Degradative Effect of I.R radiations on the Constituents of Bitumen

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Short Communication

Abstract

Sample of natural bitumen were taken from bitumen well in Agbabu town in Odigbo Local Government of Ondo State. These samples were separately irradiated with infrared radiations for a period of seven hours. Part of the sample was withdrawn at interval of One, Three and Seven hours. The withdrawn sample was later separated into maltene and asphaltene fractions. The maltene fraction was further separated into saturated, aromatic and polar fraction. The saturated and aromatic fractions were subjected to gas chromatography analysis. The Saturated and aromatic profiles of the bitumen were found to vary with the period of irradiation. Both the amount of saturated and aromatic compounds in the bitumen decreased with the period of irradiation. Thus, the radiations were found to have a degradative effect on the composition of bitumen.

Keywords: Bitumen, radiations, Gas Chromatography, degradative effect

Introduction

The greatest use of bitumen is in paving and road building, particularly flexible pavements. Examples of such areas of usage are in Highways Street and driveways, airfields, Parking areas, service (petrol) stations and industrial floors

among several others [1]. Bitumen is thermoplastic as it consistency or degree of hardness varies with temperature. On exposure to different radiation, bitumen behaves in a different manner which will affect it quality. Bitumen is very sensitive to any form of radiation which leads to a degradative effect on its constituent thereby affect its usefulness for engineering purposes[2]. However, because of external environmental effects such as oxygen [3] and ultraviolet (UV) radiation [4,5], bitumenious properties do not always satisfy operating requirements [6]. Virginie M. et al (2016) carried out a research on Study of UV rays effects on the evolution of bituminous mix behavior. The influence of UV radiation on the ageing of bituminous mixtures containing an elastomer modified bitumen cannot be totally ignored. the UV impact can be distinguished and found to be dominant for the production of carbonyl functions, the disappearance of C=C double bond of SBS and the increase of binder's hardening[7].

Nurfazilah Binti Mat Salleh(2012) work on effects of overheating bitumen on hot mix asphalt Properties. It was concluded that by increasing the bitumen heating temperatures before mixing procedure proven to oxidize and harden the bitumen at earlier stages. Meanwhile, maximum heating temperatures for 60/70 PEN bitumen before loss it HMA properties should be less or equal to 189°C. Asphaltic concrete with overheated bitumen which is not exceeded the maximum heating temperatures do increase the adhesion between the aggregate particles, durability and possibility to minimize the deformation of road wearing course[8].

Mello M.S *etal* (2011) investigated on the effects of gamma irradiation on bitumen. He Concluded that the penetration results of CAP 50/70 showed that doses below 5 Gy inhibited the influence of gamma radiation in the bitumen mechanical properties. The results of MR tests agree with Whiteoak, leading to conclusion that the gamma radiation causes aging in Bitumen and Asphaltic mixes (9).

The aim of this is to study the degradative effect of Infrared radiation on the composition of bitumen which is one of the major causes why the quality of bitumen is being compromise.

Materials and Methods

The bitumen used for the degradation experiments were collected from one of the observatory wells in Agbabu, Ondo State, Nigeria. Agbabu is one of the major towns located in the Nigerian natural bitumen belt and the place where bitumen was first discovered in Nigeria (Adegoke, 2000) [10]. The raw natural bitumen obtained from Agbabu was purified as described by the method employed by Olabemiwo *et al* (2008) .The Chemicals used for this research are products of BDH Limited which includes iso – octane (2,2,4-trimethylpentane) [11]

Irradiation of Bitumen Samples

Dry Petri dish (Pyrex) was weighed and 10g of purified natural bitumen was put on it. Thin layer of the natural bitumen was formed on the petri dish. The petri dish purified natural bitumen was subjected the radiations(wavelength) for a period of Seven hours at interval of One, Three and Seven hours respectively. Some of the irradiated sample was withdrawn into petri dish at interval of One, Three and Seven hours to be analyzed. From the withdrawn irradiated sample, 0.6g of it was carefully and accurately weighed into a beaker and 20cm³ of iso – octane was added to precipitate out the Asphaltene component. Filtration process of the solution was now carried out by making use of filter paper. From the filtration process, two components were obtained which was residue and filtrate. The residue is asphaltene and filtrate is maltene. The Maltene was collected into a sample bottle while the asphaltene was washed about five times with 20ml iso – octane. By the method employed by Olabemiwo et al (2008) using Column Chromatography, maltene fraction which is the filtrate was separated into saturated hydrocarbons, polycyclic aromatic hydrocarbon and polar compounds.

