

Assessment of the impact of untreated rubber effluent on the base cationic and mycological properties of rubber plantation soil in Calabar, Nigeria.

ABSTRACT

The study was carried out to determine the impact of rubber effluent on the cationic and mycological properties of soil in a rubber plantation through which it flows. Rubber effluent samples were collected for physicochemical and mycological analysis from the effluent discharge point of a rubber factory in Calabar, Nigeria. Three impact points (25 metres apart) were created along the flow channel of the effluent, and three sample points spaced 5m apart were created on both sides of each impact point. Top and subsoil samples were collected from the impact points and sample points for base cationic and mycological analysis. A control soil sample was also collected similarly. Correlation analysis, single-sample and two-sample were used to analyse the results. Results revealed that only temperature (26°C), sulphate (20.15mg/l) and chloride (43.87mg/l) conformed to Federal Environmental Protection Agency (FEPA) permissible limits of 40°C, 500mg/l and 600mg/l, respectively. Bacteria isolated from the rubber effluent were identified as *Pseudomonas* spp, *Micrococcus* spp, *Staphylococcus* spp, *Proteus* spp, *Klebsiella* spp, *Bacillus* spp, *Escherichia coli*, *Enterobacter* spp and *Aeromonas* spp. Fungi isolated were identified as *Aspergillus* spp, *Penicillium* spp, *Rhizopus* spp, *Mucor* spp and *Sporothrix* spp. Results also revealed that the rubber effluent impacted the soil, but parameters still recorded low values, as the effects of the effluent on the soil were altered by leaching, erosion and rubber root uptake. The study also revealed that microbiological investigation involving the use of a selective substrate can be used to augment or properly interpret results obtained from base cation studies similar to the current study, especially in a situation where pollution is not obvious or where factors like root uptake, leaching and erosion can potentially affect statistical results of base cation analysis.

Keywords: Base cations, Calabar soil, Mycological analysis, Nigeria, Rubber effluent, Rubber plantation

1. INTRODUCTION

Natural rubber is a common and easily available polyisoprenoid (biopolymers produced by living organisms). Although, over 1,500 species across 300 genera and eight families are known to produce latex-containing rubber particles, only a small number produce large quantities of rubber particles of high molecular mass [1]. Currently, natural rubber (*Hevea brasiliensis*) is the most important source of natural rubber.

Natural rubber is extensively used in the production of thousands of products in a variety of areas due to highly desirable qualities like impermeability, plasticity, flexibility, insulating and resistance properties [2]. Natural rubber is an important component of the automobile industry and it is used in the production of tyres, seats, bumpers, transmission belts, car mats, etc. Latex is used for the production of gloves, boots, baby feeding bottle teats, condoms, adhesives, balls, balloons, eraser etc [3]. Natural rubber is a highly valuable biopolymer of strategic importance which, unlike the majority of other biopolymers, cannot be completely substituted by synthetic materials in some applications.

29 Agro-based industries generate large amounts of effluent and natural rubber processing is a typical
30 example. Natural rubber processing requires large amounts of water and chemicals for its operation,
31 generating large quantities of effluent in the process. Effluent volume generated is related to the size
32 and capacity of the rubber plant. A factory that produces 20-30 metric tonnes of rubber generates an
33 average of 45,000 litres of effluent daily [4].

34 Rubber effluent, if not properly treated before disposal, can cause severe damage to man and the
35 environment. For instance, rubber effluents usually contain high levels of phosphate and ammonia
36 which makes it a suitable medium for algal growth; therefore, eutrophication of rivers and streams can
37 result if discharged without proper treatment [5]. The presence of suitable substrates and nutrients
38 (from natural latex) also makes it an ideal medium for a variety of microorganisms.

39 People living close to rubber-processing factories often complain about the foul-smelling odour from
40 the factories. Soil physicochemical and microbiological characteristics can become altered when
41 exposed to effluent. These alterations can cause toxicity problems and nutrient imbalance in the soil.
42 Pollution of the soil can also be hazardous to man and the environment when toxic chemicals move
43 through the food chain or percolate into groundwater used for drinking purposes [6]. Various
44 researchers have analysed rubber effluent in Nigeria [7, 5, 8]; however, there has been scanty
45 published research work on the peculiar physicochemical and mycological properties of this particular
46 rubber effluent, **and its impacts on surrounding soil**. Also, the ever-increasing global spotlight on the
47 environment requires that effluent properties and effluent impact be properly monitored.

