BIOCHEMICALAND GENOTOXIC EFFECTS OF ECOLOGICALLY RELEVANT CONCENTRATIONS OF NICKEL IN *CLARIAS GARIEPINUS*

ABSTRACT

This study was aimed at determining the biochemical and genotoxiceffects of ecologically relevant concentrations of Nickel (Ni) in Ologe Lagoon; which constitutes its originality. An initial field study was conducted to determine the concentrations of some heavy metals (Arsenic, nickel and lead) in surface water, sediments, and fish from Ologe Lagoon. Ten fish per test concentration were used for the bioassay. Oxidative stress indicators (superoxide dismutase, catalase, reduced glutathione, and malondialdehyde) and nuclear abnormalities were assessed in fish chronically exposed to ecologically relevant concentrations of Ni. Environmentally relevant concentrations of Ni did not have any significant effects on the levels of reduced glutathione, catalase, superoxide dismutase and malondialdehyde after 30 days of exposure. Similarly, Ni had no significant effects on all tested parameters of genotoxicity after day 30. Therefore, environmentally relevant concentrations of Ni may not have any deleterious effects in terms of oxidative stress and genotoxicity.

Keywords: *Biomarker; Ecological relevance; Genotoxicity; Nickel; Ologe Lagoon; Oxidative*

stress.

1. Introduction

The discharge of **partially** treated and untreated industrial effluents is considered as a major source of water pollution in Nigeria. Industrial growth is increasing the levels of toxicants that may have catastrophic impacts on Nigerian water bodies. These toxicants include organic compounds and heavy metals. Heavy metals refer to elements that have relatively high densities, especially above $5g/cm$ [1]. They occur naturally as trace elements, and are present in abiotic and biotic components of the ecosystem [2]. However, anthropogenic activities have increased the concentrations of these metals in the environment which have resulted in heavy metal pollution

[3]. According to Don-Pedro et al. [3], the sources of heavy metals in aquatic ecosystems are the direct discharge of domestic and industrial effluents, and runoff from urban and agricultural lands. In aquatic ecosystems, heavy metals are highly persistent and can be amplified along the food chain [4]. Heavy metals concentrations in most Nigerian rivers were found to be above acceptable and permissible levels [5,6]. The impacts of heavy metals on **human health** date back to 1956 when the cases of Minamata disease in humans caused by methyl-mercury were reported in Japan [7]. Others include Itai-Itai disease caused by cadmium poisoning [8]. High levels of manganese in drinking water induced intellectual dysfunctions in children in Araihazar, Bangladesh [9]. The effects of arsenic in aquatic organisms range from cytotoxicity in fish cell lines [10,11] to oxidative stress [12-14]. Lead, yet another toxicologically important heavy metal has been a culprit in several biological effects that include haematological [15], neurological [16], and physiological effects [17]. Carcinogenicity [18], immune-suppression [19,20], and respiratory disorder [21] have been observed in aquatic organisms exposed to Nickel. The objectives of this study were to determine the current levels of some heavy metals in surface

water, sediments and fish from Ologe Lagoon, and to conduct an ecotoxicological assessment of the predominant heavy metal in the water body.

2. Materials and Methods

2.1 Study Site

Ologe Lagoon is a part of the Lagos Lagoon system that consists of Lagos Lagoon, Lekki Lagoon, and Badagry Creek [22]. Ologe Lagoon's edges are shallow while the centre is deep [23,24]. It has an average depth of 2.42 m with a wide navigable mouth that allows for recreation, fishing, and transportation [23,24].

2.2 Field Studies and Heavy Metal Analyses

Samples were collected from three sampling sites, lying between latitudes 6°29'41"N and 6°29'50''N, and longitudes 3°5'60''E and3°6'5''E, in reference to the direction where effluents are received from Agbara Industrial Estate (Figure 1). Temperature, pH, conductivity, turbidity, salinity, dissolved oxygen, and total dissolved solids (TDS) of the lagoon surface water were determined *in-situ* with Horiba U50 Gmulti water quality meter**.** Water samples were collected using a 1 L plastic container; sediments were collected with a Venn-Grab sampler and placed in foil wraps [25]. The fish samples were collected with the aid of local fishermen, and preserved with ice packs before digestion. The digestion of samples was done according to the procedure described by Zheljazkov and Nielson [26]. Heavy metal analysis was done with a Perkin Elmer atomic absorption spectrometer.

