

# Effect of preharvest application of Silicon and saline water on postharvest quality of beet (*Beta vulgaris* L.)

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## Author's contributions

*This work was carried out in collaboration with all authors. Authors JSMF and TIS designed the study, performed statistical analysis, wrote the protocol and wrote the first draft of the manuscript. Author's MLMV and YHL managed bibliographic searches. Author AGSB, TJD transcribed the final version, placed it in the journal's rules and was responsible for submitting it to the publication. Authors TJD and JSMF managed the analyzes of the study as well as the work. All authors read and approved the final manuscript.*

## ABSTRACT

Beet is one of the vegetables richest in nutrient (bioactive compounds, folic acid and potassium). However, there are few studies on postharvest quality from the crop under irrigation conditions with saline waters, and because of that, the objective of this study was to evaluate effect of saline waters and silicon application in the preharvest on physicochemical quality of the beet. Two experiments were conducted with the objective of evaluating two forms of silicon application: via foliar (experiment 1) and via soil (experiment 2) about its influence in mitigating salt stress. In both experiments was adopted a randomized design with blocks in a 5 x 5 factorial, referring to five levels of electrical conductivity of the irrigation water (ECw): (0.5, 1.3, 3.25, 5.2 and 6.0 dS m<sup>-1</sup>) five doses of silicon (0.00; 2.64; 9.08; 15.52 and 18.16 mL L<sup>-1</sup>), they were combined according to the experimental matrix Central composite of Box totaling 10 treatments, with four replicates and three plants per plot. After harvesting, 70 days after transplanting, the following characteristics were evaluated: bulb dry mass, pH, titratable acidity (TA), total soluble solids (TSS), TSS/TA ratio and ash. **There was a significant effect for the salinity x silicon (Si) interaction applied via soil to the dry matter mass and titratable acidity. The electrical conductivities of irrigation water (ECw) and Si doses via leaf and soil influenced significantly for**

analyzed variables, except pH, total soluble solids, TSS/TA ratio, ash and titratable acidity, which were not influenced by EC<sub>w</sub> and nor by the doses of Si via soil and foliage. Irrigation with water of 6.0 dS m<sup>-1</sup> promotes better quality beet tuber. The fertilization on preharvest with silicon via soil or foliage improved postharvest quality of beet.

*Keywords:* *Beta vulgaris* L. Potassium silicate. Salt stress.

## 1. INTRODUCTION

In recent years soil salinization has been a major agricultural concerns, especially in arid and semi-arid regions [1], since in these regions besides the low rainfall and rainfall irregularity, there are still low-quality of water available for irrigation, then farmers use saline water [2], as in the case of producers of vegetables, especially beets.

Although salinity negatively interfere with the growth and development of crops, studies show that the salinity promotes higher quality postharvest of fruits and vegetables, as an example [3], working with watermelon they verified that the increase in the electrical conductivity of the irrigation water (EC<sub>w</sub>) from 2.77 to 4.91 dS m<sup>-1</sup> increased the soluble solids content. In cucumber, [4] also found soluble solids content and titratable acidity increased. The same was observed in tomato crops, where [5] found that use of water with salinity from 2.0 to 3.5 dS m<sup>-1</sup> increases the quality of tomato by increasing the TSS/TA ratio.

On the other hand, despite the increased quality of fruits and vegetables with the use of irrigation with saline water, crop productivity is reduced under salinity conditions. Restrictions on growth are observed due to the interaction of salts with nutrients in the soil, promoting imbalances in the plant nourishment [6; 7]. Thus, the accumulation of ions in tissues for long periods can cause injury and death of the plant [8].

One of the strategies to mitigate the salt deleterious effects is the application of substances that mitigate these effects, one of them is silicon [9; 10; 11; 12; 13]. Although considered a non-essential element, silicon is a beneficial element, which has the ability to reduce the impact of stressor agents [14]. Studies evidence that the supply of silicon increases crops growth, since this element indirectly acts on some photosynthetic and biochemical aspects, especially in plants under some sort of stress [15; 16; 17]. Furthermore, the increase in postharvest quality of vegetables with silicon application has been verified, such as lettuce [18] and strawberry [31].

