

1 Source Identification of Polycyclic Aromatic Hydrocarbons in Water at Point of Effluent 2 discharge point into the New Calabar River, Port Harcourt, Rivers State, Nigeria

3 4 Abstract

5 Water samples from effluents discharge points were analyzed for polycyclic aromatic
6 hydrocarbons (PAHs) concentration using chromatographic techniques. The result
7 obtained showed the presence of all the PAHs categories in the water samples. The total
8 concentrations of PAHs in the different location showed that Minipiti station was the
9 highest (44.99 mg/L), which was followed by the values obtained at the Police Post
10 station (36.89 mg/L) and finally, the lowest value was obtained at the Iwofe Jetty station
11 (11.923 mg/L). Diagnostic analysis of the concentrations of the different PAHs
12 concentrations based on the ratio of low molecular weight PAHs to high molecular
13 weight PAHs (LMW/HMW), anthracene / anthracene + phenanthrene {An/(An + Phe)}
14 fluoranthene/ fluoranthene + pyrene {Fl / (Fl + Pyr)} and benzo[a] anthracene/
15 benzo[a]anthracene + chrysene {BaA/ (BaA + Chr)} gave the predominance of pyrogenic
16 PAHs over petrogenic PAHs. Thus indicating more of human input sources than natural.
17 Ring size analysis indicated the predominance of the higher molecular weights (4-6) rings
18 over the lower molecular weights (2-3) rings. The implications of the high level of PAHs
19 within the sampled environment should give warning signals to the user of the water
20 environment'

21 **Keywords:** PAHs, diagnosis, effluent discharge, environment, source identification.

22 Introduction

23 Increase in Industrial development and expansion have led to radical socio-economic
24 advancement. This result associated with this growth is the issue or problem of
25 environmental disintegration and pollution. This has led to prevalent pollution problems from
26 discharged chemicals or derived products arising from the production processes. Some of the
27 pollutants or contaminants of the environment whose increase in any of the atmospheric
28 media are linked to human influence are polycyclic aromatic compounds (Flowers *et al.*,
29 2002). Polycyclic aromatic hydrocarbons are a large class of persistent organic compounds,
30 which are equally toxic to the environment. They are produced as derivatives of nearly every
31 form of incineration of carbon containing matter of biological organisms (Hien *et al.*, 2007).

32 The contribution of the petroleum industry to the world energy and economic development
33 over the past five to six decades is well known. One of the countries that have immensely
34 benefited from this industry is Nigeria (Ite *et al.*, 2016). The accidental releases of petroleum
35 products and other chemicals associated with exploration and exploitation activities is also a
36 common occurrence in the environmental (Ite *et al.*, 2013).

37 Particularly, the release of petroleum hydrocarbons into the atmosphere without regard to
38 laid down rules causes atmospheric pollution which has negative effect on climate change
39 (Ana *et al.*, 2010; Gorleku *et al.*, 2014), pollution of aquatic ecosystem with deleterious
40 effects, changes on aquatic ecology with subsequent antagonistic influences on biota,

41 destructive influence on tourism, fishing and recreational activities (Ite *et al.*, 2016; Yakubu,
42 2017).

43 Polynuclear aromatic hydrocarbons originated from partial burning of carbon-based materials
44 such as crude oil, timber, fire wood, diesel, fat, compost, forest fire, etc, (Srogi, 2007). Man-
45 made sources of PAHs arise from the burning of fossil fuels in the home environment,
46 discharges from automobiles, discharges from industries and gas flaring, and in the treatment
47 of both solid and liquid wastes (Ana *et al.*, 2010).

48 This study was carried out to investigate the concentrations of polycyclic aromatic
49 hydrocarbons in water and their sources at drainage discharge points into the New Calabar
50 River

51 **Materials and Methods**

52 Water samples were collected from three different locations (effluents discharge points) along
53 the New Calabar River. The samples were collected at a depth of 30 cm below surface water
54 with previously washed and dried glass bottles. They were transferred into ice packs and
55 immediately transported to the laboratory. The sampled points were: Police post, Minipiti and
56 Iwofe jetty discharge points.

57 The analysis of the water samples for the different components of PAHs was first done by
58 extraction with a mixture of hexane and dichloromethane. The extracts were further concentrated
59 and purified according to the method describe by Nwineewii and Marcus (2015) and the extract
60 analyzed with chromatographic system (HP 6890 Series GC system equipped with FID). The
61 FID was operated at a temperature 325°C.