2.3 Collection and Acclimatization of Experimental Fish

A total of 100 post-juvenile *Clariasgariepinus* (weight 18–20 g, and length 10 – 15 cm), were purchased from a fish farm in Ikorodu, Lagos State. They were transported in a 50 L capacity rectangular tank containing aerated water to the laboratory, and kept in holding tanks $(40 \text{ cm} \times 30$ cm×30 cm). During acclimatization, these fish were fed with "catfish grower", twice daily (morning and evening). The acclimatization was for a week, and water was changed every 3 days to prevent accumulation of toxic waste metabolites. Laboratory conditions were kept at 27–28 °C, 65–75 % humidity, and 10-h/14-h light/dark cycle for 2 weeks before bioassay in accordance with APHA/AWWA/WPCF [27].

Figure 1: Ologe Lagoon and Sampling Sites Localization Lagoon Adapted from [28]

2.4 Bioassay Procedure

A total of ten (10) acclimatized fish eachwere randomly caught using a plastic sieve from the stock in the holding tank, and carefully transferred to the different concentrations of the chemical as well as in control in each bioassay tanks. The respective concentrations of the test chemical were duplicated making 5 fish per test concentration. The fish were not fed for 24 hours before exposure. The test containers were labelled with each concentration and filled with 6 L of water each. The test solution was prepared using the method described by Nunes et al. [29]. In this study, NiW meanstest concentration derived from Ni surface water concentration in Ologe Lagoon while NiT is thetest concentration derived from Ni concentration in fish from Ologe Lagoon. in the holding tank, and carefully transferred to the different concentrations of the chemical
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2.5 Measurement of Anti-oxidative Stress Enzymes and Lipid Peroxidation

The reduced glutathione (GSH) of liver tissue as non-protein sulphydryls was determined according to the procedure described by Sedlak and Lindsay [30]. Superoxide dismutase (SOD) activity was determined by its ability to inhibit the auto-oxidation of epinephrine determined by the increase in absorbance at 480nm as described by Sun and Zigma [31]. Catalase (CAT) activity was determined according to the methodology of Sinha [32]. Thiobarbituric acid reactions (TBARS) assay was used to determine the levels of the lipid peroxidation product, malondialdehyde (MDA) [33].

2.6 Nuclear Abnormalities

The staining procedure was performed on blood smears obtained from fish samples. The smear of the peripheral blood collected using a heparinized syringe from the caudal vein of the fish sample was made on clean glass slides. Glass slides prepared per group were processed in accordance with Singh et al. [34]. The smeared slides were allowed to air-dry at room temperature, fixed in methanol for 15 minutes and then stained with May-Grunwald stain and allowed to air dry for 6 hours, then rinsed out with distilled water before staining with 5% Giemsa stain; then left to dry for 12 hours [35]. The slides were analyzed at $\frac{100x}{100x}$ for micronuclei and nuclear abnormalities. Micronucleus was smaller than one-third of the main nucleus and did not touch the main nucleus. Cells having two nuclei with approximately equal sizes were scored as binucleated, while cells with round appearances and basophilic cytoplasm were scored as immature erythrocytes [36].

2.7 Data Analysis

The mean and standard error (Mean±S.E.), and comparison of means were analysed using the Statistical Package for the Social Sciences (SPSS) Version 20. One-way ANOVA was used to test for significant difference between means, and differences in means were considered significant when P<0.05 and separated using Duncan's Multiple Range (DMR) test.

3. Results and Discussion

3.1 Physical and Chemical Parameters of Ologe Lagoon

The results of physical and chemical parameters evaluated in the three stations are presented in Table 1. The mean values of temperature and pH were within NESREA recommended limits [37]. TDS and turbidity values of 0.24 ± 0.02 g/L and 66.63 ± 17.04 NTU respectively were above NESREA safe limits $\left[\frac{37}{1}\right]$ (Table 1). The high amount of TDS might have been as a result of the discharge of effluents into the lagoon by nearby industries. Dan'azumi and Bichi[38] assessed the implications of industrial pollution on a source of water supply and documented that effluents discharged into water bodies by industries consequently increased the quantity of solid dissolved in the water.