48 **2. MATERIALS AND METHODS**

49 **2.1. Study area**

51 The rubber factory (N 5° 6' 80" and 8° 20' 24" E) is located on the outskirts of Calabar, which is the
52 capital of Cross River state, Nigeria. For soil samples, the study area (8° 20' 24.5" E and N 5° 6' 6.2",
53 geocoordinates for the second impact point) lies just outside the rubber factory. The factory used to
54 produce latex concentrate, however **currently** produces mainly crepe rubber. The factory has been
55 releasing untreated effluent indiscriminately into the environment for decades. Over time, a channel
56 (near the factory) of an average depth of about one metre developed, through which the wastewater
57 flows, with rainfall sometimes causing flooding of the surrounding soil. The soils sustaining the rubber
58 plantation are classified as Ultisols [9].

59 **2.2 Sample collection**

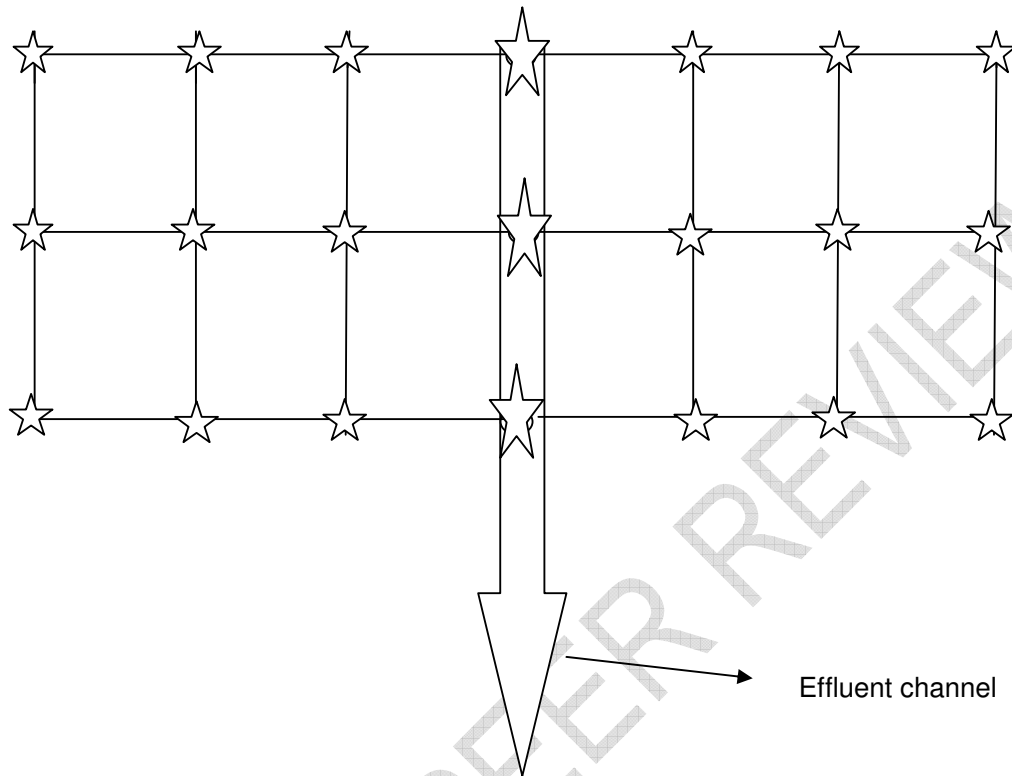
60 **2.2.1 Rubber effluent samples**

61 Rubber effluent samples were collected once per week consecutively (three times) at the discharge
62 point into sterile plastic bottles. Samples used for dissolved oxygen (DO) and biochemical oxygen
63 demand (BOD₅) analyses were collected in dark glass bottles. Parameters such as pH, conductivity,
64 and dissolved oxygen were analysed immediately. Samples were preserved (usually for 24 hours) at
65 4°C until required.

