

Physicochemical Analysis of Moringa Oleifera Seeds

ABSTRACT

The soxhlet extraction of moringa seed oil was used to determine the proximate and physicochemical screening. The parameters obtained for the proximate screening were 7.64% moisture content, 4.05% ash content, 29.65% crude fat, 34.92% crude protein and 52.30% carbohydrate while the values obtained for the physico-chemical screening were 62.45% for Iodide, 1.1% for specific gravity, 9.84 for free fatty acid, 162.84% for saponification value, 4.10% for peroxide value, 1.46% for refractive index, 10.50% for viscosity and 5.95% for acid value. The results showed that *Moringa oleifera* seeds and seed oil could be employed for edible and commercial purposes.

Key words: Physico-chemical, *Moringa oleifera*, soxhlet extraction, medicinal.

INTRODUCTION

Moringa has long been cultivated and its parts consumed and used for a variety of purposes across the tropics (John, 2005). This is because of its impressive range of nutritional and medicinal value (Bina *et al.*, 2010). All part of the moringa tree (leaves, seeds, root and flowers) are not only suitable for human consumption but also for animal consumption (Mekonnen, 2016). The leaves which are rich in protein, mineral, B- carotene and antioxidant compound are used not only for human, animal nutrition but also in traditional medicine (Gandji, *et al.*, 2015). The seeds instead have herbals containing a significant amount of oil (up to 40%) with a high quality fatty acid composition (oleic acid > 70%) after refining a notable resistance to oxidative degradation (Alessandro *et al* 2015)

Moringa oleifera has received a great amount of attention as “natural nutrition of the tropics” (Nweze *et al.*, 2014). The leaves, fruits, and immature pods and flowers of this tree are locally used as vegetable (Anwar and Bahnger, 2003). *Moringa oleifera* seeds have antimicrobial activity and are utilized for waste water treatment (Anila., *et al.*, 2014). In Sudan, dry moringa *oleifera* seeds are used in place of alum by rural woman to treat highly turbid Nile water (Gideon *et al.*, 2010). The seeds of moringa *oleifera* are considered to antipyretic, acrid, and bitter (Francis *et al.*, 2009).

It has also been found that extract obtained from the leaves of moringa is 80% ethanol contains growth enhancing principles for higher plants (Alli Rani *et al.*, 2017). In ethno medicine, moringa *Olieifera* leaves have been used by local traditional healers in treatment of various

37 ailments such as gastric discomfort, stomach ulcers, diarrhea, dysentery and skin infection
38 (Suchada *et al.*, 2010)

39 Ghasi *et al.*, (2000) have reported that administration of crude leaf extract of moringa *Oleifera*
40 along with a high fat diet decreased the high fat diet induced increases in serum, liver and kidney
41 cholesterol level by 14.4, 6.4 and 11.1% respectively. The leaves have also been found to possess
42 antitumor, antipyretic, antiepileptic, antihypertensive and antioxidant properties (Upadhyay *et al.*
43 2010). In certain case of diabetes, moringa can also be used to stabilize sugar level (Mehta *et al.*,
44 2011)

45 The seed of moringa *Oleifera* has been a good antioxidant, able to reduce oxidative damage
46 associated with aging and cancer (Abdulkadir *et al.*, 2018).

48 MATERIALS AND METHODS

49 Sample Collection:

50 Moringa *oleifera* seeds were collected from the agricultural farm of ESUT Agbani, Enugu state.
51 They were authenticated by Prof. Olive Ngwu of Department of Agronomy and Crop Science,
52 Enugu State University of Science and Technology, Enugu.

54 Sample Preparation:

55 The collected fruits were opened to release seeds embedded inside the pods and were conveyed
56 using a black polythene bag to the laboratory. Moringa seeds were cleaned, sun dried for seven
57 days so as to minimize the moisture content. The dry seeds were grounded into powdery form
58 using a grinding mill and packaged in an air tight plastic container until ready for n-hexane
59 extraction (AOCS, 2001).

60 Extraction procedure

61 About 7 g of the sample were poured into soxhlet extraction apparatus fitted with a 1-L round
62 bottom flask and a condenser. The extraction was executed using 0.6 L of n-hexane at 70°C for
63 4-5 hours until a desired amount was achieved. After which the oil was obtained by evaporating
64 the solvent using a water bath at 60°C. The sample was weighed and the difference was
65 calculated as the weight of the sample before extraction – the weight of the sample after
66 extraction multiple by 100, divided by the initial weight of the sample to give the percentage
67 yield oil. The oil was stored in a cooled place for further analysis without further treatment.

70 Proximate Analysis

71 Proximate composition of seed samples were analyzed according to method described by AOAC
72 (2005). The proximate analyses carried out involves moisture content, ash content, crude fiber,
73 protein content, fats and oil, carbohydrate.

