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# Original research papers

## **Corn Cultivation for Silage: Evaluation of Elemental Composition in the soil and Plants by Neutron Activation Analysis**

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### **ABSTRACT**

Corn cultivation for silage requires special soil management and constant applications of fertilizers and agricultural pesticides to achieve satisfactory yield levels. This study was carried out on a farm that has grown corn for several years. The soil does not have adequate management in terms of fertility and fertilization. We collected samples of the soil, roots, leaves and grains in a corn silage area to investigate the chemical elements in the soil and plants. We used the neutron activation analysis (NAA) the  $k_0$ -standardization method. This technique consists in subjecting the sample to a neutron flux, in order to produce radioactive isotopes of the nuclei present in the original sample. In addition to the AAN technique, we apply the standardization method  $k_0$ , which consists of irradiating each sample together with a neutron flux monitor, in the same irradiation position. Several samples can be irradiated simultaneously when stacked inside the irradiation vessel, interspersed with flow monitors. The concentration of the elements is calculated in relation to the monitor which is usually gold (Au), eliminating the need for patterns. Irradiations were carried out in the TRIGA MARK I IPR-R1 research reactor at Nuclear Technology Development Centre/Brazilian Commission for Nuclear Energy (CDTN / CNEN). The NAA technique identified As, Ba, Br, Ca, Ce, Co, Cr, Cs, Cu, Eu, Fe, Hf, K, La, Mo, Na, Nd, Rb, Sb, Sc, Sm, Ta, Tb, Th, U, Yb and Zn, in the samples. Although the site studied lacks adequate management of soil fertility and fertilization, we identified Ca, Cu, K, Mo and Zn, essential for corn development. We detected adequate contents for the cultivation of silage corn by assessing these nutrients and their translocation in the plant.

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*Keywords: Zea mays L, Nutrients, Neutron activation analysis, Corn, Silage, Absorption, Translocation, Soil, Plants, Method  $k_0$ .*

### **1. INTRODUCTION**

~~Corn (*Zea mays L.*) is a major crop in Brazil and is the basis of human foods and animal feeds [1].~~

Corn (*Zea mays L.*) is a major crop in Brazil and is the basis of human foods and animal feeds [1]. Growing corn for silage requires advanced technologies, and it is low and irregular in Brazil. According to [2], production of corn forage is mainly affected by the absorption and translocation of nutrients throughout crop development, affecting the elemental composition of the product.

The Neutron Activation Analysis (NAA) is one analytical technique that analyzes chemical elements in samples of different matrices, such as soil, plants and foods. It is a multielemental isotopic technique, specific for quantitative and qualitative determination of

30 chemical elements, and is applied in chemical characterization of samples in the most varied  
31 | fields of applications [3,4]. The NAA is precise and accurate, and capable of determining  
32 elemental contents within a wide range, from traces to percentages [5]. It is a non-  
33 | destructive technique, because sample solubilization is not required, an advantage over  
34 other analytical techniques as they require sample solubilization for analysis. As the NAA  
35 does not require procedures, such as chemical separation or samples dissolution, it is a  
36 more versatile technique, reducing contamination risks and avoiding fractioning or partial  
37 recovery of elements [6].

38  
39 The NAA consists of subjecting a sample to a neutron flux to produce radioactive isotopes of  
40 the nuclei in the original sample, a reaction known as activation [7]. The resulting  
41 | radioisotopes have half-lives, ranging from fractions of seconds to several years, with  
42 radioactivity measured by gamma-ray spectrometry, with germanium semiconductor  
43 detectors. This type of detector is efficient at gamma radiation and has high resolution. Each  
44 radioisotope emits gamma rays with characteristic energies, allowing its identification. The  
45 | produced spectra are analyzed by software systems that locates, identifies and calculates  
46 the area under the gamma peak. The amount of events accumulated in a photopic event of  
47 the radioisotope of interest is used to obtain contents of elements in the sample [8].

48  
49 There are several ways of applying the NAA to obtain contents of chemical elements in the  
50 sample. The most commonly used is the comparative method, which analyzes standards for  
51 each element of interest to be determined. The  $k_0$  method is another analytical method that  
52 does not use standards of the elements of interest; rather, it monitors neutron fluxes. This  
53 method is based on the knowledge of spectral parameters in the irradiation position in the  
54 reactor, such as thermal and epithermal neutron fluxes, to perform gamma spectrometry in a  
55 highly calibrated gamma spectrometer and nuclear constants available in the literature. An  
56 | advantage of the  $k_0$  method is that its cost of analysis is low, and laboratory operations are  
57 reduced [4].