Due to the scarcity in studies vising the effect of irrigation with saline water on postharvest quality of beet tuber, this study aimed to evaluate the effect of saline water and silicon application on physicochemical quality of beet.

75 **2. MATERIAL AND METHODS**

76

77 **Experimental location**

78

79 Two experiments were conducted in a greenhouse located in the Fruit sector belonging to the Federal  
80 University of Paraíba, municipality of Areia-PB, Brazil, located in the geographical coordinates  
81 6°51'47" and 7°02'04" South latitude and West longitude 35°34'13 "and 35°48'28" of the Greenwich  
82 meridian.

83

84 **Experimental design**

85

86 These experiments were conducted in order to assess two types of silicon application which were via  
87 foliar (experiment 1) and via soil (experiment 2) and their influence as a possible salt stress  
88 attenuating. Experiment 1 was conducted from August to October 2017 and experiment 2 from  
89 January to March 2018. In the experiments were adopted the experimental design with randomized  
90 blocks of factorial 5 x 5 combined according to the experimental matrix Central Compound of Box [19],  
91 referring to the electrical conductivity of irrigation water (ECw) and silicon doses (Si) with minimum  
92 values (-  $\alpha$ ) and maximum ( $\alpha$ ), respectively 0.5 to 6.0 dS m<sup>-1</sup> and from 0.00 to 18.16 mL L<sup>-1</sup>, totaling  
93 nine treatments, with four replicates and three plants per portion.

94 Beet seedlings cv. Wonder were grown in trays and planted in pots of 22 cm diameter, 16 cm in  
95 bottom diameter and 18 cm high, with a volumetric capacity of 8 dm<sup>3</sup>, and circular holes of 1 cm  
96 diameter on its bottom face, with purpose of allowing better root aeration and percolation of excess  
97 water.

98 The pots were filled with horizon soil A, collected at the depth of 0-20 cm, classified as Planossolo  
99 Háplico Eutrófico êndico/Alfisol [20], the physicochemical characteristics (Table 1) were analyzed  
100 according to the methodology of [20], respectively. The soil was air dried previously and properly  
101 homogenized, and placed in vessels accommodated previously screened (tulle fabric) and 200 g of  
102 crushed rock to prevent soil output by the vessels inferior orifices.

103

104 **Table 1:** Physicochemical characteristics of the soil used in the experiment.

Chemicals attributes		Physicals attributes	
pH	6.26	Ds	1.38
P (mg dm <sup>-3</sup> )	11.35	Pd	2.67
K <sup>+</sup> (mg dm <sup>-3</sup> )	40	Tp	0.48
Na <sup>+</sup> (cmol dm <sup>-3</sup> )	0.22	CC	78
H+Al (cmol dm <sup>-3</sup> )	1.82	PMP (g g <sup>-1</sup> )	43
Al <sup>+3</sup> (cmol dm <sup>-3</sup> )	0	Sand (g kg <sup>-1</sup> )	756.9
Ca <sup>+2</sup> (cmol dm <sup>-3</sup> )	3	Silt (g kg <sup>-1</sup> )	59.1
Mg (cmol dm <sup>-3</sup> )	1.9	Clay (g kg <sup>-1</sup> )	184
BS (cmol dm <sup>-3</sup> )	5.22	-	-

CEC (cmol dm <sup>-3</sup> )	7.03	-	-
V (%)	74.34	-	-
M (%)	0	-	-
OM (g Kg <sup>-1</sup> )	17.53	Textural classification	Sandy franc

105 Base sum (BS) = (Na<sup>+</sup>+K<sup>+</sup>+Ca<sup>2+</sup> + Mg<sup>2+</sup>); CEC = cation exchange capacity; EC = BS + (H<sup>+</sup> + Al<sup>3+</sup>); V =  
 106 (100 x SB/CTC); OM = organic matter. Ds = density of the soil; Pd = particle density; Tp = total  
 107 porosity; (1- (Ds / Dp) \* 100) Ucc =volumetric humidity level of field capacity - 0.033 MPa; Upmp =  
 108 humidity level of the permanent wilting point - 1.5 MPa.