Analysis of Gas Chromatographic

The gas chromatographic analysis used was 5890 series (Hewlet Packard) that is equipped with flame ionization detector. The stationary phase used for the analysis is a fused – silica capillary column coated with 0.25m film of HP-5. For hydrocarbons that are saturated, about $3\mu L$ of sample was injected. The column temperature started at $60^{\circ}C$, held for 2minutes isothermally and then increased to $200^{\circ}C$ at the heating rate of $10^{\circ}C$ /min for 20minutes. It was held at this temperature for 2munutes and then increased to $320^{\circ}C$ at the heating rate of $12^{\circ}C$ for 5minutes. The carrier gas used was nitrogen at a pressure of 30 psi. At pressure

of 22 and 28 psi, Hydrogen and air were introduced respectively. 300°C and 320°C were used for Injector and detector temperature respectively.

The column temperature was held for 2minutes at about 70°C column temperature for the aromatic hydrocarbons and later increase to 250°C at heating rate of 15°C for 20minutes. It was held at 260°C for 6 minutes isothermally and then increased to 320°C for 6mnutes at heating rate of 15°C and it was at this temperature for 10minutes. Nitrogen, which is the carrier gas for this experiment was used at a pressure of 35psi. At a pressure of 25 and 30 psi, hydrogen and air was introduced respectively. At temperature of 300 and 320°C was when injector and temperature was used respectively and the sample of injected volume is 2μL. By making use of the standards supplied by the Gas Chromatography equipment manufacturer, Calibration curves for the standard and aromatic hydrocarbon were prepared.

Result and discussion

Gas Chromatography result of the saturated and aromatic fraction of irradiated bitumen with infrared.

Table 1: Aliphatic Hydrocarbon Profile of Bitumen Irradiated with Infra – red Radiation

PAH	AMOUNT (g/mg)				
	RAW BT	IRO 1 SAT	IRO 3 SAT	IRO 7 SAT	
C ₁₁	304.33	371.745	227.238	194.829	
C ₁₂	3.7772	2.5146	1.5918	2.3541	
C_{13}	37.3414	25.7805	56.4245	24.6311	
C_{14}	4.4330	2.99145	1.7349	3.1931	
C ₁₅	45.1838	32.5694	19.69	27.9708	
C ₁₆	4.1187	2.7674	1.6251	2.6861	
C ₁₇	6.6893	5.0672	2.8083	4.3397	
C_{18}	4.0167	3.3583	1.9249	2.7063	
C_{19}	3.9041	2.9776	1.7255	2.5150	
C_{20}	8.9376	9.1302	5.0815	6.2955	
C_{21}	3.9808	6.6905	3.5853	4.0090	
C_{22}	11.4659	8.1386	6.0837	6.2387	
C_{23}	2.2465	2.2687	1.2912	1.6358	
C ₂₄	115.834	40.9511	40.6723	50.120	
C ₂₅	5.9590	6.3874	4.9352	5.5857	
C_{26}	39.1534	25.049	16.561	17.788	
C ₂₇	10.5277	13.1155	8.0830	9.3792	

C_{28}	7.3891	17.6333	4.0959	4.9437
C_{29}	1.0401	0.8327	0.7083	0.7928
C_{30}	1.8928 × 10 ⁻⁵	-	-	9.16933 * 10 ⁻⁵
TOTAL	620.4383	579.9684	405.8604	372.0145
PAH'S				

Table 2: Polycyclic Aromatic Hydrocarbon Profile of bitumen irradiated with infra – red

