

DEMOCRATIC REPUBLICA OF CONGO MORINGA SEEDS OIL EXTRACTION, POTENTIAL MIGHTY ANTIPOISON

ABSTRACT

Background: The poisoning, one of the nowadays most serious problem of public health in Democratic Republic of Congo since two decades, has made many victims because of the lack of informations and because of the population impoverishment unable to accede to health care. The Democratic Republic of Congo has a very rich and diversified vegetable patrimony with known therapeutic properties needing only appropriate technology to deal with the extraction process of oils or active principles.

Comment [h1]: information

Aim and objective: The overall purpose pursued is to endue the country with home technology to solve somewhat the public health problem in DRC. The kinetic study of oil transfer from liquid- solid extraction has been undertaken in view of the phenomenon uptake in order to make possible home technology of reactors sizing, nowadays absent in underdeveloped countries.

Methodology: KUNYIMA method has been successfully extended to the Moringa seeds oil extraction in petroleum ether using Soxhlet device to assess its validity. The figures have been plotted by means of Origin 8 program.

Results: When $\log \frac{1}{m_{0e}-m_e}$ is plotted as a function of time, linear behavior has been obtained at constant temperature (56 °C) in dilute medium.

The global kinetic constant of this time dependent phenomenon has been calculated [$k = 1.2607 \pm 0.0591)h^{-1}$] to make possible the reactor building for the oil production. The comparison of extraction some parameters (m_e, k, \dots) between Gourd seeds oil and Moringa seeds oil measured and calculated in the same experimental conditions shows in petroleum ether a greater kinetic activity of solvent for Gourd seeds oil than for Moringa seeds oil followed by significant extraction of Gourd seeds oil as fast as the time advanced ($k_{GSO} > k_{MSO}$).

This observation suggests the existing difference of structures between the two species as it is hereby discussed. Moreover it has been pointed out previously that if the difference Δm in absolute value is not of the same errors magnitude order it would interpret the solvent effect. It should be noted however it has been observed the ratio $\frac{n_s}{n_e} > 1$ where the kinetic constant is high (Gourd seeds oil) and $\frac{n_s}{n_e} < 1$ where the kinetic constant is low (Moringa seeds oil) and this ratio might likewise better give informations on solvent effect. Prediction is done of getting possibility of sigmoid curve in the case of the presence of solvation different equilibriums. Also the sigmoid obtaining depends likely on both the structure of extractant solvent and the structure of extracted material. In that case the kinetic constant will be calculated in the upright region of sigmoid curve.

Comment [h2]: information

Conclusion: KUNYIMA method has been successfully used in the case of Moringa seeds oil extraction. KUNYIMA Method consists in best uptake of the phenomenon, in expressing it in suitable mathematical model in order to determine its velocity through its kinetic constant before sizing the experimentation reactor. The reactor volume depends on both the sizing factor and the desired volumic debit. According to its wonderful properties, Moringa (leaves, roots and seeds) might be a potential mighty antipoison particularly for heart and for entire body in general.

Comment [h3]: Check

Keywords: KUNYIMA method, Home technology, Moringa seeds, Sizing factor, Solvent effect, Sigmoid curve.

46 **ABBREVIATIONS:** Kinetic constant Gourd seeds oil (k_{GSO}), Kinetic constant
47 Moringa seeds oil (k_{MSO}), Reactor volume (V_r), Volumic debit (Q).

48

49 1. INTRODUCTION

50 *Moringa oleifera* has been extensively studied (Moringa leave and Moringa
51 seeds). There is abundant literature where it is reported its benefactions
52 [1,2,3,4]. Its high rates in vitamins A, B and C; its important contents in
53 minerals and chemical elements such as iron, zinc, calcium, copper, potassium,
54 magnesium, manganese, phosphorus, sulfur, selenium, sodium, molybdenum ...;
55 the total absence of cholesterol; its impressive amount in fibres and the presence
56 of essential aminoacids make it called “the life tree” used in various fields of
57 science such as medicine (phytotherapy), water science, stock breeding,
58 agriculture, cosmetics and perfume, nutrition, thin paint and so forth. Moringa
59 seeds contain 40% of super quality oils like olive oil with 73% of oleic acid
60 [5,6,7,8].

Comment [h4]: Amino acids

61 With respect to the medical side, Moringa intervenes in the prevention and
62 treatment of diabetes, hypertension, cardiovascular diseases, sleep, hairs and
63 skin problems [2,5]. Several bioactive compounds isolated from:

64 - seed like glucosinolates and isothiocyanates, hemagglutinins possess anti-
65 cancer, antibiotic, anti-inflammatory and agglutinogenic effects [9].