66 **2.2.2 Soil samples**

67 The experimental layout for soil sample collection around the factory is as shown in Figure 1. The
68 larger stars represent the impact points spaced 25 metres from each other and created along the
69 effluent flow channel. Other sample points (smaller stars) were created on both sides of each impact
70 point and spaced five (5) metres from each other. From each impact and sample point, two samples
71 representing topsoil (0-15cm) and subsoil (15-30 cm) were collected and stored in sterile bags. Soil
72 sampling was done using a cylindrical T-shaped probe. A circle of diameter (30 cm) was created at
73 each sampling point and from each a decontaminated probe was vertically-driven randomly into the
74 soil three (3) times for collection of samples for mycological analysis and randomly again 3 times for
75 base cation samples. Subsoil samples were collected by driving a decontaminated probe into the
76 holes created during collection of topsoil samples. A control (pristine) soil sample was collected from
77 the vertices of an equilateral triangle (length = 5m) created 100 metres away (measured diagonally
78 from the second impact point through the rightmost sample point of the first impact point).

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93 **Fig. 1. Experimental layout of study soil**
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96 **2.3 Physicochemical analysis**

97 **2.3.1 Rubber effluent samples**

98 Temperature was determined by dipping a mercury-in-glass thermometer into the sample immediately
99 after collection. pH, conductivity, dissolved oxygen and biochemical oxygen demand (BOD₅) were
100 measured using digital pH meter (HI9813; Hanna Instruments; Rhode Island, USA), conductivity
101 meter (HI9813, Hanna Instruments, Rhode Island, USA), dissolved oxygen meter (HI76408; Hanna
102 Instruments; Rhode Island, USA), dissolved oxygen meter (HI76408; Hanna Instruments; Rhode
103 Island, USA), respectively. Calcium and magnesium were determined by titrating with 0.1M EDTA
104 while potassium and sodium were determined by flame photometry [10]. Total suspended solids
105 (TSS) and total dissolved solids (TDS) was determined by gravimetry, chemical oxygen demand
106 (COD) by open reflux method, ammonia by phenate spectrophotometry, nitrate by colorimetric
107 method, phosphate by vanado-molybdate method, sulphate by turbidimetry and chloride by silver
108 nitrate titration method [10].

109 **2.3.2 Determination of exchangeable bases of soil samples**

110 Exchangeable cations (Ca, Mg, K, and Na) were extracted with 1N ammonium acetate (pH 7.0) [11].
111 Potassium and sodium were determined by flame photometry while calcium and magnesium were
112 determined by titrating with 0.1M EDTA [11].

113 **2.4 Mycological analysis**

114 **2.4.1 Rubber effluent**

115 For serial dilution, ten (10) millilitres of rubber effluent was added to 90 ml of distilled water for the first
116 ten-fold dilution. Subsequent ten-fold dilutions were carried out by adding one (1.0) millilitres of an
117 already diluted sample to nine (9.0) millilitres of distilled water.

118 *2.4.1.1 Enumeration of heterotrophic fungi*

119 Potato dextrose agar (Criterion C6621, USA) was prepared according to manufacturer's instructions
120 and supplemented with 100 µg/ml of chloramphenicol to inhibit bacterial growth. Zero-point-one (0.1)
121 ml of 10^{-3} to 10^{-5} dilutions were each spread-plated out in triplicates. The colony forming units
122 (CFU/ml) was determined after incubation at room temperature for 2-3 days.

123 *2.4.1.2 Enumeration of rubber effluent utilizing fungi*

124 Rubber effluent was added to mineral salts agar [12] at 2% (third rubber effluent sample analysed
125 was used) concentration and incorporated with 100 µg/ml of chloramphenicol as the anti-bacterial
126 agent. Zero point one (0.1) millilitres of 10^{-2} to 10^{-4} dilutions were each spread-plated out in triplicates.
127 The colony forming units (CFU/ml) was determined after incubation at room temperature for 4-5 days.