62 The results obtained were then analyzed according to the method described by Ilechukwu *et al.*,
63 (2016) to determine, predict and apportion the source of the PAHs into the environment.

64 **Results and Discussion**

65 The concentrations of the different PAHs compounds from the various stations is given in Table
66 1. Table 1: Concentration of polycyclic aromatic hydrocarbons at the different station. The result
67 showed that naphthalene, acanaphthylene and phenanthrene were undetected at the Iwofe Jetty.
68 All the PAHs were detected at the Minipiti station, while at the Police Post station, naphthalene,
69 acanaphthene and Benzo (k) fluoranthene were not detected.

70 The elevated concentrations of PAHs in the present work possibly originated from fall out from
71 flared gases and waste water from petroleum based industries close to the sample points.
72 Recalling that Rumuolumeni axis of Port Harcourt is replete with plethora of onshore and
73 offshore petroleum activities, deport, illegal oil bunkering and destruction of stolen crude
74 through direct burning. This observation corroborates the findings of Esumang *et al.*, (2009),
75 that observed that the entry route of PAHs to the surface water of Accra, Ghana metropolis came
76 from atmospheric fallouts which were either precipitated by gravity or rain into the water, urban-
77 run offs, public and private waste discharges and industrial wastewater.

78 The global attention given to PAHs is due to their mutagenicity, carcinogenicity, teratogenicity
 79 and toxicity to humans (Stabenaut et al., 2006, Nwineewii and Marcus, 2015). The presence of
 80 these PAHs in the environment therefore portends danger to the users of the water. It has been
 81 established that the higher molecular weight PAHs constitute more danger to the environment,
 82 since they are potential carcinogens (Chrysikou et al., 2008).

83 Exposure to this water which contains all the categories of PAHs at appreciable level can be
 84 injurious to pregnant women and infants. It can inflict behavioural disorder, impair intelligent
 85 quotient, cause asthmatic breathing, and other health abnormalities in both adults and children
 86 (Nwinewii and Ibok, 2014).

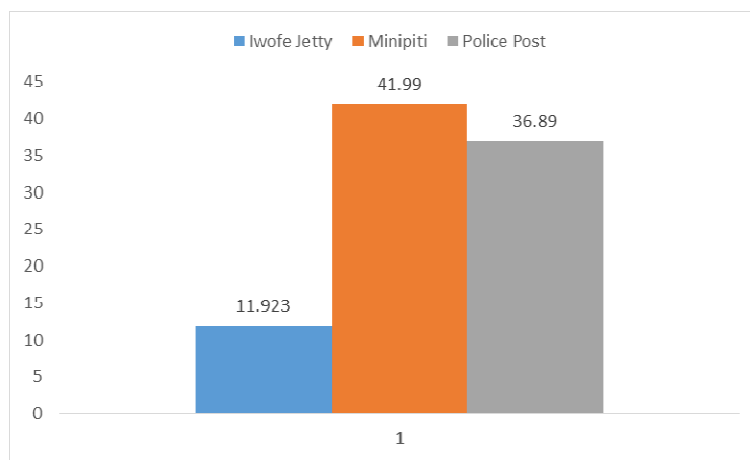
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| PAHs (mg/L) | Stations | | |
|---------------------------|-------------|----------|-------------|
| | Iwofe Jetty | Minipiti | Police Post |
| Napthalene | - | 4.25297 | - |
| Acenaphthylene | - | 3.34848 | 0.914637 |
| Acenaphthene | 0.09024 | 0.63742 | - |
| Flourene | 0.1743 | 0.94282 | 2.85662 |
| Phenanthrene | - | 0.90227 | 0.94504 |
| Anthracene | 0.46641 | 0.75535 | 1.13056 |
| Fluoranthene | 0.13007 | 0.66517 | 0.68168 |
| Pyrene | 0.38137 | 0.05029 | 11.01870 |
| Benzo (a) anthracene | 0.567288 | 7.69750 | 11.13948 |
| Chrysene | 0.62411 | 0.60233 | 3.97761 |
| Benzo (b) fluoranthene | 1.81437 | 0.22058 | 0.229111 |
| Benzo (k) fluoranthene | 0.08823 | 6.12916 | - |
| Benzo (a) pyrene | 1.58213 | 3.92882 | 0.57 |
| Dibenzo (a,h) anthracene | 1.42191 | 6.2427 | 0.5304 |
| Benzo (g,h,i) perylene | 1.06531 | 5.09045 | 2.05952 |
| Indeno (1,2,3-c,d) pyrene | 1.09051 | 0.5246 | 0.8291 |