74 Physicochemical Analysis

75 Specific Gravity

76 An empty specific gravity bottle was weighed and recorded as w_1 . Then another specific gravity
77 bottle was filled with distilled H₂O and kept in a water bath at 50°C for 40 minutes, then weight

78 was taken and recorded as w_2 . After drying, the bottle was filled with the extracted oil and the
79 weight was recorded as w_3 . The process was repeated to get the final weight.

80

81 **Free Fatty Acid (FFA) value**

82 0.5 ml of sample was weighed into a 250 ml conical flask using a pipette. 20 ml of ethanol was
83 added into the conical flask containing the sample with constant stirring. Then three drops of
84 phenolphthalein indicator was added and titrated with 0.1N NaOH solution for 20s until it
85 changed faint pink with thorough shaking.

86 **Acid Value**

87 0.5 ml of oil was weighed using a pipette into a conical flask. Three drops of phenolphthalein
88 indicator and 20 ml of ethanol were added into the conical flask. The mixtures were titrated with
89 0.1N NaOH solution until a pink coloration was obtained.

90 **Saponification value**

91 3 g of the sample was weighed into 200 ml conical flask. 40 ml of alcoholic potassium hydroxide
92 was added into the container containing the sample with constant stirring. The resulting mixture
93 was refluxed for an hour thirty minutes until the entire oil dissolved. Two drops of indicator was
94 added and titrated against 0.5 N HCl with continuous shaking until the pink color changes to
95 colorless.

96 **Iodine Value**

97 0.5 g of sample was weighed into a conical flask and 15 ml of chloroform was added. 25 ml of
98 wiji's reagent was added and the mixture was stirred thoroughly using a glass rod. The flask was
99 covered tightly and placed in the dark corner for 1 hour. 40 ml of 15 % potassium iodide and 100
100 ml of distilled water was added and shaken vigorously. The mixtures were titrated against 0.1N
101 solution of sodium thiosulphate until the reddish solution almost disappears. Small amount of
102 starch indicator was added and titrated until the blue black colouration completely disappeared
103 after vigorous shaking.

104

105 **Peroxide Value**

106 0.5g of oil sample was weighed into a conical flask. 1g of potassium iodide and 20 ml of mixture
107 of DMSO and acetic acid was added into the conical flask containing the oil sample. It was
108 heated for 4 min. 15 ml of 3 % potassium iodide was added and was titrated with 0.02 sodium
109 thiosulphate until yellow color almost disappeared. 0.5 ml starch indicator was added and
110 shaken vigorously and was titrated carefully until blue color disappears.

111

112 **Viscosity**

113 The viscosity was determined using Brookfield viscometer (LVII, Brookfield Inc., USA) using
114 spindle no.5 with shear rate 100 rev/ min.

115 **Refractive Index**

116 The Refractive Index of oil was determined using an Abbe's refractometer. Two-three drops of
117 sample applied and reading was recorded.

118 **RESULTS AND DISCUSSION**

119 **Table 1 shows the result of proximate analysis of moringa oleifera oil**

120 **Proximate analysis result**

Parameters	Values in (%)
Moisture	7.64
Crude Protein	34.92
Oil	38.84
Crude fat	29.65
Ash	4.05
Carbohydrate	52.30

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122 **Table 2 shows the result of physico-chemical analysis of moringa oleifera oil**

123 **Physic-chemical analysis result**

Parameters	Values in (%)
Iodine value	62.45
Specific gravity	1.00
Free fatty acid (MgKOH/g)	9.84
Saponification value(MgKOH/g)	162.84
Peroxide value (MgEq/Kg)	4.10
Refractive index	1.46
Viscosity (MM ² /S)	10.50

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125 DISCUSSION

126 The 38.8% oil percentage yield agrees with 35-40% yield reported by solade (2008). The value
127 of the 62.84% for saponification and 62.45% for iodine value was in agreement with Orhevba *et*
128 *al.*, (2013). The crude fat value of 29.65% and 10.50% for viscosity is in contrary to what
129 Nzikou *et al.*, (2009) and Olaleye *et al* (2018) reported. The 4.05% for ash and 52.30% for
130 carbohydrate obtained agrees with 4.2% and 56.42% reported by Nzikou *et al* .,(2009) and
131 Orhevba *et al.*, (2013) respectively. 7.64% value for moisture content, 1.1% for specific gravity,
132 4.10% for peroxide value, 5.95% for acid value and 1.46% for refractive index was in agreement
133 with 7.51% for moisture, 0.896 for specific gravity, 5.00 for peroxide, 6.35 for acid value and
134 1.457 for refractive index as reported by Olaleye *et al.*, (2018). Orhevba *et al.*, (2013) reported
135 8.27% free fatty acid and Nzikou *et al.*, (2009) reported 37.6% for crude protein.

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138 CONCLUSION

139 The extracted oil from moringa oleifera seed could be utilized successfully as a source of edible
140 oil for human consumption based on its high saturated oil and also be used for other industrial
141 applications.

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