128 **2.4.2 Soil samples**

129 For serial dilution, 10 grams of soil was added to 90 ml of distilled water for the first ten-fold dilution.
130 Subsequent ten-fold dilutions were carried out by adding one (1.0) millilitres to nine (9.0) millilitres of
131 distilled water.

132 *2.4.2.1 Enumeration of heterotrophic fungi*

133 Potato dextrose agar (Criterion C6621, USA) was prepared according to manufacturer's instructions
134 and supplemented with 100 µg/ml of chloramphenicol to inhibit bacterial growth. Zero-point-one (0.1)
135 millilitres of 10^{-2} to 10^{-3} dilutions (topsoil) and 10^{-1} to 10^{-2} (subsoil) dilutions were each spread-plated
136 out in triplicates. The colony forming units (CFU/g) was determined after incubation at room
137 temperature for 2-3 days.

138 *2.4.2.2 Enumeration of rubber effluent utilising fungi*

139 Rubber effluent was added to mineral salts agar [12] at 2% (third rubber effluent sample analysed
140 was used) concentration and incorporated with 100 µg/ml of chloramphenicol as the antibacterial
141 agent. One (1) millilitres of 10^{-1} to 10^{-2} dilutions (topsoil) and 10^{-1} dilution (subsoil) were each spread-
142 plated out in triplicates. The colony forming units (CFU/g) was determined after incubation at room
143 temperature for 4-5 days.

144 **2.5. Isolation and preservation of pure culture**

145 Potato dextrose agar (Criterion C6621, USA) was used. Using a sterile inoculating loop, each
146 morphologically distinct colony from water and soil samples were sub-cultured twice and incubated at
147 64 hrs, before being transferred to agar slant for preservation. Inocula were obtained from the
148 respective tubes, sub-cultured on potato dextrose agar for 3 days for identification and
149 characterization purposes.

150 **2.6. Identification and characterization of fungal isolates**

151 Characterization of fungal isolates was based on macroscopic and microscopic appearances which
152 comprised pigmentation, colour of aerial and substrate hyphae, shape and kind of asexual spore,
153 presence of special structures, sporangiophore or conidiophores and characteristic of the spore head.
154 Isolates were determined using the scheme of Domsch et al. [13] and Barnett and Hunter [14].

155 2.7. Statistical analysis

156 Microsoft Excel 2013 (Microsoft Inc.) and R Statistical Software (R Software Foundation) were used
157 for a variety of statistical analyses which included Pearson's correlation, single-sample and two-
158 sample t-tests. The following includes definitions of terms and how statistical tests were employed.
159 Sample point: refers to any soil sample point collection, excluding impact points. Impact point: refers
160 to any soil sample collection point along the channel of effluent only. Sample line: refers to all sample
161 points on both sides of an impact point excluding the impact point. Correlation (Pearson's): carried out
162 between successive values of a parameter on both sides of an impact point and sampling distance
163 (excluding the particular impact point). One-sample t-test: was carried out between the value of a
164 parameter at a particular impact point and values of its sample line. One-sample t-test was also used
165 to compare control (pristine) soil and study soil parameters. Two-sample independent t-test was used
166 to compare topsoil and subsoil for each parameter.

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168 3. RESULTS AND DISCUSSION

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170 Physicochemical and mycological analysis of the effluent revealed that only temperature, sulphate
171 and chloride conformed to FEPA [15] standards for inland waters (Table 1). The fungi isolated from
172 the effluent were identified as *Aspergillus* spp (33%), *Penicillium* spp (24%), *Rhizopus* spp (20%),
173 *Mucor* spp (14%) and *Sporothrix* spp (9%).

174 The mean temperature (26°C) falls below the permissible limit (40 °C) set by FEPA [15]. Similarly,
175 Senthil et al. [16] obtained a mean value of 25.64°C. The mean pH value (5.8) indicates slight acidity.
176 This value falls outside the range of 6-9 set by FEPA [15]. pH values in the range of 5-8.1 have been
177 recorded by other authors [17,18,5,16,7]. Although effluent limit standard does not exist for
178 conductivity, an abrupt change in conductivity of a water body can be indicative of pollution [19]. This
179 study recorded a mean conductivity value of 4,457 µS/cm.