66 - moringa leaves such as glucosinolates, thio carbamates and carbamates as well
67 as other nitrile groups. These compounds are responsible for several beneficial
68 effects such as: hypotensive, hypolipidemic and antiatherosclerotic,
69 hypoglycemic, antifungal, regulation of the thyroid status [10,11,12,13,14].

Comment [h5]: responsible

70

71 The action of the roots is mainly antiseptic, anti-inflammatory, sedative,
72 cardiogenic, potentiator of some analgesic and antidepressive drugs. These
73 actions are mainly due to the presence of alkaloids such as Moringin,
74 Moringinine a powerful fungicide, bactericide Pterygospermine or Anthonin and
75 Spirochine
76 [15,16].

77 This medical aspect of Moringa seeds has interested our laboratory (LACOPA)
78 because indeed it has been already tried successfully in it to use Moringa as an
79 antipoison. Its effect has been highly appreciated. By adjunction Curcuma to
80 Moringa seeds the antipoison effect has been extraordinarily enhanced.

81 The tests are continuing and will wait for a fit required scientific sample to be
82 published. Anyway the preoccupation of Laboratory is to make possible the
83 heart normal acting beyond one century without any problem by improving and
84 stabilizing the cardiac exergetic yield [17,18].

85 **2. MATERIALS AND METHODS**

86 **a) Materials**

87 Drying oven, Balance (mark OHAUS), watch glass, desiccator, heating skull
88 cap, spade, beaker, cellulose cartridge (33 × 205 mm), thermometer, gel of
89 silicone, aluminium paper, thermostat, rotary evaporator, chronometer, mortar
90 and pestle, burettes, distillate water and petroleum ether have been used. Origin
91 8 program has served in this work. Moringa seeds with the following
92 characteristics have been used in this research [19,20,21]:

- 93
94 Reign: *Plantae*
95 Family: *Moringaceae*
96 Division: *Magnoliophyta*
97 Order: *Capparales*
98 Class: *Magnoliopsida*
99 Gender: *Moringa*
100 Species: *Oleifera*



101 **Fig.1. Moringa Tree**

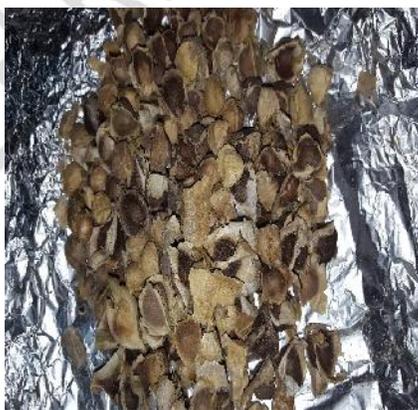


Fig.2. Moringa Seeds



Fig.3. Moringa Powder



Fig.4. Moringa Seeds Oil at 27 °C

102

103 b) Methods

104 • Oil extraction protocol

105 Moringa seeds have been husked, afterwards dried at 50 °C for four days in the
106 drying oven (mark memmert). These seeds were pounded and the obtained
107 powder was preserved in a desiccator. Ten grammes of this powder were
108 introduced in cellulose porous cartridge of 33 × 205 mm and put in Soxhlet
109 extractor.

110 In a three necked flask fitted with a thermometer, 450 mL of petroleum
111 ether (40 – 60 °C, $\rho_{\text{sol}} = 0.65 \text{ L}$) were used for the extraction.

112 The fitting out of Soxhlet was done on heating skull cap (mark thermos
113 scientific) in fixing the temperature at 56 °C. To maintain constant the
114 temperature during experiments, the heating skull cap was covered of
115 aluminium as a heat insulator. The ambient temperature has been kept at 24 – 25
116 °C [22,23,24,25].

117 After a given extraction time, the cartridge was dried in a drying oven at 50 °C
118 for 24 hours in order to get rid of traces of solvent. The solvent in the oil –
119 solvent mixture was recovered using a rotary evaporator at 60 °C and placed
120 finally in drying oven to get totally rid of solvent traces. After this, the balloon
121 flask with oil has been cooled in a desiccator and weighed. The difference
122 between the balloon flask containing oil and the empty one determines the
123 extracted oil mass at a t time in gramme.

124 • Equation (1) has been used in calculation [22].