58  
59 In 2003, the standardized  $k_0$  method was established at the Nuclear Technology  
60 Development Centre, Brazilian Commission for Nuclear Energy (CDTN/CNEN), in Belo  
61 Horizonte, capital of city Minas Gerais State, Brazil. Since then, it has been used to  
62 determine the elemental composition of several matrices [9].

63  
64 In this work, we used the standardized  $k_0$  method of neutron activation [4,9] to analyze the  
65 elemental composition of the soil, roots, leaves and corn samples. The objective was to  
66 determine the chemical elements in samples in a corn growing area for silage and evaluate  
67 these nutrients in terms of the cultivated area and planted species.

## 68 69 **2. MATERIALS AND METHODS**

70  
71 We sampled the soil, roots, leaves and corn grains on a farm in the municipality of  
72 Biquinhas, Minas Gerais State (18°46'58" S and 45°30'08" W), at an average altitude of 629  
73 meters in the crop year of 2017.

74  
75 We performed systematic sampling, that is, we sampled soil/plant randomly on the first crop  
76 row and in zigzag. The collection was carried out in plots based on the harvest period  
77 (maturation) of the grains and the variety planted (Agrocere 5055).

### 78 79 **2.1 Soil sampling and preparation**

80  
81 Thirty samples consisting of 150 grams of soil were collected at 15 cm from the roots of the  
82 corn plant from 0 to 20 cm. The samples were conditioned in identified plastic bags and sent

83 to preparation at the laboratory at CDTN/CNEN, where the soil was dried at room  
84 temperature until reaching a constant weight. Then, the samples were milled in porcelain  
85 grains, sieved and packaged in sealed polyethylene bottles, forming a composite sample.  
86 Approximately 200 mg aliquots were weighed and placed into polyethylene vials suitable for  
87 irradiation.

## 88 89 **2.2 Collection and preparation of corn plant samples**

90  
91 The plants were cut close to the ground, and separated into roots, leaves and corn cobs.  
92 The samples were stored in plastic bags, tagged, and sent to the preparation at the  
93 laboratory at CDTN/CNEN.

94  
95 The soil aggregated to the roots was removed. The roots were washed in running water,  
96 deionized water and dried with paper towel. Then, the samples were packaged in plastic  
97 beakers and placed in the freezer for further freeze-drying. Regarding the leaves, whole  
98 parts in the opposite position and below the first spike were collected. The samples were  
99 stored in plastic bags and tagged. In the laboratory, the samples were washed in running  
100 and deionized water to remove any soil particles from the leaves. They were wrapped in  
101 plastic beakers and placed in the freezer. One corn ear was collected per plant. The samples  
102 were then placed in plastic bags, tagged and sent to the laboratory at CDTN/CNEN, where  
103 they were threshed, and the grains were washed in running and deionized water and then  
104 placed in the freezer.

105  
106 After reaming in the freezer for a minimum of 12 hours at  $-10^{\circ}\text{C}$ , the samples were freeze-  
107 dried and weighed to obtain the moisture percentage. Root and leaf samples were ground in  
108 a Grindomix GM 200 knife mill and packed in a bottle with a lid. The corn kernels were  
109 packed without crushing.

## 110 111 **2.3 Preparation of samples for analysis**

112  
113 For irradiation, aliquots of 150 mg of root and leaves were weighed and packed in a  
114 polyethylene sample holder. The corn samples were weighed in triplicate and stored in the  
115 sample holders with masses of 2.5 g. These sample masses were in agreement with the  
116 methodology of analysis of large samples, recently established at CDTN [10].

117  
118 For the application of the  $k_0$  method, the samples were conditioned in a larger sample vial,  
119 intercalated by neutron monitors, disks (6 mm diameter and 1 mm thick) [11] of Al-Au  
120 (0.1%), IRMM-530RA, supplied by the Institute for Reference Materials and Measurements  
121 (IRMM), Belgium.

## 122 123 **2.4 Irradiation**

124  
125 The irradiations were carried out for 8 hours in the nuclear research reactor TRIGA MARK-I  
126 IPR-R1 in at CDTN/CNEN. The samples were irradiated at the carousel in the irradiation  
127 channel IC-7, with the reactor operating at 100 kW. In this position, the spectral parameters  $f$   
128 (ratio of thermal and epidermal neutron fluxes) and  $\alpha$  (distance from the epithermal neutron  
129 profile) were 22.32 and - 0.0022, respectively, and the thermal neutron flux was  $6.35 \times 10^{11}$   
130  $\text{cm}^{-2} \text{s}^{-1}$  neutrons [9].