180 This study recorded an average value of 2,802 mg/l for TDS, which is higher than FEPA (1991) 2,000
181 mg/l. Non-isoprene constituents such as carbohydrates, sugar, proteins, lipids, carotenoids, inorganic
182 chemicals and a variety of chemicals used during processing make up the effluent from natural rubber
183 processing [20]. The high contents of many of these components likely contributed to the high TDS of
184 this rubber effluent. Similarly, Shruthi et al. [18], Girish [21] and Pillai and Girish [17] recorded mean
185 values of 2,240 mg/l, 2,397 mg/l and 2,240 mg/l, respectively from their studies. However, Iyagba et
186 al. [5] and Asia and Akporhonor [7] reported mean values of 550 mg/l and 450.0 mg/l, respectively.

187 The average value of 1,638 mg/l obtained for total suspended solids (TSS) is higher than the 30 mg/l
188 limit set by FEPA [15]. The high mean value recorded can be attributed to the heavy presence of latex
189 particles, microorganisms and inorganic matter in the effluent. Several authors have also recorded
190 high mean values for TSS [16,7,17,21].

191 The effluent has a low (anoxic) mean dissolved oxygen level (3.1 mg/l). Rubber effluents typically
192 have low DO levels, as revealed by Iyagba et al. [5], 0 mg/l; Asia and Akporhonor [7], 4.70 mg/l;
193 Senthil et al. [16], 1.16 mg/l. The mean BOD₅ value (3,038 mg/l) is higher than the 30 mg/l limit set by
194 FEPA [15]. High BOD values can be attributed to the presence of large amounts of latex particles,
195 proteins, sugars, and other organic matter. Similarly, high values ranging from 1,340-2,610 mg/l have
196 been reported by many researchers [17,7,18,21]. However, Senthil et al. [16] and Iyagba et al. [5]
197 reported low rather low BOD₅ values of 326 and 189 mg/l, respectively. The high mean COD value
198 (4,531 mg/l) indicates that the waste also contains substantial amounts of inert organic matter and
199 inorganics. This high COD result is consistent with the results of other authors [17,7,18,21].

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Table 1: Physicochemical and mycological properties of rubber effluent and FEPA standards

Parameters	First sample	Second sample	Third sample	Mean ± SEM	FEPA standards
Temperature (°C)	26	25	26	26±0.33	40
pH	5.6	5.8	6.1	5.8±0.14	6-9
Conductivity (µS/cm)	6,075	4,245	3,050	4,457±880	-
DO (mg/l)	1.7	3.4	4.2	3.1±0.737	-
BOD ₅ (mg/l)	4,504	2,900	1,710	3,038±810	30
COD (mg/l)	6,200	4,749	2,643	4,531±1,033	-
TSS (mg/l)	2,164	1,550	1,200	1,638±282	30
TDS (mg/l)	3,874	2,635	1,898	2,802±576	2000
Calcium (mg/l)	48.50	30.59	22.81	33.97±7.60	200
Magnesium (mg/l)	11.02	7.54	8.44	9.00±1.042	200
Potassium (mg/l)	34.76	29.33	16.42	26.84±5.44	-
Sodium (mg/l)	4.46	1.35	0.89	2.23±1.12	-
Phosphate (mg/l)	95.92	73.28	46.73	71.98±14.21	5
Nitrate (mg/l)	52.60	40.11	27.68	40.13±7.19	20
Ammonia (mg/l)	1.22	0.90	1.32	1.15±0.12	-
Sulphate (mg/l)	27.70	16.42	16.33	20.15±3.78	500
Chloride (mg/l)	59.4	39.5	32.7	43.87±8.0	600
HFC (CFU/ml)	5.40±2.08 x 10 ⁶	2.20±1.73 x 10 ⁵	1.30±1.15 x 10 ⁵	1.91±1.65 x 10 ⁶	-
RUFC (CFU/ml)	1.70±1.20 x 10 ⁵	4.70±2.18 x 10 ⁴	2.30±1.76 x 10 ⁴	8.00±1.71 x 10 ⁴	-