125
$$\log\left(\frac{1}{m_0 - x}\right) = \log\left(\frac{1}{m_{0e} - m_e}\right) = \frac{k}{2,3} t + \log\frac{1}{m_0} \quad (1)$$

126 • For the sizing of extraction reactor, the following equation has been used
 127 [22,26]:

128
 129
$$V_r = \frac{\gamma}{k(1-\gamma)} Q = A Q \quad (2)$$

130 where A is the sizing factor γ the conversion degree [22].

131 **3. RESULTS AND DISCUSSION**

132 The extraction phenomenon depends on physical properties of extractant solvent
 133 such as its dielectric constant, its polarity, its polarizability, its refractive index
 134 and on the structure of the extracted material. Also the experimental conditions
 135 such as temperature, medium p^H , committed concentrations, sampling... etc.
 136 should be taken into account.

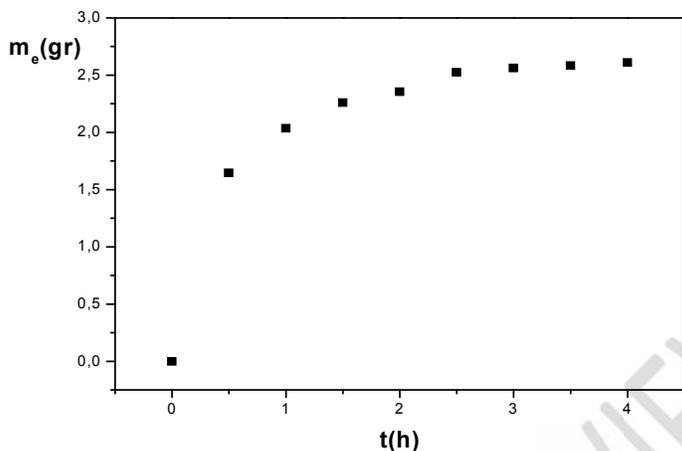
Comment [h6]: Be specific

137 In any extraction process two steps are to be considered: solubilisation and
 138 transfer (diffusion). The spontaneous phenomena of solubilisation occur when
 139 $\Delta G < 0$ (generally exothermic phenomena). There exists however the willing
 140 endothermic phenomena promoted usually by the increase of temperature.

141 **Table 1. Shows measured and calculated parameters of Moringa seeds oil extraction.**

Time (h)	$\frac{m_e}{m_{0e}}$	$\frac{m_e}{m_{0e}}$	$\frac{m_e}{m_{0e}}$	$\frac{m_e}{m_{0e}}$	$\log\frac{m_e}{m_{0e}}$	$\log\frac{m_e}{m_{0e}}$
0	0.0000±0.0000	2.6105±0.2901	2.6105	0.3831	-0.4167	-0.4167
0.5	1.6457±0.1767	2.6105±0.2901	0.9648	1.0365	0.0156	-0.4167
1	2.0361±0.2650	2.6105±0.2901	0.5744	1.7409	0.2408	-0.4167
1.5	2.2611±0.0960	2.6105±0.2901	0.3494	2.8620	0.4567	-0.4167
2	2.3550±0.1454	2.6105±0.2901	0.2555	3.9139	0.5926	-0.4167
2.5	2.5244±0.2889	2.6105±0.2901	0.0861	11.6144	1.0650	-0.4167
3	2.5635±0.3031	2.6105±0.2901	0.0470	21.2765	1.3279	-0.4167
3.5	2.5831±0.3080	2.6105±0.2901	0.0274	36.4964	1.5622	-0.4167
4	2.6105±0.2901	2.6105±0.2901	0.0000	-	-	-

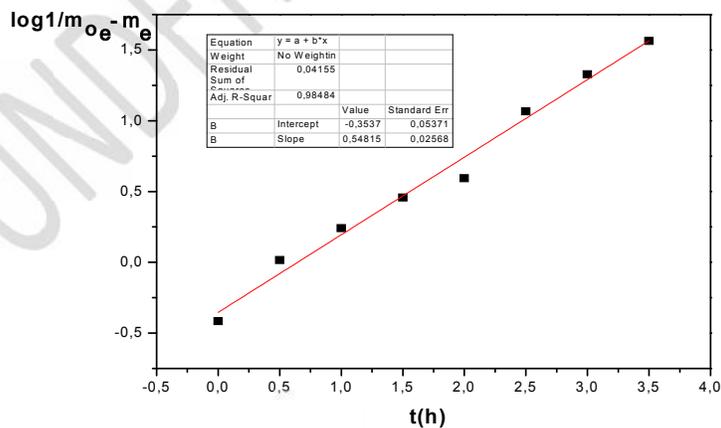
142
 143 In this table m_e is experimental extracted mass of Moringa seeds oil while m_{0e}
 144 is total experimental extractable mass of Moringa seeds oil for a given solvent
 145 hereby petroleum ether and for a given sample which can be obtained as shown
 146 in figure 5.