The mean concentration of Arsenic was 0.0007 ± 0.0003 mg/kg in sediments while Lead levels were 11.89 \pm 1.61mg/kg and 1.34 \pm 1.02 mg/L in sediments and surface water respectively (Table 2). The mean concentrations of Nickel in surface water and sediments were 6.56±2.35 mg/kg and 1.85±0.22 mg/L respectively (Table 2). Arsenic and Lead were not detected in the homogenate sample of fish, but Nickel was found to be 3.72 mg/kg in the homogenate sample (Table 2). The Bioaccumulation Factor (BAF) and Biota to Sediment Accumulation Factor (BSAF) for nickel were 2.01 and 0.57 respectively (Table 2).

Table 1: Physicochemical Characteristics of Ologe Lagoon's Surface Water

NESREA -National Environmental Standards and Regulatory Enforcement Agency NA -Not Available

Table 2: Levels of Arsenic (As), Lead (Pb), and Nickel (Ni) in Sediments, Surface water and

K.

U

Fish Tissues from Ologe Lagoon

ND: Not detected using AAS. ND means values < 0.001

3.2 Oxidative Stress

After 15d of exposure, the reduced glutathione (GSH) activities in fish exposed to NiT were significantly ($p < 0.05$) increased while NiW had no significant ($p > 0.05$) effect on GSH activities. The mean values ranged from 19.86 ± 4.79 to $43.07\pm2.72\mu$ mol/ml and 18.10 ± 3.11 to 27.54 \pm 5.14 µmol/ml on the 15th and 30th day respectively (Figure 2). However, NiT and NiW did not have any significant (*p > 0.05*) effects on the activities of GSH after 30 days of exposure. The results of GSH obtained afterNiTexposure agree with the transitory pattern observed by Kuroshima[39] and Lange et al. [40]. GSH is the primary line of defence against reactive oxygen species (ROS) [41]. The initial elevation of GSH may be attributed to an increased flux of Ni through the liver, with the removal of GSH by metals probably stimulating the synthesis of GSH [42].

The results showed that there was no significant difference $(p > 0.05)$ between the **superoxide** dismutase (SOD) levels of control and the exposed groups after 15 and 30 days of exposure. The mean values ranged from 4.28 ± 0.20 to 4.61 ± 0.50 µmol/mland 4.35 ± 0.11 to 5.01 ± 0.62 µmol/ml on the $15th$ and $30th$ day respectively (Figure 3). The activities of catalase (CAT) in fish from the treated groups were not significantly different ($p > 0.05$) from those of the control after 15 and 30 days of exposure. The mean values ranged from 17.85±2.39 to 25.13±3.53µmol/mland 17.37 \pm 4.59 to 27.00 \pm 4.76 µmol/ml on the 15th and 30th day respectively (Figure 4). Keramati and Ramin [43] observed that the activities of catalase in the liver of *Rutilusrutilus*exposed to diazinon were not altered. Ahmad et al. [44] associated the inactivity of CAT to the high activity of Glutathione peroxidase (GPX), which acts as a defence against the production of H_2O_2 .

There was no significant difference $(p > 0.05)$ between the level of lipid peroxidation product, malondialdehyde (MDA) in the control and treated fish after 15 and 30 days of exposure. The mean values ranged from 1.80±0.19 to 2.05±0.15 µmol/mland 1.36±0.08 to 1.42±0.29 µmol/ml on the $15th$ and $30th$ day respectively (Figure 5). Palermo et al. [45] observed an increase in the level of MDA in the liver of *Prochiloduslineatus*exposed to 2500 µg/L of Ni, however, the level of MDA in the gills of *P. lineatus*remained unchanged. Liu et al. [46] reported that MDA level remained unchanged despite a marked increase of ROS generation observed in *Oreochromisniloticus*exposed to 15 mg L⁻¹ of PFOS. Liu et al. [46] went further to state that the reasons for the insignificant change in MDA level in the fish were unknown. After 30 days of exposure, environmentally relevant concentrations of Ni had no significant ($p > 0.05$) effects on GSH, SOD, CAT, and MDA levels in the test fish. Therefore, environmentally relevant concentrations of Ni did not induce oxidative stress in *C. gariepinus.*

Figure 3: SOD levels in fish exposed to NiW and NiT for 15 and 30 days

Day

 $DAY3O$

DAY 15

DAY 30

DAY 15

Conc (mg/l)

Ni (Lagoon Fish Tissue
Conc.)