205 *KEY: DO = Dissolved oxygen, BOD = Biological oxygen demand, COD = Chemical oxygen demand, TSS = Total*
 206 *suspended solids, TDS = Total dissolved solids, HFC = Heterotrophic fungi count, RUFC = Rubber effluent*
 207 *utilising fungi, NTU = Nephelometric turbidity unit, µS/cm = microSiemens per centimeter, mg/l = Milligram per*
 208 *litre, CFU/ml = Colony-forming unit per millilitre, SEM = Standard error of the mean, FEPA = Federal*
 209 *Environmental Protection Agency*

210 Mean values of calcium (33.97 mg/l) and magnesium (9.00 mg/l) were within FEPA [15] limit of 200
 211 mg/l. An average ammonia value of 1.15 mg/l was recorded in this study. The relatively low ammonia
 212 value was likely due to the fact that ammonia was not used to preserve the field latex. Similarly, Asia
 213 and Akporhonor [7] obtained a low mean of 4.49 mg/l. High ammonia values ranging from 39.3-230
 214 mg/l have been obtained [5,21,18,16,17], pointing to the use of ammonia for preservation.

215 In this study, a mean nitrate value of 40.13 mg/l was obtained against a limit of 20 mg/l set by FEPA
 216 [15]. Iyagba et al. [5] obtained 0.07 mg/l and Asia and Akporhonor [7] recorded 1.36 mg/l. However,
 217 Senthil et al. [16] obtained a high value (149 mg/l). A mean phosphate value of 71.98 mg/l, which
 218 exceeds the 5 mg/l limit set by FEPA [15] was recorded. This result is consistent with high values (48-
 219 94.3 mg/l) recorded by other authors [16,18,21,5,17]. However, Asia and Akporhonor [7] reported a
 220 mean of 1.32mg/l. The mean sulphate value was 20.15 mg/l against 500 mg/l set by FEPA [15]. The
 221 mean chloride content was 43.87 mg/l against a limit of 600 mg/l set by FEPA [15]; however, Senthil
 222 et al. [16] recorded a mean chloride value of 1, 386 mg/l. Differences in the type and quantity of water
 223 and chemicals utilised, type of rubber processing or processing conditions are likely responsible for
 224 the big variations in physicochemical results obtained by different authors.

225 This study recorded a high mean TFC of 1.91 x 10⁶ CFU/ml. Iyagba et al. [5] also recorded a similarly
 226 high value of 3.8 x 10⁷ CFU/ml. The high fungal count of this study can be attributed to the nutrient-
 227 rich nature of rubber effluent which favoured the proliferation of fungi, the kind of water used in
 228 processing, or poor sanitary practices by the factory workers. Some of the fungi obtained in this study
 229 have been isolated in previous studies [21,16] and many are pathogenic. Rubber effluent utilizing
 230 fungi count (RUFC) indicates the presence of fungi that can utilize the rubber effluent.

231 Table 2 presents the overall, topsoil and subsoil means for impact points, sample points and control
 232 soil for the parameters. The overall means of exchangeable calcium, potassium and sodium,
 233 according to the classification of Landon [22], indicates low contents, except for magnesium. The low
 234 base contents can be attributed to erosion, leaching, clay fixation of these base cations. Also, rubber
 235 plantations can cause base cations values of soil to decline over time [23,24]. The moderate
 236 magnesium content of the study soil indicates that the soil is moderately rich in magnesium minerals
 237 like dolomite and serpentine. Rubber effluent utilizing fungi count (RUFC) indicates the presence of
 238 fungi that can utilize the rubber effluent. The RUFC was lower than HFC due to the probable toxicity
 239 of the effluent to some fungi or lack of suitable substrates or nutrients for others.