88

89 The total concentrations of PAHs from the different sample station showed that Minipiti had the
 90 highest concentration of PAHs, which was followed by the values obtained at the Police post
 91 station and then the Iwofe Jetty station (Figure 1). The values of PAHs observed in the different
 92 stations were higher than the acceptable limit of 10 mg/L stipulated by FMENV, (1992) for
 93 drinking water. One would ordinarily expect that the Iwofe Jetty station would have had more
 94 concentrations considering the major oil based activities (legal or illegal) take place, but this was
 95 not so. The reason might be due to constant flow or movement of the tidal water. The high value
 96 recorded at the Minipiti station may probably be due to the proximity of an abattoir where
 97 different activities (roasting of cow skin, discharge of the gut content of animals, burning of the
 98 discarded bones of animals) take place. Secondly, the effluents from the university (Ignatius
 99 Ajuru University of Education) are also discharged through this point and thirdly, the stagnancy
 100 of the effluents discharge at this point for a longer time before the tidal water removes them
 101 during high tide. The same factors for the Minipiti station (except the university and the abattoir

102 may have contributed to the values of PAHs observed at the Police Post station. However, at the
103 Police Post station, fallouts from nearby filling stations can also be a contributory factor in PAHs
104 input.



105
106 Figure 1: Total polycyclic aromatic hydrocarbon content in the different stations

107 The result of the diagnostic ratios of the different PAHs combinations is given in Table 2. The
108 present study observed higher values of HMW PAHs as against the LMW PAHs. The ratio
109 between the LMW PAHs/HMW PAHs in the various stations revealed that the PAHs sources
110 were pyrogenic, this is based on the fact that a ratio of less than 1 was obtained in all the stations.
111 Diagnosis using $An / An + Phe$, gave values of 1, 0.456 and 0.545 at Iwofe jetty, Minipiti and
112 Police Post stations respectively. Values less than 0.1 indicated petrogenic or natural sources
113 while values greater than 0.1 showed pyrogenic or anthropogenic sources. From the values
114 obtained in the various stations, it is evident that the sources were pyrogenic. For the higher
115 molecular weight components of PAHs, (4 member rings) $Fl / Fl + Pyr$, values greater than 0.5
116 shows pyrogenic origin of PAHs, while values less than 0.4 shows petrogenic or natural origin.
117 The value obtained using this ratio indicated that PAHs in Iwofe Jetty and Police Post were of
118 petrogenic origin, while those of Iwofe originated from pyrogenic sources. For the 5-6 member
119 rings, the ratio between $BaA / (BaA + Chr)$ gave values of 0.717, 0.867 and 0.125 for Iwofe Jetty,
120 Minipiti and Police Post stations respectively. When the values from this ratio exceeds or
121 becomes less than 0.2, the origin of the PAHs are from petrogenic sources, while values above
122 0.35 shows pyrogenic sources of PAHs. The implication of the observation made with respect to
123 $BaA / (BaA + Chr)$ ratio implies that PAHs from Iwofe Jetty and Minipiti stations originated
124 from pyrogenic sources, while those of Police Post originated from petrogenic sources.

125 The diagnostic investigation or source identification results indicated diverse sources of PAHs
126 input into the water environment similar to observations of earlier studies in Niger delta
127 environment (Anyakora et al., 2011; Adeyemo and Ubiogoro, 2012; Moses *et al.*, 2015),
128 although pyrogenic sources of PAHs origin dominates over petrogenic sources in this present
129 work. This observation is in consonance with the finding of other authors in Niger Delta
130 environments (Inengite *et al.*, 2010; Adeyemo and Ubiogoro, 2012; Moses *et al.*, 2015). These
131 numerous input sources according to the authors are (oil exploration and exploitation activities,
132 agricultural inputs, forest and agricultural fire, abattoir activities, runoffs, flaring of gases, etc.