## 131 132 **2.5 Gamma Spectrometry**

133  
134 After irradiation, we waited the necessary time for the decay of radionuclides of shorter and  
135 interfering half-lives and, then, gamma spectrometry was performed on a HPGe coaxial

136 detector, with 50% of nominal efficiency, model GC 5019 CANBERRA, associated to  
 137 appropriate electronics and the Genie 2K spectra acquisition program, CANBERRA. The  
 138 counts were performed on the characteristic gamma energies of the radioisotopes produced.  
 139 For the analysis of the gamma spectra, we used the HyperLab program [8,12], which is  
 140 specific for this analysis, while for the calculation of the elemental content, we used the  
 141 Kayzero for Windows® [13] program.

## 142 143 2.6 Quality control

144  
 145 To verify the method performance, two certified reference materials were analyzed: Tomato  
 146 Leaves (SRM 1573a) and ~~a~~-sediment (BCR-320R) using the same method used for the corn  
 147 samples (Tables 1 and 2).

148  
 149 To evaluate the method efficiency, the  $E_n$  [14] test was applied, which takes into account for  
 150 the calculations, the expanded uncertainty of the experimental and certified values with a  
 151 coverage factor  $k = 2$ . This means that true results are 95% likely to be within the confidence  
 152 interval.

153  
 154 The following equations were used in the calculation of  $E_n$ :

$$155 \quad E_n = \frac{Valor_{exp} - Valor_{certified}}{\sqrt{U_{exp}^2 + U_{certified}^2}} \quad (1)$$

156  
 157  
 158 | where,  $\gamma_{exp}$  means experimental,  $U_{exp}$  means the expanded uncertainties with  $k = 1$ , of the  
 159 experimental results and  $U_{certified}$  is the certified values,  $k = 2$ .

$$160 \quad U_{exp} = 2 \cdot U_{exp\_Comb} \quad \text{where:} \quad U_{exp\_Comb} = \sqrt{u_{AREA}^2 + u_{method}^2} \quad (2)$$

161  
 162  
 163  $u_{AREA}$  is the net area uncertainty of the peak gamma studied and  $u_{method}$  is the total  
 164 uncertainty of method  $k_0$  established as 3.5% in the CDTN.

165  
 166 The method performance was evaluated by the criterion  $|E_n| \leq 1$ , meaning that the  
 167 performance was satisfactory, that is, the results of the method are within the 95%  
 168 | confidence interval. If  $|E_n| > 1$ , indicating the unsatisfactory performance of the method.

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 170  
 171 **Table 1. Experimental results and certified values for SRM 1573a, Tomato**  
 172 **Leaves, and statistical evaluation,  $E_n$ -score.**

| Elements | SRM 1573a                    |                         | $E_n$ -score |
|----------|------------------------------|-------------------------|--------------|
|          | Experimental Results,<br>k=1 | Certified Values<br>k=2 |              |
| Ca       | 47180 ± 1922                 | 50500 ± 900             | -0.84        |
| Co       | 0.55 ± 0.02                  | 0.57 ± 0.02             | -0.43        |
| Fe       | 378 ± 12                     | 368 ± 7                 | 0.41         |
| K        | 27190 ± 492                  | 27000 ± 500             | 0.17         |
| Na       | 133 ± 2                      | 136 ± 4                 | -0.49        |
| Rb       | 15 ± 1                       | 14.89 ± 0.27            | -0.22        |

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|----|---------------|---------------|------|
| Sb | 0.063 ± 0.003 | 0.063 ± 0.006 | 0.03 |
|----|---------------|---------------|------|

**Table 2. Experimental results and certified values for BCR-320R, Channel Sediment, and statistical evaluation,  $E_n$ -score.**

| Elements | BCR-320R, Channel Sediment   |                         |              |
|----------|------------------------------|-------------------------|--------------|
|          | Experimental Results,<br>k=1 | Certified Values<br>k=2 | $E_n$ -score |
|          | mg kg <sup>-1</sup>          | mg kg <sup>-1</sup>     |              |
| As       | 24 ± 1                       | 21.7 ± 2                | 0.94         |
| Co       | 11 ± 1                       | 9.7 ± 0.6               | 0.61         |
| Cr       | 67 ± 6                       | 59 ± 4                  | 0.58         |
| Fe       | 27825 ± 6833                 | 25700 ± 1300            | 0.28         |
| Hg       | 1.01 ± 0.07                  | 0.85 ± 0.09             | 0.95         |
| Sc       | 6 ± 1                        | 5.2 ± 0.4               | 0.21         |
| Th       | 6 ± 2                        | 5.3 ± 0.4               | 0.08         |
| U        | 1.6 ± 0.5                    | 1.56 ± 0.2              | 0.00         |
| Zn       | 352 ± 124                    | 319 ± 20                | 0.13         |