240 **Table 2: Means of exchangeable bases and mycological properties of study soil and control**
 241 **soil**

Parameters*	Impact points means			Sample points means			Control soil	
	Overall	Topsoil	Subsoil	Overall	Topsoil	Subsoil	Topsoil	Subsoil
Ex. Ca	3.90±0.09	3.93±0.18	3.87±0.07	3.97±0.07	3.92±0.07	4.03±0.12	3.8	3.6
Ex. Mg	1.83±0.15	1.9±0.29	1.73±0.13	1.5±0.05	1.57±0.07	1.50±0.07	1.4	1.3
Ex. K	0.11±0.00	0.11±0.01	0.11±0.01	0.11±0.00	0.11±0.00	0.11±0.00	0.11	0.11
Ex. Na	0.07±0.01	0.08±0.01	0.07±0.01	0.06±0.00	0.07±0.00	0.06±0.00	0.07	0.08
HFC	5.90±1.42 x 10 ³	8.53±1.25 x 10 ³	3.27±1.58 x 10 ³	1.33±2.61 x 10 ⁴	2.24±2.30 x 10 ⁴	4.19±2.92 x 10 ³	1.90±1.73 x 10 ⁴	4.9±2.03 x 10 ³
RUFC	2.73±1.76 x 10 ³	3.57±2.16 x 10 ³	1.90±1.36 x 10 ³	4.32±2.77 x 10 ³	3.10±3.28 x 10 ³	1.60±2.03 x 10 ³	2.70±1.45 x 10 ³	1.30±1.20 x 10 ³

242 *Mean±standard error of mean (SEM)
 243 *Units: Ex. Ca, Ex. Mg, Ex. K, Ex. Na = cmol/kg; HFC, RUFC = CFU/g
 244 KEY: Ex. Ca = Exchangeable calcium, Ex. Mg = Exchangeable magnesium, Ex. Mg = Exchangeable potassium,
 245 Ex. Mg = Exchangeable sodium, cmol/kg = centimoles/kg, HFC = Heterotrophic fungi count, RUFC = Rubber
 246 effluent utilising fungi
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248 Table 3 presents the results of correlation analysis relating sample lines (distance) to each of the
 249 parameters. There were significant negative correlations for sodium ($r = -0.97$, $P < 0.01$) at the third
 250 sample line of subsoil and for RUFC at first ($r = -0.83$, $P < 0.05$) and third ($r = -0.95$, $P < 0.01$) sample
 251 lines of topsoil; however, there were no significant correlations ($P > 0.05$) for calcium, magnesium,
 252 potassium and HFC. The significant negative correlation for sodium implies that other potentially
 253 significant correlations were cancelled out by erosion, leaching and rubber root uptake. No significant
 254 correlations were observed for HFC (topsoil and subsoil) since the media used was not selective. The
 255 significant negative correlations for RUFC highlights the receding effect of the effluent on the study
 256 soil. The sample points closer to impact channels were impacted more, leading to stimulation of
 257 metabolically capable fungi. The significant correlation for RUFC also indicates that other potentially
 258 significant correlations were cancelled out by leaching, erosion and rubber root uptake.

259 **Table 3: Coefficients of correlation (r) relating sample lines (distance) to each of the**
 260 **parameters**

Parameters	Topsoil			Subsoil		
	1st SL	2nd SL	3rd SL	1st SL	2nd SL	3rd SL
Ex. Ca	-0.50	-0.38	0	-0.74	0.45	0.23
Ex. Mg	-0.30	0.34	0.30	0.21	0.39	0
Ex. K	-0.23	-0.35	0	-0.22	-0.35	0
Ex. Na	0.65	-0.49	-0.76	0.35	-0.68	-0.97**
HFC	-0.15	0.61	-0.31	0.16	-0.48	-0.60
RUFC	-0.83*	0.20	-0.95**	0.25	-0.11	-0.71

261 *Correlation is significant at 0.05 alpha level (two-sided)
 262 **Correlation is significant at 0.01 alpha level (two-sided)
 263 KEY: Ex. Ca = Exchangeable calcium, Ex. Mg = Exchangeable magnesium, Ex. Mg = Exchangeable potassium,
 264 Ex. Mg = Exchangeable sodium, cmol/kg = centimoles/kg, HFC = Heterotrophic fungi count, RUFC = Rubber
 265 effluent utilising fungi count
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267 One-sample t-test results for study soil and control soil comparisons for the parameters are presented
 268 in Table 4. For topsoil, the test revealed significant results for exchangeable magnesium ($P < 0.05$)
 269 and RUFC ($P < 0.01$), while there were no significant results ($P > 0.05$) for exchangeable calcium,