Figure 5: MDA levels in fish exposed to NiW and NiT for 15 and 30 days

3.3 Nuclear Abnormalities

Plates 1(a-s) illustrate the nuclear abnormalities in the blood of fish exposed to environmentally relevant concentrations of Nickel. The results showed that after 15 days of exposure, the frequency of micronuclei in the fish exposed to NiT was significantly (*P <0.05*) higher than that of the control while other nuclear abnormalities were not significantly $(P > 0.05)$ induced (Table 3). Çavaş and Ergene-Gözükara [47] stated that nuclear abnormalities were indicators of genotoxic damage. Previous studies have shown that heavy metals induce genotoxic effects in fishes by increasing the number of micronuclei in their blood [47,48]. Formation of micronuclei in animals could be an effect of chromosomal breakage or dysfunction of the spindle mechanism by genotoxic agents such as heavy metals [49].However, at day 30 post-exposure there was no significant difference in the frequency of micronuclei in the control group and exposed group (Table 3). It is possible for organisms to undergo DNA repair after exposure to genotoxic agents [50].

Plates 1a-s: Photomicrograph of blood samples of *C*. *gariepinus* showing normal cells (control group) and nuclear abnormalities (exposed groups) (100x). **a&j-** Normal cells**b,k&s**micronucleated cell (arrow)**c,l&r-** binucleated cell (arrow)**d&m-** nuclear bud (arrow) **e&n**notched nucleus (arrow) **f&o-** 8-Shaped nucleus (arrow) **g&p-** blebbed nucleus (arrow) **h&q**polychromatic erythrocyte (PCE) (arrow) **i-** lobed nucleus (arrow).

Nuclear Abnormalities	Duration(days)	Control	NiW	NiT	
Micronuclei (%)	15	0.00 ± 0	0.03 ± 0.03	$1.2 \pm 0.51*$	
	30	0.13 ± 0.09	0.57 ± 0.33	0.13 ± 0.13	
Binuclei $(\%)$	15	0.10 ± 0.56	0.00 ± 0	$0.47 + 0.27$	
	30	0.07 ± 0.03	1.43 ± 1.23	0.50 ± 0.32	
Buds $(\%)$	15	0.10 ± 0.10	0.00 ± 0	0.00 ± 0	
	30	0.00 ± 0	0.03 ± 0.03	0.67 ± 0.03	
Notched nuclei $(\%)$	15	0.00 ± 0	0.03 ± 0.03	0.20 ± 0.15	
	30	0.00 ± 0	0.07 ± 0.03	0.13 ± 0.09	
8-Shaped Nuclei (%)	15	0.80 ± 0.70	0.30 ± 0.10	0.40 ± 0.21	
	30	0.00 ± 0	0.07 ± 0.03	0.03 ± 0.03	
Blebbed Nuclei $(\%)$	15	0.03 ± 0.03	$0.07 + 0.07$	0.07 ± 0.03	

Table 3: Frequencies of Nuclear Abnormalities in the Blood of the Fish (*Clariasgariepinus***)**

Alla

**means significantly different at p < 0.05 in rows while different letters (superscript) in the upper case means significantly different at p < 0.05 between durations of exposure.*

4. Conclusion

The current levels of the **physical and chemical** parameters of Ologe Lagoon showed that temperature and pH were within NESREA's safe limits whereas turbidity and NTU were above NESREA's safe limits. The field assessment of heavy metals in the lagoon indicated that Nickel was the predominant metal in surface water and fish from the lagoon.

Results from the biochemical analysis indicated that environmentally relevant concentrations of Ni in Ologe Lagoon may not be a causal factor of oxidative stress or peroxidative damage.

Data from the genotoxicity assaysuggest that the test fish may have the ability to recover from the genotoxic effect of the current concentrations of Ni in the lagoon. However, there is a need for the continuous monitoring of the impact of ecologically relevant concentrations of Ni especially with other biomarkers not utilized in the present study.

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