109

110 The plants were irrigated daily, at the beginning raising the soil moisture around 80% of the field  
 111 capacity (CC). The different ECw were obtained using the NaCl, CaCl<sub>2</sub>.2H<sub>2</sub>O e MgCl<sub>2</sub>.6H<sub>2</sub>O salts, in  
 112 the proportion of 7: 2: 1 according to the characteristics presented in Table 2. Irrigation with the water  
 113 sources of different salinities was started 10 days after the emergence. In the first DAE, the slide was  
 114 calculated by the equation proposed by [21]. The total required irrigation (ITN), in mm, was calculated  
 115 by the equation of [22], considering 100% efficiency of irrigation application.

116

117 **Table 2:** Chemical characteristics of the waters used in the experiment

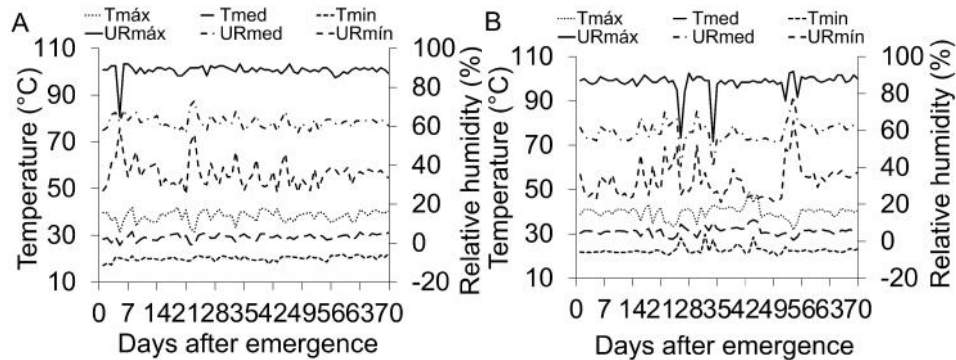
Attributes	Electric conductivity (dS m <sup>-1</sup> )				
	0.50	1.30	3.25	5.20	6.00
	<b>Values</b>				
pH	7.00	7.50	7.40	7.30	7.40
SO <sub>4</sub> <sup>-2</sup>	3.22	3.70	3.67	3.35	3.90
Mg <sup>2+</sup>	1.33	1.78	1.93	2.03	2.98
Na <sup>+</sup>	1.70	5.92	12.57	20.5	24.20
K <sup>+</sup>	0.20	0.21	0.20	0.20	0.21
Ca <sup>2+</sup>	0.73	1.58	1.78	1.88	2.53
CO <sub>3</sub> <sup>-2</sup>	0.00	0.00	0.00	0.00	0.00
HCO <sub>3</sub> <sup>-</sup>	2.75	3.50	4.00	4.25	4.25
Cl <sup>-</sup>	3.40	10.90	30.40	48.90	58.15
SAR (mmol L <sup>-1</sup> ) <sup>0.5</sup>	1.28	1.87	2.60	3.23	2.96
Classification	C2S1	C3S3	C4S4	C4S4	C4S4

118 EC= electrical conductivity at 25 °C; SAR = sodium adsorption ratio [Na<sup>+</sup>/(Ca<sup>2+</sup>+Mg<sup>2+</sup>/2)<sup>1/2</sup>]; CO<sub>3</sub><sup>2-</sup> =  
 119 Absent. Water classification according [23].