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### 3. RESULTS AND DISCUSSION

The chemical elements were analyzed based on their nuclear characteristics, that is, their respective radioisotopes, energies and half lives.

The results for the elements in the different matrices studied (soil, roots, leaves and corn grains) with their respective standard deviations are shown in Table 3. Six essential elements for plant growth, namely Cu, Fe, Mo and Zn as well as Ca and K were determined (Table 3). Co and Na, beneficial elements for the plants, were also analyzed.

#### 3.1 Elements in soil samples

Fe content in the soil samples was higher than in the other samples (root, leaves and corn grains). Fe is considered an important element for energy transformation in plants [15]; however, it is not usually available for absorption due to the low solubility in its oxidized form. In the plants, Fe is related to several metabolic activities, participating in the formation of some enzymes, besides it is indispensable in the processes of respiration, photosynthesis, N<sub>2</sub> fixation and electron transfer through the cycling between Fe<sup>2+</sup> and Fe<sup>3+</sup>. Thus, Fe availability is linked to crop yield. However, in Brazilian soils, responses of corn crops to Fe applications are practically non-existent. Fe contents in soils of the Cerrado region in Brazil is satisfactory for the development of the plants [16].

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**Table 3. Elements determined in the soil samples; root, leaf and corn grains (dry mass).**

| Elements | Content (mg kg <sup>-1</sup> ) |       |        |        |
|----------|--------------------------------|-------|--------|--------|
|          | Soil                           | Root  | Leaf   | Grains |
| Ag       | < 1                            | < 0.2 | < 0.08 | < 0.02 |

|    |              |                |               |               |
|----|--------------|----------------|---------------|---------------|
| As | 6.9 ± 0.3    | < 0.1          | < 0.08        | < 0.08        |
| Ba | 267 ± 12     | < 10           | < 4           | < 2           |
| Br | 5.9 ± 0.2    | 20 ± 1         | 7.0 ± 0.3     | 0.30 ± 0.06   |
| Ca | < 2*         | < 1.0*         | 7.0 ± 0.6*    | < 0.2*        |
| Ce | 81 ± 3       | 2.0 ± 0.1      | 0.36 ± 0.06   | < 0.1         |
| Co | 4.5 ± 0.2    | 0.30 ± 0.02    | < 0.008       | 0.010 ± 0.001 |
| Cr | 100 ± 4      | 3.8 ± 0.2      | 0.71 ± 0.07   | < 0.1         |
| Cs | 7.5 ± 0.3    | 0.2 ± 0.01     | 0.08 ± 0.01   | < 0.005       |
| Cu | < 3          | < 1            | < 0.7         | 388 ± 31      |
| Eu | 1.2 ± 0.1    | < 0.01         | < 0.004       | < 0.002       |
| Fe | 44 ± 2*      | 0.78 ± 0.03*   | 83 ± 5        | 13 ± 1        |
| Hf | 17 ± 1       | 0.20 ± 0.01    | < 0.008       | < 0.006       |
| Hg | < 0.7        | < 0.1          | < 0.08        | < 0.3         |
| K  | 16.9 ± 0.6*  | 24 ± 1*        | 21 ± 1*       | 3.0 ± 0.4*    |
| La | 21.9 ± 0.8   | 0.60 ± 0.02    | 0.60 ± 0.02   | < 0.001       |
| Mo | < 1          | < 0.8          | < 0.5         | 0.30 ± 0.01   |
| Na | 0.50 ± 0.02* | 0.105 ± 0.004* | 50 ± 2        | 6 ± 1         |
| Nd | 16.7 ± 2     | < 0.9          | < 0.8         | < 0.4         |
| Rb | 124.0 ± 5.0  | 27 ± 1         | 26 ± 1        | 4.4 ± 0.4     |
| Sb | 1.2 ± 0.05   | 0.07 ± 0.01    | < 0.02        | < 0.002       |
| Sc | 17.4 ± 0.6   | 0.31 ± 0.01    | 0.010 ± 0.001 | < 0.0003      |
| Se | < 2,0        | < 0.2          | < 0.2         | < 0.08        |
| Sm | 4,2 ± 0,2    | 0.089 ± 0.004  | 0.030 ± 0.001 | < 0.001       |
| Ta | 2.0 ± 0.1    | 0.04 ± 0.01    | < 0.01        | < 0.003       |
| Tb | 0.70 ± 0.03  | < 0.02         | < 0.01        | < 0.003       |
| Th | 17.9 ± 0.6   | 0.31 ± 0.01    | < 0.02        | < 0.01        |
| U  | 4.4 ± 0.2    | < 0.5          | < 0.3         | < 0.01        |
| Yb | 3.8 ± 0.2    | < 0.05         | < 0.04        | < 0.01        |
| Zn | 42 ± 4       | 8.5 ± 0.6      | 14 ± 1        | 22 ± 4        |