147
148
149

Fig.5. Experimental extracted mass as a function of time.

150 In this table each value is a mean value of three measurements. When
151 $\log \frac{1}{m_{oe}-m_e}$ is plotted versus time using origin 8 program a right line is obtained;
152 its slope gives the kinetic constant, a measure of kinetic activity of solvent, and
153 its intercept gives m_{os} called total statistical experimental extractible mass of
154 Moringa seeds oil in petroleum ether and for a given sample different from m_{oe}
155 as it is shown in figure 6.



156
157

Fig.6. $\log \frac{1}{m_{oe}-m_e}$ versus time

158 The kinetic constant has been found $k = (1.2607 \pm 0.05691)\text{h}^{-1}$. The intercept
 159 allows to calculate m_{o_s} as follows.

160
$$\log m_{o_s} = 0.3537 \pm 0.0537$$

161
$$\bar{m}_{o_s} = \bar{m}_{o_{th}} = 10^{(0.3537 \pm 0.0537)}$$

162
$$m_{o_{1s}} = m_{o_{1th}} = 10^{(0.3537 + 0.0537)} = 2.55541$$

163
 164
$$m_{o_{2s}} = m_{o_{2th}} = 10^{(0.3537 - 0.0537)} = 1.9954$$

165
$$\bar{m}_{o_s} = \bar{m}_{o_{th}} = \frac{m_{o_{1s}} + m_{o_{2s}}}{2} = 2.2754 \quad (3)$$

166
$$\bar{m}_{o_s} = 2.2754 \pm 0.2800$$

167 So the statistical mass (m_s) of Moringa seeds oil as a function of time has been
 168 calculated by means of the relation

169
$$m_s \approx m_{o_s} (1 - e^{-kt}) \quad (4)$$

170 All those values have been inscribed in table 2 and compared to Gourd seeds oil
 171 extraction values [22].

172 **Table 2. Comparison of the extraction parameters at 56 °C between Gourd seeds oil and Moringa seeds**
 173 **oil.**

Time (h)	Gourd Seeds Oil				k	Moringa Seeds Oil			
	\bar{m}_{o_s} (gr)	\bar{m}_{o_s} (gr)	\bar{m}_{o_s} (gr)	\bar{m}_{o_s} (gr)		\bar{m}_{o_s} (gr)	\bar{m}_{o_s} (gr)	\bar{m}_{o_s} (gr)	\bar{m}_{o_s} (gr)
0	0.0000±0.0000	0.0000±0.0000	0.0000	0.0000	1.6411±0.0712	0.0000±0.0000	0.0000±0.0000	0.0000	0.0000
0.5	—	—	—	—		1.6457±0.1767	1.0640±0.1567	0.5817	0.6465
1	4.0485±0.8790	4.8339±0.8432	0.7854	1.1940		2.0361±0.2650	1.6304±0.1464	0.4057	0.8007
1.5	4.6640±0.5862	5.4843±0.9174	0.8203	1.1760		2.2611±0.0960	1.9320±0.1388	0.3291	0.8544
2	5.1423±0.5888	5.7706±0.9399	0.6283	1.1222		2.3550±0.1454	2.0926±0.1334	0.2624	0.8886
2.5	5.2212±0.3131	5.8966±0.9453	0.6754	1.1294		2.5244±0.2889	2.1781±0.1296	0.3463	0.8628
3	5.2587±0.3562	5.9521±0.9458	0.6934	1.1318		2.5635±0.3031	2.2236±0.1644	0.3399	0.8674
3.5	5.2950±0.4075	5.9765±0.9451	0.6815	1.1287		2.5831±0.3080	2.2478±0.1256	0.3353	0.8702
4	5.3118±0.4193	5.9872±0.9444	0.6754	1.1272		2.6105±0.2901	2.2607±0.1246	0.3498	0.8660

174
 175 When this table 2 is analyzed, it can be firstly observed that the kinetic activity
 176 of petroleum ether is greater in Gourd seeds oil than in Moringa seeds oil
 177 ($k_{GSO} > k_{MSO}$). Secondly the extracted oil as a function of time in Gourd seeds
 178 is large amount compared to Moringa seeds oil. This observation suggests the
 179 existing difference of structures between Gourd seeds and Moringa seeds.
 180 Indeed, in a crystal the molecules are organized. In amorphous compound, there
 181 is no organization. In melting, a crystal is converted from an organized state to
 182 an unorganized one.