120

121 Silicon was applied as liquid potassium silicate (K<sub>2</sub>SiO<sub>3</sub>) with 12% Si and 15% K<sub>2</sub>O. In the first  
 122 experiment, Si doses were applied through a hand sprayer and in the second experiment the  
 123 application was carried out directly in the vessels. In these experiments, there was compensation of  
 124 K<sub>2</sub>O via foliar and via soil in the applications used in beets with the purpose of supplying the same  
 125 amount of potassium for all plants. The silicon application was done weekly, totaling 7 applications

126 during the beet cycle. The doses of (Si) were diluted in 1.2 L of distilled water and 50 ml of this solution  
 127 was applied to each plant according to the each via of application.  
 128 During the experiments conduction, the atmospheric data (Figure 1) were recorded daily with the HT-  
 129 600 Instruthermr® digital thermohygrometer installed inside the experimental area, at plants height. In  
 130 both experiments, mean temperature and weather values were close to the ideal range (25 °C) during  
 131 the crop cycle, according to [24].  
 132



133  
 134 **Figure 1:** Graphical representation of the relative air humidity and temperature in the experiments  
 135 conduction period, experiment 1 (A) and experiment 2 (A). Air maximum temperatures (Tmax), mean  
 136 (Tmeas) and minimum (Tmin) in ° C; maximum relative humidity of the air (URmáx), mean (URmed)  
 137 and minimum (URmin) in%.

138  
 139 Cover and fertilization of the plantation were done with 40, 180 and 90 kg ha<sup>-1</sup> of NPK, respectively,  
 140 with urea, simple superphosphate, and potassium chloride according to the chemical analysis of the  
 141 soil and fertilization recommendation for the state of Pernambuco, Brazil [25]. Phytosanitary controls of  
 142 pests and weeds were carried out during the experiment.

143  
 144 **Determination of physico-chemical parameters**

145  
 146 After harvesting, 70 days after transplanting, the following characteristics were evaluated: bulb dry  
 147 mass, pH, titratable acidity (TA), total soluble solids (TSS), TSS/TA ratio and ash.  
 148 Dry matter mass was obtained after weighing the bulb in an analytical balance, after drying in oven at  
 149 65°C; pH was measured with bench PH meter; the titratable acidity (g pulp 100 g<sup>-1</sup> citric acid), by  
 150 titration of the juice with NaOH a 0,1 M solution; Soluble solids content, determined with digital  
 151 refractometer; the TSS/TA ratio was through division; for ash determination, the methodology of the  
 152 [26] was adopted, from the incineration in muffle for 6 hours at a temperature of 550 °C.

153  
 154 **Data analysis**

155  
 156 The obtained data were submitted to analysis of variance and polynomial regression, and the mixed  
 157 model (MIXED) was used for the repeated evaluations in time using the statistical software SAS®  
 158 University [27].

159 **3. RESULTS AND DISCUSSION**

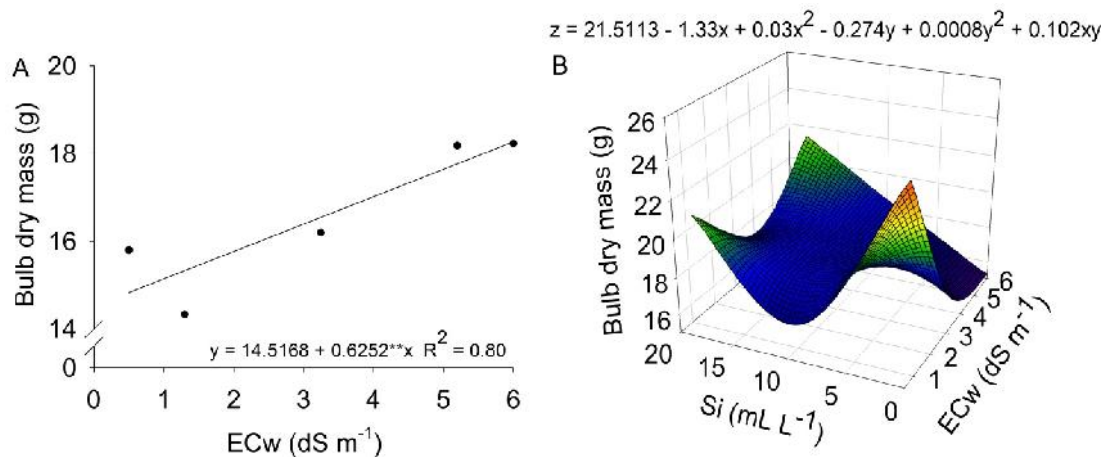
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161 There was a significant effect for the salinity x silicon (Si) interaction applied via soil to the dry matter  
 162 mass and titratable acidity. The electrical conductivities of irrigation water (ECw) and Si doses via leaf  
 163 and soil influenced significantly for analyzed variables, except pH, total soluble solids, TSS/TA ratio,  
 164 ash and titratable acidity, which were not influenced by ECw and nor by the doses of Si via soil and  
 165 foliage.