205 (g kg<sup>-1</sup>) \*; <, smaller than.

206

207

208 Zn content was 42 ± 4 mg kg<sup>-1</sup> in the soil sample. According to [17], its share in soil contents  
 209 ranges from 60 to 89 mg kg<sup>-1</sup> depending on the rock of origin and deposition sources. Zn  
 210 deficiency is augmented by prolonged cultivation, especially in sandy soils and in the  
 211 Cerrado region [18].

212

213 Cr was also identified in the soil at content 100 ± 4 mg kg<sup>-1</sup>. For [19] reported that lettuce  
 214 plants grown in a soil supplemented with 200 mg kg<sup>-1</sup> Cr<sup>3+</sup> showed a content of 11.1 mg kg<sup>-1</sup>  
 215 in the tissue and a 60% reduction in dry matter weight 60 days after application, in relation to  
 216 the control samples.

217

218 Br contents varied according to samples with the highest value found in the roots at 20 ± 1  
 219 mg kg<sup>-1</sup>. Most toxic elements are found in the roots, as the plants attempt to contain the  
 220 toxic effect they cause to metabolism. Br is not considered an essential chemical element for  
 221 plants and animals; however, its excessive consumption can be harmful to human health  
 222 [20]. In the soil, Br contents range from 5 to 40 mg kg<sup>-1</sup>, confirming the data obtained in this  
 223 study.

224

225 Ce, Eu, La, Nd, Sc, Sm, Tb, Yb are rare elements and were detected in the soil samples.  
 226 Ce, La, Sc and Nd had the highest contents, below the toxic limit for the plants, also  
 227 confirming the data obtained by [21].

228

### 229 **3.2 Elements found in roots samples**

230

231 Table 3 shows the elements found in the roots of corn plants. Fe, K and Na showed higher  
232 content, indicating greater absorption of these elements by the roots.

233

234 | Br, Co, Cr, Cs, Hf, Rb, Sb, Ta, Th, Zn were also found in the roots, and the number of  
235 elements absorbed by the roots is smaller than in the soil. This is because the plant may  
236 have blocked these toxic elements to avoid absorption. In addition, plants have root  
237 | exclusion mechanisms, and when none of these mechanisms is sufficient, they have  
238 physiological mechanisms to contain their toxic effect on metabolism [22]. There was Br  
239 accumulation in the root,  $20 \pm 1 \text{ mg kg}^{-1}$ , since Br content in the soil is smaller,  $5.9 \pm 0.2 \text{ mg}$   
240  $\text{kg}^{-1}$ .

241

242 Rare earth elements, namely Ce, La, Sc, and Sm, were also identified in the root samples, at  
243 contents much lower than in soil samples.

244

### 245 **3.3 Elements in leaf samples**

246

247 Fe, K and Na were detected in the leaf samples. Ca, which was not detected in the other  
248 samples, was found in the leaf sample at a content  $6.8 \pm 0.6 \text{ g kg}^{-1}$ . For [23] states that the  
249 suitable range for Ca, in nutritional terms, for corn plants is 2.5 to  $8.0 \text{ g kg}^{-1}$ .

250

251 K and Ca had the highest contents in the leaf samples and Ca accumulation in leaves is due  
252 to its low mobility in the phloem [24].

253

254 Br, Cr, Cs, Rb and Zn were also detected in the leaf samples; however, the elements were  
255 detected at lower contents than in the root samples. Except for Zn, which showed a larger  
256 translocation from roots to leaves.