PAH	AMOUNT (g/mg)				
	RAW BT	IRO 1 SAT	IRO 3 SAT	IRO 7 SAT	
Napthalene	0.7122	0.1160	0.12459	0.11831	
Acenapthylene	0.0000	0.0000	0.0000	0.0000	
Acephathene	0.0083	0.01394	0.005290	0.00575	
Fluorine	0.1980	0.08268	0.06521	0.026631	
Phenathrene	0.12949	0.07757	0.062855	0.02038	
Anthracene	0.02884	0.02988	0.010598	0.00679	
Fluoranthene	0.02289	0.012464	0.006254	0.00255	
Pyrene	0.01795	0.03190	0.01187	0.006787	
Benzo(a)	-	0.01750	0.0085474	0.0031555	
anthracene					
Chrysene	-	0.01833	0.07822	0.002763	
Benzo (b)	0.02399	0.003922	-	7.858 * 10 ⁻⁴	
fluoranthene					
Benzo (k)	0.10900	-	-	0.0010659	
fluoranthene					
Benzo (b)	0.0000	-	-	0.0000	
pyrene					
Indeno (1,2,3	0.08438	-	-	-	
- cd)					
Dibenzo (a,h)	-	-	-	-	
anthracene					
Benzo (g,h,i)	0.0246	_	-	-	
TOTAL	1.3596	0.4041	0.3734	0.1949	
PAH'S					

Bitumen samples irradiated with Infra – red radiations

Saturated Fractions: The total amount in g/kg of the aliphatic hydrocarbons was found to decrease as the period of exposure of the bitumen to ultraviolet radiation increases. This can be contributed to cracking and recombination of product. The irradiation of bitumen brought about the cracking of some higher molecular mass hydrocarbons to lower molecular mass radicals.[5]

Aromatic Fractions: The total amount in g/kg of the polycyclic aromatic hydrocarbons was found to decrease as the period of exposure of the bitumen to infra-red radiation increases. Benzo (a) anthracenes and chrysene which were absent in the control sample appeared after irradiation with infra-red radiations. Indeno (1, 2, 3-cd) pyrene and Benzo (g, h, i) perylene which were present in the control sample disappeared after irradiation with infra-red radiations. Acenaphthylene, Benzo (a) pyrene and Dibenzo (a, h) anthracene were absent in the control and irradiated sample. Benzo (b) fluoranthene which was present after one hour of irradiation disappeared after three hours of irradiation. Benzo(k) fluoroanthene which was present in the control sample disappeared after three hours of irradiation and later reappeared in minimal amounts of seven hours if irradiation.

Conclusion

The Gas Chromatogram result for Aromatic fraction of bitumen irradiated with infrared radiation respectively shows that the total number of polycyclic hydrocarbons decreases as the time of irradiation is increases. This reduces the quality of bitumen used for construction purposes or any other purposes. Therefore, bitumen to be used for construction or any other purposes should not be exposed for radiation. Olabemiwo et al(2010) reported the same thing that exposure of Agbabu natural bitumen to sunlight caused a decrease in its total aliphatic hydrocarbons. Distribution of individual aliphatic hydrocarbon was also found to vary with extent of irradiation of the bitumen with sunlight. The polycyclic hydrocarbon profile of the bitumen was also found to depend on the period of exposure of sunlight volatilization, cracking. The changes in the composition of the bitumen will, no doubt, reduce its strength. A reduction in strength will definitely increase the rate of ageing of the bitumen (6).Hence, Gas

chromatography analysis shown that bitumen consists of Sixteen Polycyclic hydrocarbon and these are responsible for the quality of bitumen.

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Appendix

Definition of abbreviation

PAH: Polycyclic Aromatic Hydrocarbon

RAW BT: Raw Bitumen

IRO SAT 1: Saturated fraction of bitumen irradiated with infrared for one hour

IRO SAT 3: Saturated fraction of bitumen irradiated with infrared for three hours

IRO SAT 7: Saturated fraction of bitumen irradiated with infrared for Seven hours