.0090
C ₂₂	11.4659	8.1386	6.0837	6.2387
C ₂₃	2.2465	2.2687	1.2912	1.6358
C ₂₄	115.834	40.9511	40.6723	50.120
C ₂₅	5.9590	6.3874	4.9352	5.5857
C ₂₆	39.1534	25.049	16.561	17.788
C ₂₇	10.5277	13.1155	8.0830	9.3792
C ₂₈	7.3891	17.6333	4.0959	4.9437
C ₂₉	1.0401	0.8327	0.7083	0.7928
C ₃₀	1.8928 $\times 10^{-5}$	-	-	9.16933 $\times 10^{-5}$
TOTAL PAH'S	620.4383	579.9684	405.8604	372.0145

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82 **Table 2: Polycyclic Aromatic Hydrocarbon Profile of bitumen irradiated with infra – red**

PAH	AMOUNT (g/mg)			
	RAW BT	IRO 1 SAT	IRO 3 SAT	IRO 7 SAT
Napthalene	0.7122	0.1160	0.12459	0.11831
Acenaphthylene	0.0000	0.0000	0.0000	0.0000
Acephathene	0.0083	0.01394	0.005290	0.00575
Fluorine	0.1980	0.08268	0.06521	0.026631
Phenathrene	0.12949	0.07757	0.062855	0.02038
Anthracene	0.02884	0.02988	0.010598	0.00679
Fluoranthene	0.02289	0.012464	0.006254	0.00255
Pyrene	0.01795	0.03190	0.01187	0.006787
Benzo(a) anthracene	-	0.01750	0.0085474	0.0031555
Chrysene	-	0.01833	0.07822	0.002763
Benzo (b) fluoranthene	0.02399	0.003922	-	$7.858 * 10^{-4}$
Benzo (k) fluoranthene	0.10900	-	-	0.0010659
Benzo (b) pyrene	0.0000	-	-	0.0000
Indeno (1,2,3 - cd)	0.08438	-	-	-
Dibenzo (a,h) anthracene	-	-	-	-
Benzo (g,h,i)	0.0246	-	-	-
TOTAL PAH'S	1.3596	0.4041	0.3734	0.1949

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Bitumen samples irradiated with Infra – red radiations

88 **Saturated Fractions:** The total amount in g/kg of the aliphatic hydrocarbons was
 89 found to decrease as the period of exposure of the bitumen to ultraviolet radiation
 90 increases. This can be contributed to cracking and recombination of product. The
 91 irradiation of bitumen brought about the cracking of some higher molecular mass
 92 hydrocarbons to lower molecular mass radicals.[5]

93 **Aromatic Fractions:** The total amount in g/kg of the polycyclic aromatic
94 hydrocarbons was found to decrease as the period of exposure of the bitumen to
95 infra-red radiation increases. Benzo (a) anthracenes and chrysene which were
96 absent in the control sample appeared after irradiation with infra-red radiations.
97 Indeno (1, 2, 3-cd) pyrene and Benzo (g, h, i) perylene which were present in the
98 control sample disappeared after irradiation with infra-red radiations.
99 Acenaphthylene , Benzo (a) pyrene and Dibenzo (a, h) anthracene were absent in
100 the control and irradiated sample. Benzo (b) fluoranthene which was present after
101 one hour of irradiation disappeared after three hours of irradiation. Benzo(k)
102 fluoroanthene which was present in the control sample disappeared after three
103 hours of irradiation and later reappeared in minimal amounts of seven hours if
104 irradiation.

105 **Conclusion**

106 The Gas Chromatogram result for Aromatic fraction of bitumen irradiated with
107 infrared radiation respectively shows that the total number of polycyclic
108 hydrocarbons decreases as the time of irradiation is increases. This reduces the
109 quality of bitumen used for construction purposes. Therefore, bitumen to be used
110 for construction or any other purposes should not be exposed for radiation. Hence,
111 Gas chromatography analysis shown that bitumen consists of Sixteen Polycyclic
112 hydrocarbon and these are responsible for the quality of bitumen.

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Appendix

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Definition of abbreviation

129 PAH: Polycyclic Aromatic Hydrocarbon

130 RAW BT: Raw Bitumen

131 IRO SAT 1: Saturated fraction of bitumen irradiated with infrared for one
132 hour133 IRO SAT 3: Saturated fraction of bitumen irradiated with infrared for three
134 hours135 IRO SAT 7: Saturated fraction of bitumen irradiated with infrared for Seven
136 hours

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