166 The increase of salinity in the irrigation water promoted a linear increase in the beet bulb dry matter  
 167 mass, presenting maximum values in the irrigated plants with water of 5.2 and 6.0 dS m<sup>-1</sup>, obtaining up  
 168 to 18 g. Plants irrigated with the salt water of 1.3 dS m<sup>-1</sup> presented the lowest dry matter mass of the  
 169 bulb, obtaining 14.2 g, a reduction of 21.11% in plants irrigated with water of 0.5 dS m<sup>-1</sup> compared to  
 170 those under 5.2 and 6.0 dS m<sup>-1</sup> (Figure 2A).

171 The dry matter mass presented decrease with the increase of the water salinity and with silicon doses  
 172 via soil, the highest values were obtained when these plants were irrigated with water of low salinity  
 173 (0.5 dS m<sup>-1</sup>) and without application of silicon (0 mL L<sup>-1</sup>) was obtained 23 g of dry matter mass of the  
 174 bulb. By increasing the values of ECw and the silicon doses there was a reduction until 5.2 dS m<sup>-1</sup> and  
 175 9.08 mL L<sup>-1</sup>, in this order, with later addition until a dose of 18.16 mL L<sup>-1</sup> of silicon (Figure 2B).

176



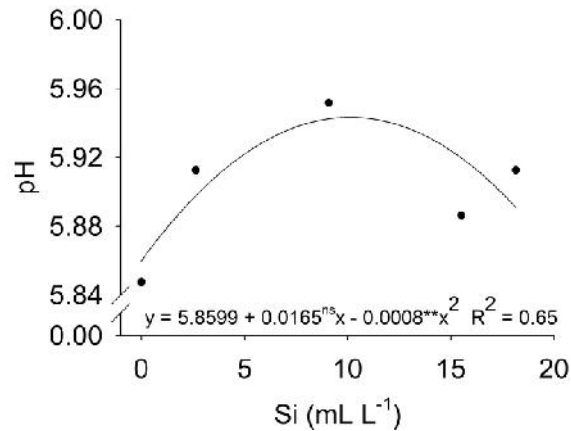
177

178 **Figure 2:** Electrical conductivity effect of the irrigation water (A) and the electric conductivity of silicon  
 179 doses applied to the soil and in the irrigation water (B) in the dry matter of beet bulb

180

181 The pH of the beet pulp adjusted itself to the quadratic regression model, with an increase due to the  
 182 increase in Si doses via foliage; the pH values were raised to 5.94 relating to estimated optimal dose  
 183 of 9.08 mL L<sup>-1</sup> (Figure 3). Si doses above 9.08 mL L<sup>-1</sup> have reduced the beet pulp pH to the dose of 15.52  
 184 mL L<sup>-1</sup> with further increase up to 18.16 mL L<sup>-1</sup>. The values found in this study are considered low. pH  
 185 works as an indication of the vegetable flavor, having relation inverse to the acidity [28].