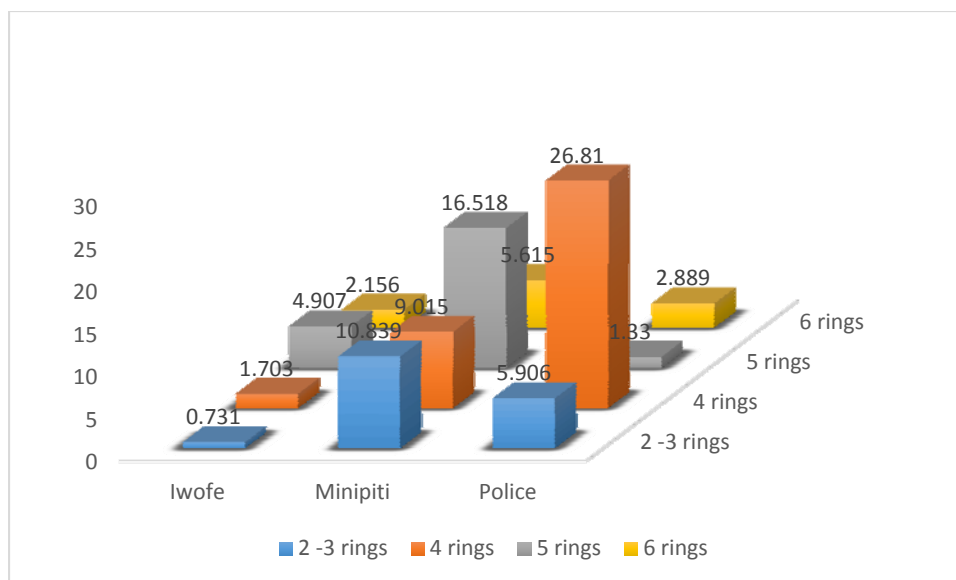
134 Table 2: Diagnostic ratios of PAHs in water from the different sample stations

| Stations | \sum LMW PAHs | \sum HMW PAHs | LMW/HMW | An/(An + Phe) | Fl/(Fl + Pyr) | BaA/(BaA + Chr) |
|-------------|-----------------|-----------------|---------|---------------|---------------|-----------------|
| Iwofe Jetty | 0.7309 | 7.700 | 0.095 | 1 | 0.254 | 0.717 |
| Minipiti | 10.839 | 30.040 | 0.361 | 0.456 | 0.930 | 0.867 |
| Police Post | 5.906 | 31.036 | 0.190 | 0.545 | 0.058 | 0.125 |

135

136 The concentration, predominance and occurrence of the different PAHs rings is shown in figure
 137 2. The result of the concentrations of the different PAHs ring sizes showed that the five member
 138 rings and the six member ring were predominant over the lower molecular weights rings in the
 139 Iwofe Jetty station. In the Minipiti station, though there was the predominance of the five
 140 membered rings over the others, yet they all occurred in appreciable concentrations, with the
 141 lowest members (2-3) having the nearest concentration to it. Sediment PAHs in the PolicPost
 142 station was dominated by the 4 ringed members, which was followed by the 2-3 ringed members.
 143 These observations in the predominant nature of one ring or the other over the rest is not in
 144 agreement with the values obtained in Some Creeks of South East Rivers State (Niger Delta)
 145 Nigeria (Nwineewii and Marcus, (2015) where no particular ring size dominated the others and
 146 those of Okoro, (2008) in in Ekpan Creeks, in Warri Delta state, where the values of the
 147 individual PAHs components were within the same concentration range. However, this
 148 observation corroborates the findings of Inengite et al., (2010) and Agbozu *et al.*, (2017) in Kolo
 149 Creek and different environments within Warri respectively.

150 The ring size possessed by PAHs tend to give the characteristics exhibited by the individual
 151 PAHs. The higher ring PAHs are less soluble in water, but the reverse becomes the case with
 152 their solubility in fats and oil. Therefore, the higher molecular weights accumulates more in fat
 153 tissues of animals and man. Due to the solubility of the lower molecular weights, they have the
 154 tendency to be easily absorbed by bio-organisms (Johnsen *et al.*, 2005; Agbozu *et al.*, 2017) and
 155 can easily be assessed or taken in by water consumers. The lower concentrations of the low
 156 molecular weights may not be unconnected with their volatility and uptake by plants and animal
 157 as against the non-availability of the higher molecular weight PAHs to plants and animals and
 158 their non- volatile nature and partial solubility (Haritash and Kaushik, 2009).



159
 160 Figure 2: Concentrations (mg/L) of the different PAHs ring categories in water samples from the
 161 different sample stations

162 **Conclusion**

163 The concentration of polycyclic aromatic hydrocarbons at the effluents discharge points were
 164 higher than required standards for water quality. This could pose negative consequences on the
 165 end user. The source determination of the PAHs implicated human activities, although the higher
 166 rings were more concentrated. However, knowing the adverse negative effects associated with
 167 PAHs, users of the sampled environments should to caution inorder to avoid future health
 168 challenges. Government should adequately monitor the content of discharged effluents and other
 169 pyrogenic activities within the area to cub further increase in concentrations of the PAHs
 170 component.

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