257

258 The Cr changed its content of  $3.8 \pm 0.2 \text{ mg kg}^{-1}$  in the roots to  $0.71 \pm 0.07 \text{ mg kg}^{-1}$  in the  
259 leaves, confirming [25]. The authors explain that, in plants, most Cr is retained in the roots  
260 and only a small portion is transported to the shoots.

261

262 Rare elements (Ce, La, Sc, and Sm) continued to be translocated to the leaves at contents  
263 much lower than in the root samples, as these elements are absorbed by the plants, their  
264 distribution among different organs differs considerably. Research has shown that contents  
265 of rare elements in the roots are higher than in the other plant organs. For [26] showed  
266 decreases in contents of rare earth elements towards roots > leaves > stems > flowers > fruits  
267 or grains in various crops, such as corn, wheat and rice.

268

### 269 **3.4 Elements in samples of corn grains**

270

271 Fe, K and Na were also detected in samples of corn grains, while Ca was at its detection  
272 limit. K content was lower in the soil, root and leaf samples, contrary to data obtained by  
273 [15]. The authors reported that higher Ca contents increase grain production. Greater  
274 | transport and storage of photo assimilation in corn grains increase K content, since K  
275 participates in the transport of sucrose and photo assimilation from the source to the drain  
276 [15].

277

278 Na contents ranged from  $500 \text{ mg kg}^{-1}$  in the soil to  $6 \text{ mg kg}^{-1}$  in the grains. No reference  
279 value was found for this mineral in the literature; however, it is known that Na can partially  
280 replace K when Na contents are low in the soil. Still, Na is essential for some species,  
281 usually grasses, which photosynthesize via C-4 metabolism [27].

282  
283 Trace elements were also found in samples of corn grains, such as Co, Mo, Rb and Zn. Co  
284 was detected at a low content. For [28] reported that Co could be absorbed as  $\text{Co}^{2+}$ ,  
285 transported from the roots to shoots by the xylem, via the transpiratory current. The authors  
286 report that the highest contents are found in the roots, followed by the leaves, and the lowest  
287 contents are found in the stems. Mo also had a low content at  $0.30 \pm 0.01 \text{ mg kg}^{-1}$ . For [29],  
288 approximately  $0.08 \text{ mg kg}^{-1}$  of Mo in corn seeds is sufficient to allow normal growth and  
289 development of plants.

### 290 291 **3.5 General evaluation of the elements found**

292  
293 The high Zn contents found in the samples must be related to its greater absorption and  
294 storage in plant organs with later translocation to the grains. According to [30], the higher the  
295 Zn content in plant organs, the greater the translocation and accumulation in the grains.  
296 Foliar applications become a more effective strategy to increase Zn contents in the grain  
297 [31]; since Zn is absorbed by the leaf epidermis, remobilized and transferred to the grain  
298 through the phloem.

299  
300 Rare earth elements were not detected in the corn samples, as expected, due to their low  
301 solubility and bioavailability to reproductive organs. For [32], describes that not all elements  
302 are equally retained in the roots of different species, suggesting that tolerance to an element  
303 does not necessarily guarantee tolerance to another.

304  
305 Therefore, the most important elements, based on contents found in the soil, root, leaves  
306 and corn grains, were Fe, with high content in the soil, along with Br, K, Na, Rb and Zn,  
307 since all these elements were detected in all the samples studied. Several toxic elements  
308 and rare earth elements were also detected in soil samples, roots and leaves; nevertheless,  
309 all contents were below the threshold established by the Brazilian legislation. Essential  
310 elements detected were absorbed and moved to all plant parts, except for corn roots, which  
311 limited translocation of toxic elements (Co, Hf, Sb, Ta and Th) to the shoots.

## 312 313 **4. CONCLUSION**

314  
315 The neutron activation analysis, the  $k_0$ -method, identified and quantified many elements,  
316 namely As, Ba, Br, Ca, Ce, Co, Cr, Cs, Cu, Eu, Fe, Hf, K, La, Mo, Na, Nd, Rb, Sb, Sc, Sm,  
317 Ta, Tb, Th, U, Yb, and Zn in the samples, despite the different matrices.

318  
319 Although the studied site did not have ~~an~~ adequate management of soil fertilization, we  
320 identified six essential elements for crop development. The evaluation of contents of these  
321 nutrients and translocations in the plant showed the nutrients have adequate contents for the  
322 cultivation of corn for silage.

## 323 324 325 **COMPETING INTERESTS**

326  
327 Authors have declared that no competing interests exist.

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