186 These results corroborate to those obtained by Korkmaz et al. (2017) in tomato, they found that the  
 187 increase in ECw from 0 to 4.4 dS m<sup>-1</sup> promoted an increase of the fruit pulp pH. [5] also observed that  
 188 which pH of the tomato pulp increased with salinity increasing in irrigation water, with a maximum  
 189 value of 4.8 in plants under 5.0 dS m<sup>-1</sup>.



191

192 **Figure 3:** Effect of silicon doses via foliage in the pH of beet pulp

193

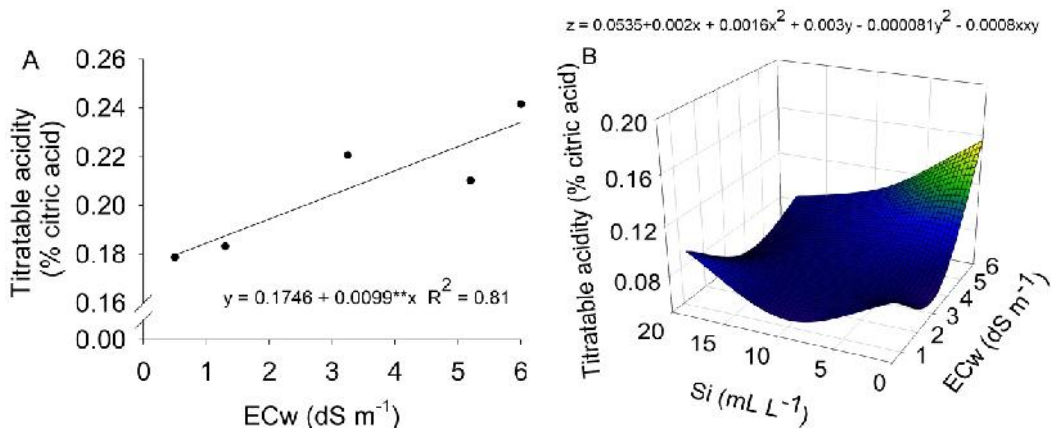
194 It was observed that the increase in EC<sub>w</sub> promoted increase in titratable acidity of beet pulp, then  
 195 plants subjected to the highest EC<sub>w</sub> (6.0 dS m<sup>-1</sup>) had the highest titratable acidity (0.23 g citric acid)  
 196 and plants irrigated with water of lower EC<sub>w</sub> (0.5 dS m<sup>-1</sup>) had the lowest titratable acidity (0.17 g citric  
 197 acid), 29.16% reduction comparatively (Figure 4A).

198 As increased salt concentration in the irrigation water has been found that the titratable acidity of 0.08  
 199 g of citric acid under 0.5 dS m<sup>-1</sup> increased 0.16 g of citric acid in the beet pulp irrigated with water of  
 200 highest salinity, increasing up to 50%. [29] also found that the titratable acidity on melon had increased  
 201 with the increase of the salinity of irrigation water, which observed increase of 0.107% citric acid EC<sub>w</sub>  
 202 0.49 and 1.75 dS m<sup>-1</sup>, however, the authors also noted that from these salinity the titratable acidity  
 203 reduced 4.7% when had water salinity increased until 2.4 dS m<sup>-1</sup>.

204 On the other hand, it was verified that the increase in Si doses reduced the titratable acidity of the beet  
 205 pulp up to the Si dose of 9.08 mL L<sup>-1</sup> with subsequent increase until reaching the maximum dose  
 206 (18.16 mL L<sup>-1</sup>), yielding 0.10 g of acid with the this application (Figure 4B). This behavior was also  
 207 verified by [18] in lettuce, which observed that the treatment with 28 mg L<sup>-1</sup> of silicon, obtained 140 mg  
 208 of citric acid 100 g<sup>-1</sup> of fresh matter, being greater than in the plants of the control that presented  
 209 average values of 114 mg of citric acid 100 g<sup>-1</sup>.