183 The are intermediate states between crystalline state and amorphous state called
184 mesomorph substances that can be divided into two classes according to the type
185 of organization namely smectic compounds where the molecules are oriented in
186 parallel and on parallel surfaces and nematic compounds where the molecules
187 are oriented in parallel without any order. Concerning some substances, there
188 are successive passages [27,28] :

189 Crystal \rightarrow Smectic state \rightarrow Nematic state \rightarrow Amorphous state

190 The organic materials with long chains are generally mesomorphs (Smectic and
191 Nematic structures) for example ammonium oleate.

192 So Gourd seeds can be identified to amorphous compound while Moringa seeds
193 to mesomorph material. Furthermore it has been previously [27,28] signalized
194 that if $\Delta m = m_s - m_e$ in absolute value is not of the same order of magnitude
195 of errors it would give informations on the solvent effect. It should be noted
196 however that the observation of this table 2 shows the ratio $\frac{n_s}{n_e} > 1$ for Gourd
197 seeds oil where the kinetic constant is high while $\frac{n_s}{n_e} < 1$ for Moringa seeds oil
198 where the kinetic constant is low. This ratio might be used likewise to diagnose
199 the solvent effect. Prediction can be raised on the possibility of obtaining
200 sigmoid curves in the case of the presence of solvation different equilibriums.

201 Also the sigmoid getting may depend likely on both the structure of extractant
202 solvent and the structure of the extracted material. The structure containing the
203 extracted oil and the committed concentrations should be taken into account.

204 In that case the kinetic constant will be calculated in the upright region of
205 sigmoid curve. Now it is possible to calculate the sizing factor and to size a
206 discontinuous stirrer vat reactor of oil production by the below-mentioned
207 equation [22,26,29] :

208 $V_r = \frac{\gamma}{k(1-\gamma)} Q$, as it is shown in table 3 and figure 7.

209 γ has been found $26.0827 \pm 1.1212 \%$

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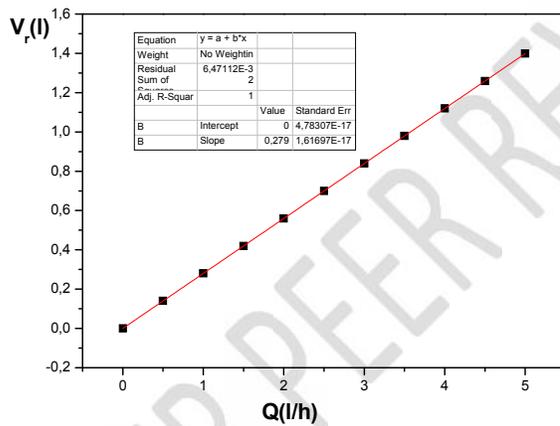
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215 **Table 3. Reactor volume (V_r) as a function of desired volumic debit (Q)**

216

Q (l/h)	V_r (l)
0	0.0000±0.0000
0.5	0.1399±0.0147
1	0.2798±0.0294
1.5	0.4197±0.0440
2	0.5596±0.0587
2.5	0.6995±0.0734
3	0.8394±0.0881
3.5	0.9793±0.1028
4	1.1192±0.1174
4.5	1.2591±0.1321
5	1.3990±0.1468

217



218

219

Fig.7. V_r versus Q

220

4. CONCLUSION

221 KUNYIMA method [22] has been successfully used in the case of Moringa
 222 seeds oil extraction. The laboratory is interested in extraction of medicinal plant
 223 oil aiming to better the heart acting.

224 According to its wonderful properties, Moringa (leaves, roots and seeds) might
 225 be a potential mighty antipoison particularly for heart and for entire body in
 226 general.

227 Moringa seeds can be identified to mesomorph structure while Gourd seeds are
 228 (made) of amorphous texture [28,30,31].

229 Gourd seeds oil, Moringa seeds oil, Curcuma, Ginger, Garlic, Arachis hypogaea
 230 oil and so forth are intensively studied in laboratory looking for home

231 technology [22,32,33].Their conjunctions in different proportions give
232 extraordinary healing properties.

233

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