270 potassium, sodium and HFC. For subsoil, the test revealed significant results ($P < 0.01$) for
 271 exchangeable calcium, magnesium, sodium and RUFC, while there were no significant results ($P >$
 272 0.05) for exchangeable potassium and HFC. The significant differences recorded between study soil
 273 and control (pristine) soil base cation parameters indicate the effect of the effluent on the study soil.
 274 Heterotrophic fungi count (HFC) of study soil was not significantly different from that of control
 275 (pristine) soil. This means that stimulation of rubber effluent utilising fungi did not lead to an increase
 276 in the total number of fungi in the study soil, even when RUFC increased. RUFC of study soil was
 277 significantly different from that of control (pristine) soil due to stimulation of metabolically capable fungi
 278 by the effluent in the study soil, leading to their increment. This stimulation was near-absent in pristine
 279 soil with little or no exposure to rubber effluent, causing smaller RUFC.

280 **Table 4: One-sample t-test comparing exchangeable bases and mycological parameters of**
 281 **study soil with control soil**

Parameters	Topsoil/Topsoil (<i>P</i> -values)	Subsoil/subsoil (<i>P</i> -values)
Exchangeable calcium	0.1212	0.001685**
Exchangeable magnesium	0.02781*	0.00896**
Exchangeable potassium	0.2151	0.6309
Exchangeable sodium	0.2307	0.0009409**
HFC	0.05644	0.09867
RUFC	0.0001241**	0.0002231**

282 *Significant at 0.05 alpha level (two-sided)
 283 **Significant at 0.01 alpha level (two-sided)
 284 KEY: HFC = Heterotrophic fungi count, RUFC = Rubber effluent utilising fungi count
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286 Table 5 shows the results of a two-sample t-test comparing topsoil and subsoil values for each
 287 parameter. The test showed significant results for HFC ($P < 0.01$) and RUFC ($P < 0.05$), but no
 288 significant results ($P > 0.05$) for the base cations. There was no significant difference between the top
 289 and subsoil for exchangeable cations probably due to rubber root uptake. HFC and RUFC decreased
 290 with depth in this study soil. This can be attributed to more vegetal cover, better soil structure and
 291 more organic matter in the topsoil [25]. The fungi isolated in the study soil were *Aspergillus* spp,
 292 *Penicillium* spp, *Rhizopus* spp, *Fusarium* spp, *Mucor* spp, *Cladosporium* spp, *Absidia* spp and
 293 *Chrysosporium* spp.

294 **Table 5: Two-sample independent t-test comparing topsoil and subsoil values of each**
 295 **parameter**

Parameters	<i>P</i> -values
Exchangeable calcium	0.4172
Exchangeable magnesium	0.4059
Exchangeable potassium	0.6993
Exchangeable sodium	0.5802
HFC	2.947×10^{-12} **
RUFC	0.01129*

296 *Significant at 0.05 level (two-sided)
 297 **Significant at 0.01 level (two-sided)
 298 KEY: HFC = Heterotrophic fungi count, RUFC = Rubber effluent utilising fungi count
 299

300 4. CONCLUSION

301
 302 The study revealed that the effluent should be treated before discharge into the environment since
 303 some parameters recorded values above permissible limits. The mycological investigations added
 304 more weight to the body of evidence in support of the impact of the wastewater on the study area
 305 since the stimulation of rubber utilising fungi in a receding manner from the flow channel evidently
 306 points to an impact decreasing with increasing distance from the flow channel of the wastewater.
 307 Hence, correlation analysis performed on data from microbiological investigation involving the use of a
 308 selective substrate can be used to augment or properly interpret results obtained from correlation
 309 analysis involving base cation parameters, especially in a situation where, like in this study, pollution
 310 is not obvious or where factors like root uptake, leaching and erosion can potentially cancel out

311 significant correlation results of base cation parameters. Also, the significantly different RUF of study
312 soil from that of control soil reflects the stimulation (and hence increment) of **metabolically-capable**
313 **fungi in the study soil due to continuous exposure to effluent, an exposure** that was absent in control
314 soil. Although the soil was impacted by the rubber wastewater, most base cation parameters still
315 recorded low values due to leaching, erosion and rubber root uptake.

316

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