210 The increase in the titratable acidity of the beet pulp in function of the silicon doses corroborates the  
 211 data obtained by [31] in strawberry, in which the highest values for titratable acidity were with the  
 212 application of Si via foliage, in which 1.19 mg of citric acid 100 g<sup>-1</sup> was obtained with the application of  
 213 69 mg kg<sup>-1</sup> of Si, and via soil the highest value was 1.13 mg of citric acid 100 g<sup>-1</sup>, with the application of  
 214 97.5 mg kg<sup>-1</sup> of Si.

215



216

217 **Figure 4:** Electrical conductivity effect of the irrigation water and doses of silicon applied via foliage (A)

218 and via soil (B) in the titratable acidity of beet

219

220 There was an increase in the soluble solids contents of the pulp when the beet plants were irrigated  
 221 with saline water and silicon application, in both Experiment I (Figure 5A) and Experiment II (Figure  
 222 5B). In the experiment I, data were adjusted to the linear model increasing, and then the increase in  
 223 the ECw promoted an increase in the soluble solids content, in which the maximum value of 15.7° Brix  
 224 was obtained in the plants irrigated with water of 6.00 dS m<sup>-1</sup> (Figure 5A). In the experiment II, data  
 225 were adjusted to the quadratic regression model, where the maximum value in the soluble solids  
 226 content was obtained in the plants that have received 9.08 mL L<sup>-1</sup> of silicon, obtaining 14.3° Brix  
 227 (Figure 5B).

228 Similar results were observed by [3] in melon, when they verified that the increase in salinity in the  
 229 irrigation water from 2.77 to 4.91 dS m<sup>-1</sup> promoted an increase in the soluble solids content of the  
 230 'Shadow' cultivar, increasing the soluble solids values from 3.58 to 5.08%, respectively. [4] also  
 231 verified that saline increase in irrigation water increases the content of soluble solids in cucumber.

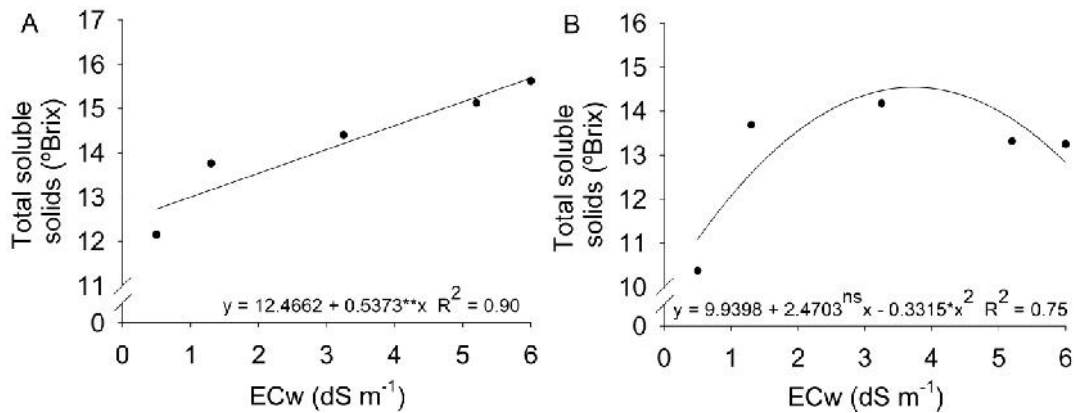
232 The increase in soluble solids content in the beet pulp with increase of water salinity may have been  
 233 due to the increase in the concentration of photoassimilates (solutes), since the increase in the soluble  
 234 solids is mainly due to the deleterious effects of the salts that reduce the bulbs average weight, but  
 235 they increase soluble solids [30].

236

237

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239

240

**Figure 5:** Electrical conductivity effect irrigation water (A) and application of silicon via soil (B) on soluble solids contents of beet pulp

241

242

#### 243 4. CONCLUSION

244

245 Using water with an electrical conductivity of 6.0 dS m<sup>-1</sup> promotes better tuber beet quality;

246 The fertilizations on preharvest with silicon, applied via soil or foliage, improved the postharvest quality  
247 of the beet.

248

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